

Benzoannulation of aromatic heterocycles: advances in the 21st century

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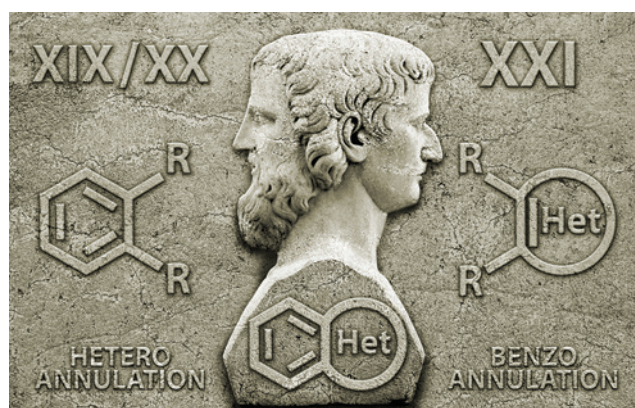
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Benzoannulated aromatic heterocycles (BAHs), also known as benzoheterocycles, are key building blocks in the development of functional materials and pharmaceuticals. They are involved in a variety of biochemical processes in nature. The prevalence and widespread use of these molecules stimulates the chemical community's ongoing interest in developing methods to construct carbazole, indole, quinoline, isoquinoline and benzo[*b*]thiophene motifs. The most common strategy for preparing them is the heteroannulation of functionalized benzene derivatives. Over the last two decades, an alternative approach based on the annulation of heterocyclic derivatives has been developed: benzoannulation, also known as benzannulation. Compared to classical heteroannulation, this approach has several advantages and has led to significant progress in the availability of a variety of benzo-heterocycles in recent years. This review is the first to analyze the development of benzoannulation methods for aromatic heterocycles in the 21st century. We highlight the advantages of the benzoannulation strategy, including the versatility of the methods, the availability of starting compounds and the ability to obtain products with specified substituents in the benzene ring. This review aims to help chemists with the synthesis of benzoheterocycles of a specific structure for various applications, ranging from the design of biologically active compounds and the synthesis of natural products to materials chemistry.

The bibliography includes 298 references.

Keywords: heterocycles, benzoannulated heterocycles, benzoannulation, benzannulation, carbazole



Dedicated to Academician of the Russian Academy of Sciences, V.P. Ananikov, on the occasion of his anniversary

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1. Introduction

Benzoannulated aromatic heterocycles (BAH) are condensed bi- and polycyclic aromatic systems consisting of one heterocyclic and one or two benzene units. The most common representatives of such compounds include indole, carbazole, benzo[*b*]-

thiophene, dibenzofuran, quinoline, isoquinoline, acridine, *etc.* (Fig. 1a). Benzo-fused heterocycles are frequently found in the structures of natural compounds. The tricyclic core of carbazole, for example, appear in many natural alkaloids isolated from plants,^{1,2} indole derivatives are found in marine organisms (sponges, shellfish, algae, *etc.*³), benzothiazole alkaloids are

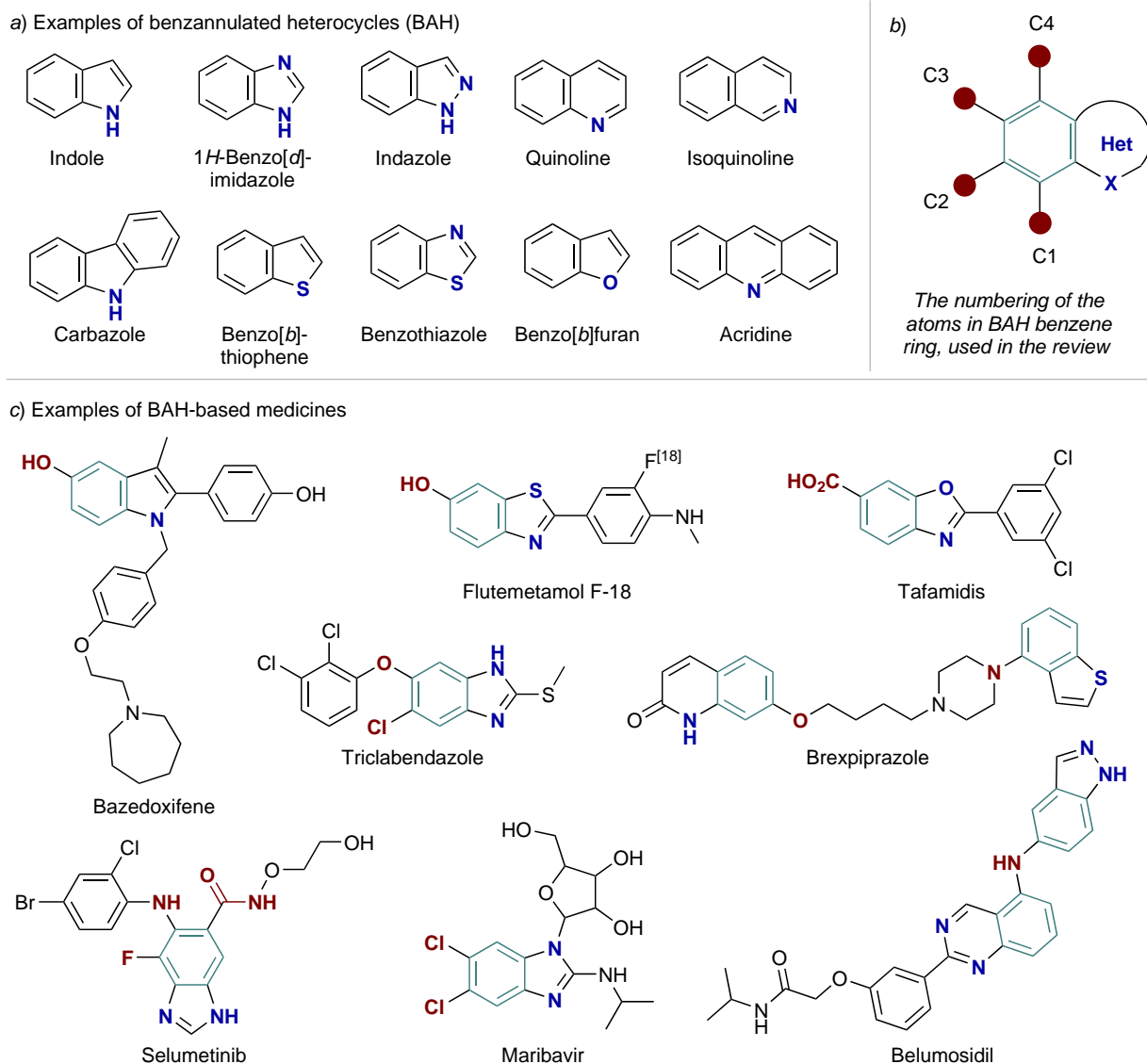


Figure 1. Exemplary benzoannulated heterocycles (a) and BAH-containing medicines (c); numbering of carbon atoms in the benzene ring (b).

found in some plants,⁴ dibenzofuran derivatives are mainly found in lichens and ascomycetes,⁵ and most quinoline and isoquinoline alkaloids are secondary plant metabolites from various families.⁶ The total synthesis of natural alkaloids based on benzoheterocycles (primarily carbazole^{7–10} and indole^{11–13}) is one of the most notable areas of organic synthesis.

The synthesis of heterocyclic compounds with specified properties is a major challenge in modern organic chemistry.^{14–22} Heterocyclic moieties are widely used in the design of innovative materials for various fields of science and technology,^{23–28} and are

also found in biologically active compounds and drugs.^{29–34} According to the US Food and Drug Administration (FDA), around 20% of drugs approved in the last 10 years contain BAH motifs in their structure.^{35–39} Some examples are shown in Fig. 1c. Bazedoxifene, for example, is an indole derivative used to treat the negative effects of menopause. The drug Flutemetamol F-18 contains a benzothiazole moiety and is used to diagnose Alzheimer's disease. Tafamidis, a benzothiazole-based drug, is used to treat transthyretin amyloidosis. Triclabendazole, which contains a benzimidazole moiety, is recommended as an anthelmintic drug for

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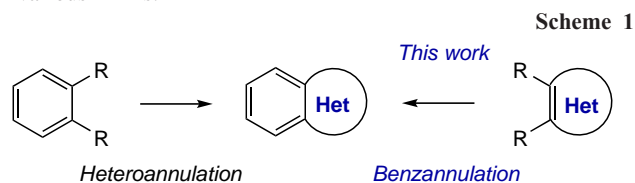
Current research interests: organic synthesis, chemistry of heterocyclic compounds, synthetic photochemistry, photochromic compounds and materials.

Translation: N.M.Vinogradova

fascioliasis. The benzamidazole derivatives Selumetinib and Maribavir are used to treat children with nervous system diseases and to control post-transplant cytomegalovirus infection, respectively.

Some drugs comprise two or more BAH scaffolds. For example, Belumosudil, a selective ROCK2 inhibitor, contains indazole and quinazoline moieties and is used to treat complications after stem cell or bone marrow transplantation. Brexpiprazole, which is based on benzothiophene and quinoline, is an atypical antipsychotic agent. In all of the above examples, the benzene ring in the BAH moiety is functionalized with various substituents, such as hydroxy, carboxyl, amino groups or halogen atoms.

There are two main strategies for synthesizing BAHs (Scheme 1): heteroannulation (a heterocycle is formed on the basis of benzene derivatives) and benzoannulation (a benzene ring is assembled from substituents of the heterocycle, also known as the ‘back-to-front’ approach.^{40,41} Almost all classical methods are based on the heteroannulation of benzene derivatives. Previously, many methods were developed for obtaining indoles,^{42–47} carbazoles,^{48,49} benzo[*b*]furans,⁵⁰ benzo[*b*]thiophenes⁵¹ and other classes of BAHs⁵² which have proven highly effective. However, towards the end of the 20th century, researchers became interested in an alternative approach to BAH synthesis, *viz.*, the benzoannulation of heterocycles. Compared to the traditional method, this approach has several advantages, including greater potential for substrate pre-functionalization and the ability to produce compounds with specific substituents in the benzene ring. This is due to the lower aromaticity and higher reactivity of heteroaromatic compounds. In general, heterocyclic moieties are easier and more convenient to functionalize than benzene rings. This has led to significant progress in the availability and application of various BAHs.



Since the beginning of the 21st century, there has been a clear interest within the chemical community in developing methods for the benzoannulation of aromatic heterocycles as a more convenient alternative to the traditional approach. In recent years, several reviews have been published on the synthesis of indoles⁵³ and carbazoles.^{54,55} Significantly more works have analyzed all possible methods for synthesizing BAHs, regardless of the production strategy, including indoles,^{56–58} carbazoles,^{59–62} benzothiophenes,⁶³ quinolines,^{64,65} and others.^{66,67} To the best of our knowledge, there is no review in the current literature that summarizes the strategies for the benzoannulation of all known aromatic heterocycles.

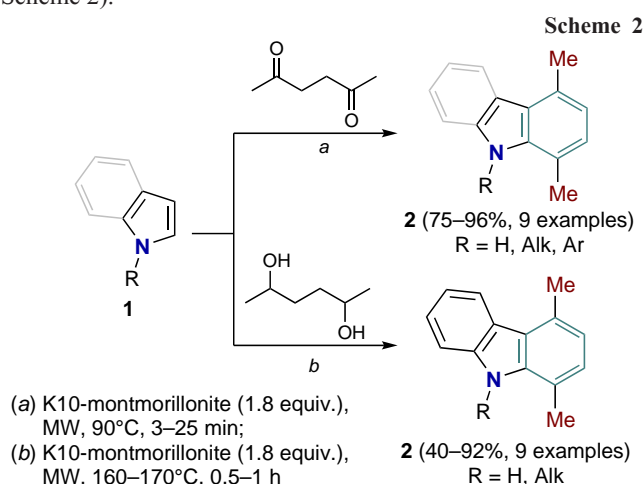
This paper is the first to cover methods of creating benzene rings for various classes of aromatic heterocycles that have been developed since 2000. Examples of recently introduced drugs (see Fig. 1 *c*) demonstrate that the presence of substituents in the benzene ring is crucial for the manifestation of the necessary biological activity of the compound. Therefore, rather than considering benzoannulation methods from the perspective of cyclization types and benzene moiety construction, as is done in many previously published reviews, this paper considers them in terms of the nature and position of substituents in the resulting benzene ring. To unify the position of substituents, carbon atom in all BAGs are designated as C1–C4 (see Fig. 1 *b*). Our proposed analysis of methods may be useful for developing

approaches to obtaining bi- and tricyclic fused heterocycles containing O-, S-, C- and N-substituents in specific positions on the resulting benzene ring.

2. Benzoheterocycles with hydrocarbon substituents

2.1. Alkyl-, alkenyl- and alkynyl-substituted heterocycles

This Section discusses synthetic approaches to heterocycles that contain alkyl, alkenyl or alkynyl substituents in the benzene ring. Schemes 2–4 illustrate the reactions used to produce alkyl-substituted indoles and carbazoles. Products **2**,[†] which have methyl groups at positions C1 and C4, are formed by the annulation of 2,3-unsubstituted indole and pyrrole (compound **1**) with either hexane-2,5-dione⁶⁸ or hexane-2,5-diol (see Scheme 2).⁶⁹



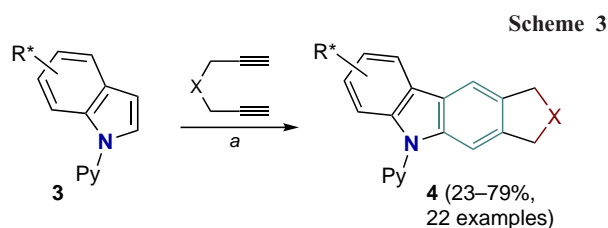
These reactions are completed in a relatively short time in good yields under microwave (MW) activation in the presence of the acidic mineral K10 montmorillonite. The first modification of this method, namely, benzoannulation with hexane-2,5-dione, tolerates various substituents, including phenyl and alkyl groups, at the pyrrole or indole nitrogen atoms.⁶⁸ The second modification is only described for unsubstituted and methyl-substituted indoles.⁶⁹ Subsequently, it was demonstrated⁷⁰ that this approach is also suitable for preparing aryl- and ester-substituted carbazoles (see, *e.g.*, Scheme 32).

The synthesis of carbazoles **4**, annulated with saturated carbocyclic and heterocyclic rings at positions C2 and C3, was carried out *via* a manganese(I)-catalyzed reaction of 2,3-unsubstituted indole-3-derivatives with 1,6-diene (see Scheme 3[‡]).⁷¹ In this case, the presence of a directing group (pyridin-2-yl) is necessary at the nitrogen atom of the heterocycle. Using this method, a large number of the corresponding products was synthesized, achieving yields ranging from 23% to 79%.

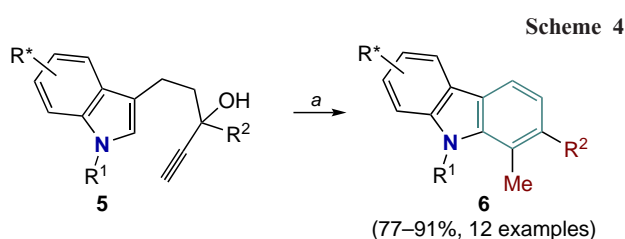
N-substituted carbazoles **6** bearing alkyl groups at the C1 and C2 positions were obtained from (indol-3-yl)pentin-3-ols **5** by

[†] In this review, the compounds are numbered according to their position in the Schemes, from substrate to product. Light grey fragments and dotted lines indicate their potential presence in the molecule.

[‡] The symbol R* denotes heterocyclic substituents that do not significantly affect substrate reactivity during benzoannulation. No definition is provided for these substituents.



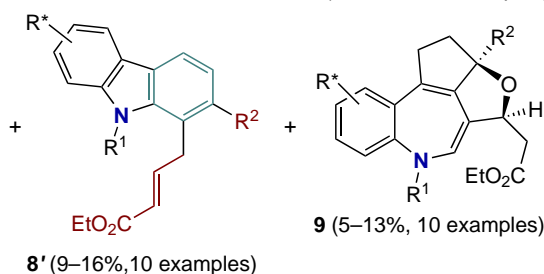
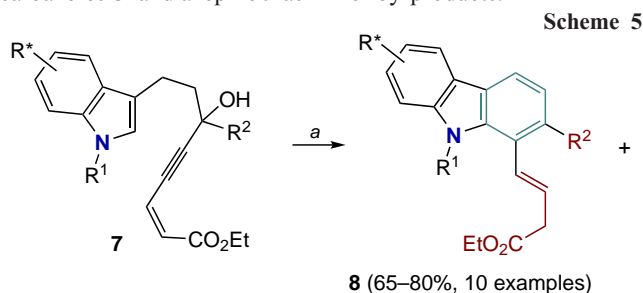
(a) $[\text{Mn}(\text{CO})_5\text{Br}]$ (0.1 equiv.), KOAc (2.4 equiv.), pyridine N-oxide (2 equiv.), 1,4-dioxane, 120°C , N_2 , 24 h; $\text{X} = \text{C}(\text{C}(\text{O})\text{R})_2$, O, NTs, NMs, $\text{C}(\text{CN})_2$; Py is pyridin-2-yl, Ts is *p*-toluenesulfonyl, Ms is methanesulfonyl (mesyl)



(a) TsOH (0.25 equiv.), DCM, 55°C , 5–18 h; DCM is dichloromethane; $\text{R}^1 = \text{Alk}$, Ar; $\text{R}^2 = \text{Alk}$;

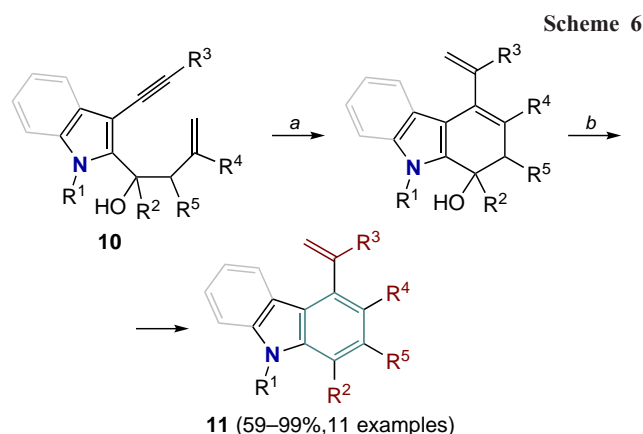
acid-catalyzed cyclization of propargyl alcohols in 77–91% yields (see Scheme 4).⁷² This method allows variation in the C2-alkyl substituent.

Introducing an additional vinyl group into compound **5** (to give compound **7**) opens an access to carbazoles **8** with an alkenyl substituent at the C1 position (Scheme 5).⁷³ The yields of the products are 65–80%. The reaction produces isomeric carbazoles **8'** and azepine **9** as minor by-products.



(a) MsOH (0.1 equiv.), DCM, 55°C , 2 h; $\text{R}^1, \text{R}^2 = \text{Alk}$

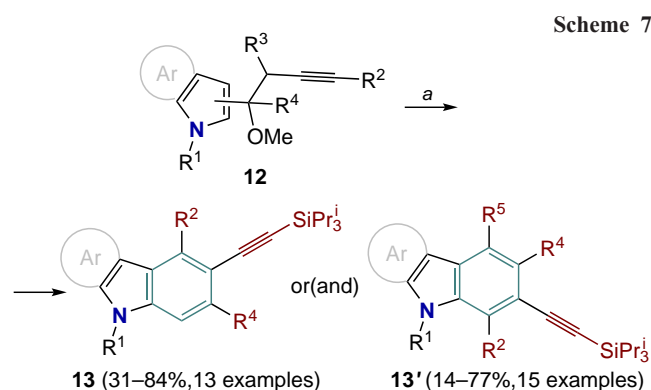
The synthesis of indoles and carbazoles **11** with an alkenyl-functionalized C4 position of the benzene ring was carried out through the intramolecular metathesis of enyne **10**, catalyzed by the Grubbs first-generation catalyst, followed by dehydration (Scheme 6).^{8,74} This method requires the presence of arylsulfonyl substituents at the nitrogen atom and allows to obtain products bearing additional substituents at positions C1–C3 in yields ranging from 59% to 99%.



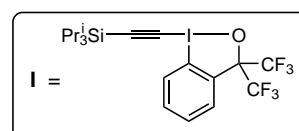
(a) 1st generation Grubbs catalyst (0.075 equiv.), PhMe, 80°C , 12 h, ethylene gas; (b) TsOH·H₂O (0.1 equiv.), rt, 1 h; $\text{R}^1 = \text{SO}_2\text{Ar}$; $\text{R}^2, \text{R}^4 = \text{H}, \text{Me}$; $\text{R}^3 = \text{H}, \text{Alk}, \text{Ar}$; $\text{R}^5 = \text{H}, \text{Me}, \text{CO}_2\text{Me}$

Alkynylated indoles, carbazoles and thieno[3,2-*b*]indoles **13**, **13'** were obtained *via* the platinum(II) bromide-catalyzed cyclization of 1-hetarylbut-3-in-1-ol derivatives **12** in the presence of the benziodoxol derivative **I** (Scheme 7).⁸¹ The alkynyl substituent can be located at position C2 or C3, depending on the structure of the starting heterocycle. For a number of 2-substituted substrates with aromatic substituents at the triple bond, the simultaneous formation of C2- and C3-substituted BAHs was observed, with the C3-substituted BAHs predominating. The yields of the products of this reaction for most substrates are good, reaching 84%. The possibility of desilylation while forming the ethynyl substituent was demonstrated.

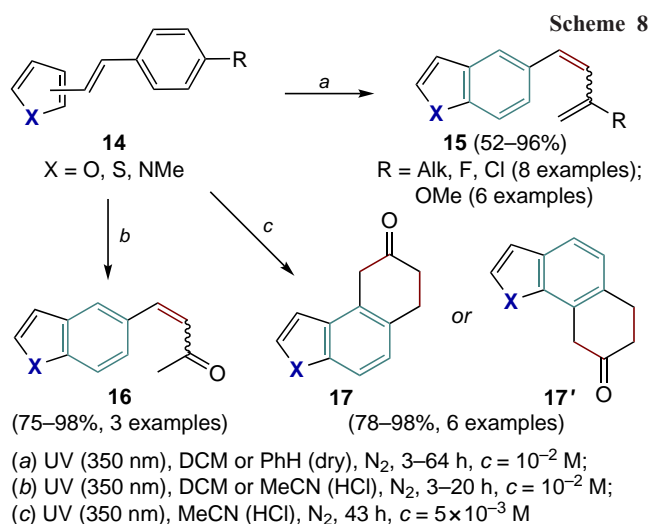
Photorearrangements of 1,2-diarylethenes **14** under inert conditions can be used to access benzo[*b*]thiophenes, benzo[*b*]furans and indoles with different alkyl and alkenyl substituents (Scheme 8). These reactions proceed through a UV-induced



(a) PtBr_2 (0.1 equiv.), **I** (2 equiv.), NaHCO_3 , THF–MeCN (5:1), 72 – 120°C ; $\text{R}^1 = \text{Alk}$, Ar; $\text{R}^2, \text{R}^5 = \text{H}, \text{Alk}, \text{Ar}$; $\text{R}^3, \text{R}^4 = \text{H}, \text{Alk}$; rt is room temperature

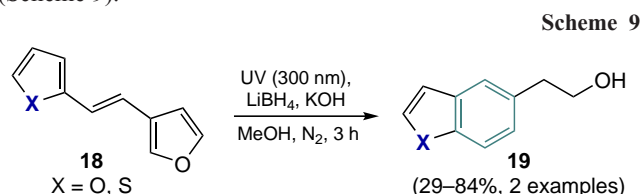


[§] Examples of the preparation of alkenyl-substitute BAHs are also given in other Sections (the position of the alkenyl group within the benzene ring is indicated): these include C3-substituted carbazoles (Scheme 17),⁷⁵ C1-substituted indoles (Scheme 46),⁷⁶ C2- or C3-substituted carbazoles (Schemes 117⁷⁷ and 56⁷⁸), C1-substituted indoles (Scheme 173)⁷⁹ and C1-substituted carbazole (Scheme 174).⁸⁰



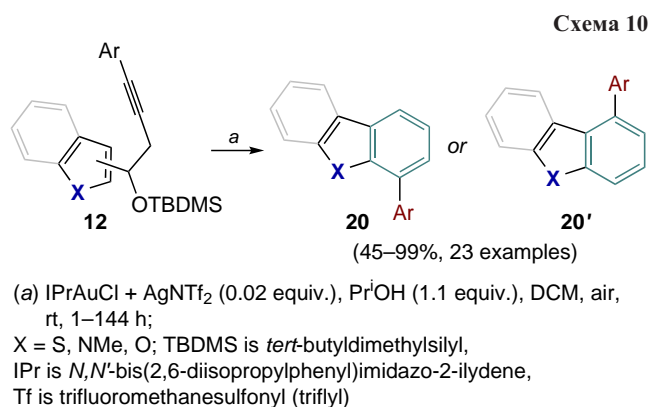
6 π -electrocyclization step. Depending on the structure of the substrates (*i.e.* the nature of the substituents in the benzene ring) and the conditions (*i.e.* the presence of an acid), BAHs substituted with butadienyl (**15**),^{82,83} or alkenyl (**16**)⁸³ groups are formed, as well as products (**17**) with a cyclohexanone moiety annulated at the C1, C2 or C3, C4 atoms.⁸⁴

In a related process involving the furan ring-opening in substrates **18** in the presence of a base and a reducing agent, BAHs **19** with a C3-hydroxyethyl group were obtained (Scheme 9).⁸⁵

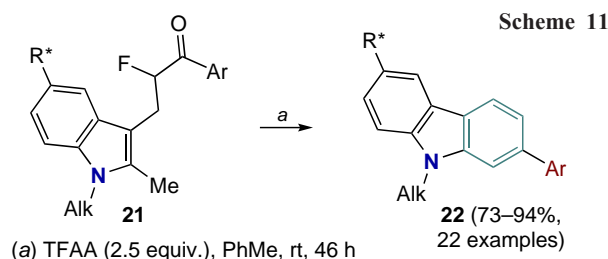


2.2. Aryl-substituted heterocycles

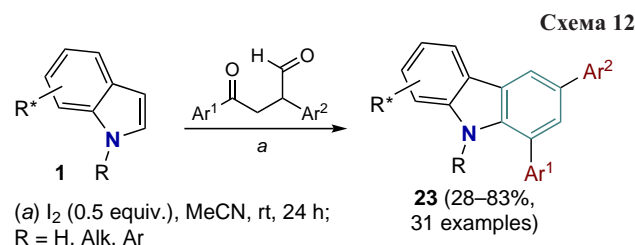
The synthesis of BAHs with one or two aryl substituents in the benzene ring has been reported in the literature. For example, the gold(I)-catalyzed cyclization of homopropargyl ethers **12** derived from a variety of 2-substituted heterocycles (*e.g.* furan, benzo[*b*]furan, thiophene and benzo[*b*]thiophene), affords compounds **20** with various aryl substituents at position C1, with yields ranging from 45% to 99% (Scheme 10).^{86,87} Interestingly, the cyclization of both 2- and 3-substituted indoles gives C4-substituted carbazoles **20'**. In the case of one of the benzo[*b*]thiophene-based substrates, two isomeric (at the C1 or C4 positions) products were isolated in equal amounts.



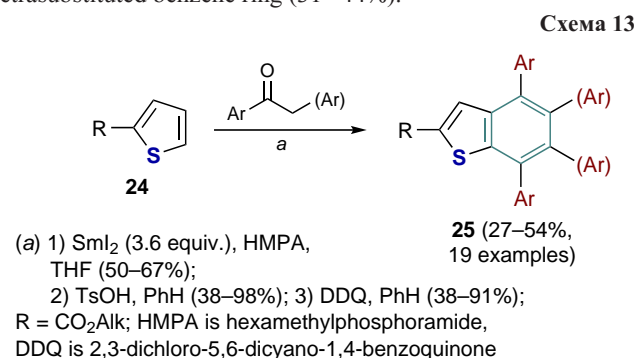
2-Substituted carbazoles **22** were obtained by the cyclization of 2-fluoro-3-(indole-3-yl)propanones **21** in the presence of trifluoroacetic anhydride (TFAA) (Scheme 11).⁸⁸ The target compounds containing benzene (including those bearing electron-donating groups), naphthalene, and benzo[*b*]furan moieties as aromatic substituents were obtained in high yields.



1,3-Diarylcarbazoles **23** were synthesized from 2,3-unsubstituted indoles **1** and β -formylketones in the presence of a catalytic amount of molecular iodine (Scheme 12).⁸⁹ This method produces carbazoles in yields ranging from 28% to 83%.



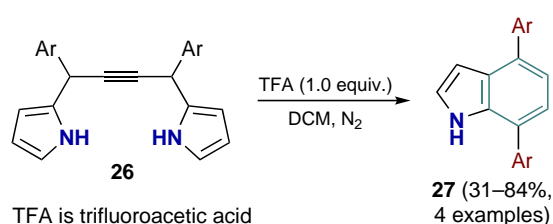
The synthesis of BAHs containing a 1,4-diaryl-substituted benzene moiety can be accomplished in two different ways. The first approach involves a three-step transformation of thiophene-2-carboxylates **24** into benzo[*b*]thiophenes **25** (Scheme 13). This process includes SmI_2 -promoted electrophilic substitution reactions, dehydration in an acidic medium and 6 π -electrocyclization of the hexatriene system, followed by oxidation by DDQ.⁹⁰ The C1 and C4 aromatic substituents can be the identical or different derivatives of benzene and thiophene. The product yield over three steps ranges from 27% to 54%. The same method provides an access to benzothiophenes with tetrasubstituted benzene ring (31–44%).



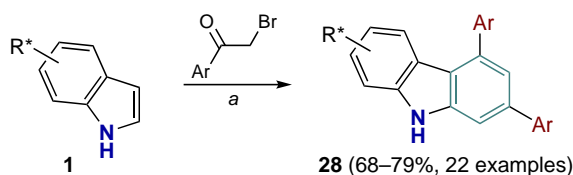
The synthetic approach to C1,C4-diaryl-substituted indoles **27** is based on the acid-catalyzed rearrangement of 1,4-diaryl-1,4-dipyrrolylbutynes **26** (Scheme 14).⁹¹ This reaction was performed with substrate loadings of dozens of milligrams, with yields ranging from 31% to 84%, and its high efficiency has been proven by NMR monitoring.

A method for the synthesis of carbazoles **28** with C2 and C4 aryl substituents, based on the aerobic reaction of 2,3-di-

Scheme 14



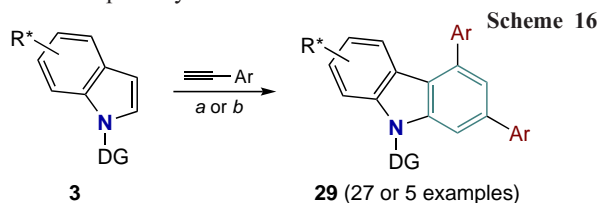
Scheme 15



(a) DMAP (0.5 equiv.), MeCN, air, 60°C, 0.4–4 h;
DMAP is 4-dimethylaminopyridine

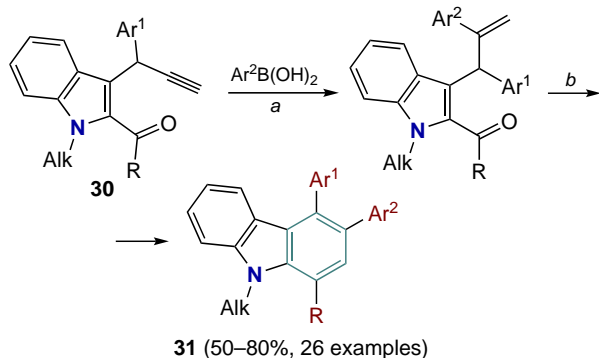
substituted indoles with α -bromoacetophenone derivatives in the presence of DMAP (Scheme 15).⁹² This method delivers carbazoles **28** bearing electron-donating or electron-withdrawing substituents rapidly and regioselectively, with yields ranging from 68% to 79%.

Target carbazoles **29** were obtained from 2,3-unsubstituted indoles and arylacetylenes using rhodium(III)⁹³ or manganese(I) catalysts (Scheme 16).⁹⁴ The first case implies a much broader range of substrates and higher yields (31–89% vs. 14–29%). Substituted benzenes (including those with electron-deficient substituents) and thiophene derivatives can act as aromatic substituents. Both methods require the presence of the directing pyrimidinyl substituent in the indole, which can be removed by base in an acceptable yield.



(a) [Cp*Rh(MeCN)₃][SbF₆]₂ (0.1 equiv.), 1,4-dioxane, argon, 80°C, 12 h (31–89%); (b) MnBr(CO)₅ (0.1–0.2 equiv.), DIPEA (0.2–0.4 equiv.), Et₂O, argon, 80°C, 24 h (yield 14–29%); DG (directing group) = pyrimidin-2-yl, Cp* is pentamethylcyclopentadienyl, DIPEA is diisopropylethylamine

Scheme 17

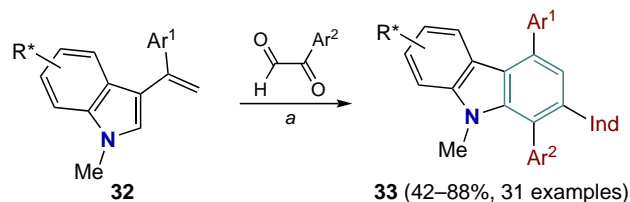


(a) Pd(PPh₃)₄ (0.05 equiv.), AcOH or 1,4-dioxane, 80°C, 4–8 h;
(b) TfOH (0.2 equiv.), rt, 0.5 h; R = H or Ar

Di- and triaryl-substituted carbazoles **31** were obtained by reacting 2-carbonyl-3-propargylindoles **30** with arylboronic acids (Scheme 17).⁷⁵ This reaction, carried out in a one-pot fashion, involves sequential steps of palladium(0)-catalyzed hydroarylation of the triple bond and acid-catalyzed cyclization. Carbazoles with C3, C4-diaryl and C1, C3, C4-triaryl-substituted benzene rings were obtained. The yields of the corresponding products ranged from 50% to 80%. Benzene and thiophene derivatives were used as aromatic substituents.

Carbazoles with C1-, C2- and C4-triaryl-substituted benzene rings are accessible through two different approaches. A one-pot variant of acid-catalyzed annulation involving intramolecular nucleophilic addition to phenylglyoxal derivatives, substitution and cyclization was developed starting from 3-vinylindoles **32** (Scheme 18).⁹⁵ Ninety-five Brønsted acids based on phosphoric acid were employed as catalysts, while benzene and indole derivatives served as aromatic substituents. The yields of carbazoles **33** ranged from 42% to 88%.

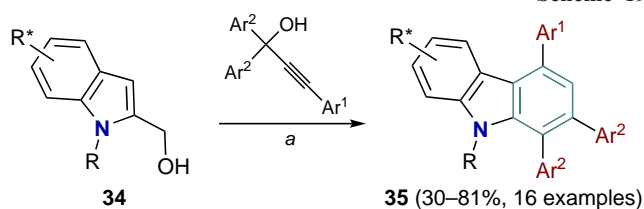
Scheme 18



(a) 1) PhMe, 60°C, 6 h; 2) indole (0.8–1.0 equiv.), diarylphosphoric acids (0.1–0.2 equiv.), H₂O, PhMe, 80°C, 12 h; Ind is indole-3-yl

Triaryl-substituted carbazoles **35** were synthesized from indolylmethanol **34** and propargyl alcohols in the presence of a Lewis acid catalyst (Scheme 19).⁹⁶ This cascade reaction involves the indole allenylation, a [1,5]-hydride shift or 6 π -electrocyclization, and the Wagner–Meerwein rearrangement. Various benzene derivatives were used as aryl substituents, and the product yields ranged from 30% to 81%.

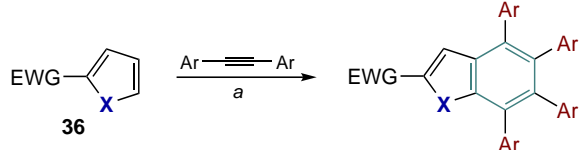
Scheme 19



(a) Yb(OTf)₃ (0.1 equiv.), DCE, Δ , 18 h; R = H, Me, propargyl; DCE is 1,2-dichloroethane

Tetraaryl-substituted BAHs **37** were obtained via palladium(II)-catalyzed aerobic annulation of 2,3-unsubstituted furans or pyrroles **36** containing electron-withdrawing

Scheme 20



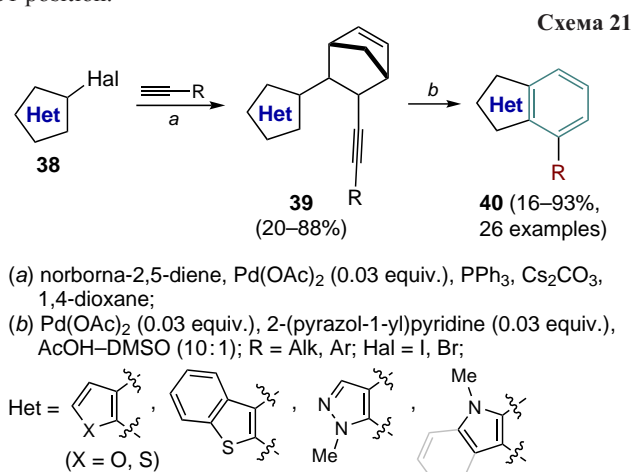
(a) Pd(OAc)₂ (0.05 equiv.), RCO₂H (1.0 equiv.), RCO₂Na (1.0 equiv.), DMF, O₂, 120°C; X = O, NAlk; EWG (electron-withdrawing group) = CO₂R, C(O)NR₂, C(O)R, C(O)H, CN; R = Alk

substituents, with diarylalkynes (Scheme 20).⁹⁷ The target benzo[*b*]furans and indoles were isolated in yields ranging from 9% to 84%.

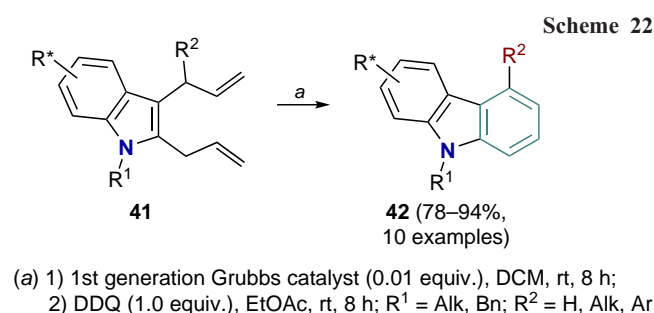
2.3. Alkyl- or aryl-substituted heterocycles

This Section highlights synthetic approaches to benzoannulated heterocycles, enabling products with both alkyl and aryl substituents in the resulting aromatic moiety. The material is organized according to the degree of increase in the number of such groups, ranging from monosubstituted to tetrasubstituted.

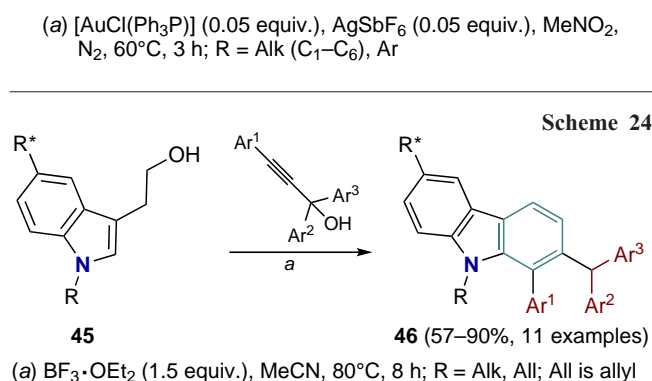
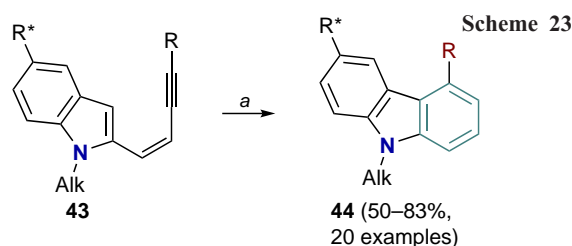
Monosubstituted benzoannulated heterocycles can be synthesized in three ways. A series of C1-substituted compounds **40** (benzothiophenes, dibenzothiophenes and indazoles, as well as individual representatives of the indole, carbazole and benzofuran families) were obtained in two steps from the corresponding halogenated heterocycles **38** via norbornene-substituted intermediates **39** (Scheme 21).⁹⁸ Both steps are based on palladium(II)-catalyzed coupling and annulation reactions. The product yields at these steps range from acceptable to high. The method was tested using benzothiophenes containing a variety of aromatic and aliphatic substituents at the C1 position.



The synthesis of BAHs with aryl or alkyl substituents at the C4 position can be carried out in two ways. Reactions involving the metathesis of 2,3-diallylindoles **41**, followed by oxidation deliver carbazoles **42** with a methyl or phenyl substituent at the C4 atom (one example involving a C3-tethered methyl group is also known) (Scheme 22).⁹⁹ The yields over the two stages ranged from 61% to 95%. Effective debenzoylation of the protected nitrogen atom was demonstrated for these compounds.



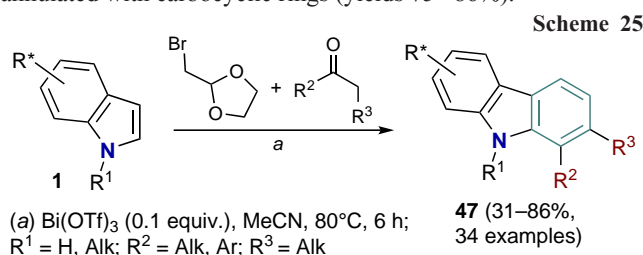
Carbazoles **44** with a C4 substituent in the benzene ring are obtained by the gold(I)-catalyzed intramolecular hydroarylation of (*Z*)-2-(enynyl)indoles **43** (Scheme 23).¹⁰⁰ Most examples used alkyl substituents, but one example of a *p*-tolyl-substituted



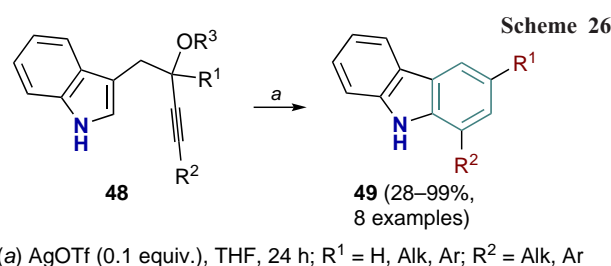
carbazole has been described. The product yields were in the range of 50–83%.

Benzoheterocycles with two substituents (aryl and/or alkyl) in the benzene ring can also be obtained in several ways. For example, carbazoles substituted at the C1 and C2 positions were synthesized by the reaction of 3-(2-hydroxyethyl)indoles (tryptophol, **45**) with propargyl alcohols in the presence of boron trifluoride etherate in 57–90% yields (Scheme 24).¹⁰¹ Products **46** contain a C1-aryl group at the position and a C2-diarylmethyl group.

The synthesis of carbazoles **47** bearing alkyl and aryl substituents at the C1 and/or C2 positions, *via* bismuth(III) triflate-catalyzed annulation of 2,3-unsubstituted indoles **1** with acetals of α -bromoacetaldehyde and ketones is described (Scheme 25).¹⁰² Methyl, aromatic, and heteroaromatic groups were used as substituents. The yields of such products ranged from 31% to 86%. In addition, cyclic ketones and 1,3-dicarbonyl compounds were used in the reaction to give carbazoles annulated with carbocyclic rings (yields 75–86%).

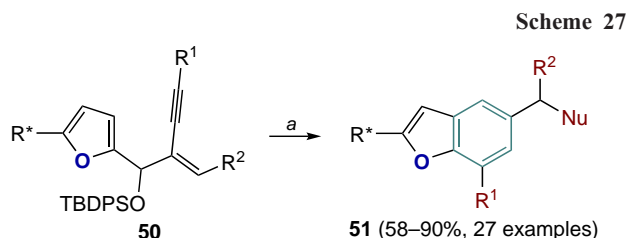


Carbazoles **49**, substituted at the C1 or C1 and C3 positions, were obtained by the silver(I) triflate-promoted cyclization of 1-(indol-3-yl)but-3-en-2-ols derivatives **48**, which are based on



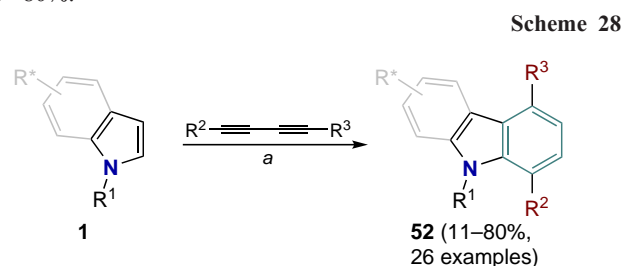
propargyl alcohol (Scheme 26).¹⁰³ Products with alkyl, cycloalkyl, aryl and thienyl substituents were isolated in yields ranging from 28% to 99%.

1,3-Disubstituted benzo[*b*]furans **51** were obtained *via* a gold(I)-catalyzed cascade reaction of functionalized furans **50** followed by the addition of nucleophiles, such as water, alcohols, aromatic amines and 1,3,5-trimethoxybenzene (Scheme 27).¹⁰⁴ The yields of benzo[*b*]furans **51** ranged from 58 to 90%.



(a) [JohnPhos(MeCN)AuSbF₆] (0.05 equiv.), NuH (5 equiv.), THF, rt or 50–80°C, 10 min–24 h;
R¹ = Alk, Ar; R², R³ = H, Alk, Ar; Nu = OH, OR, NHAH;
TBDPS is *tert*-butyldiphenylsilyl, NuH is nucleophile,
JohnPhos is 2-(*di-tert*-butylphosphino)biphenyl

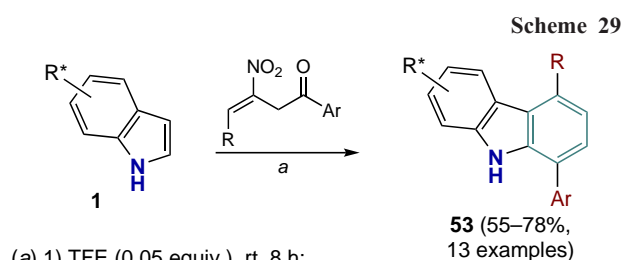
1,4-Disubstituted indoles and carbazoles **52** were synthesized by a gold(I)-catalyzed [4+2] reaction of 1,3-diynes with 2,3-unsubstituted pyrroles and indoles **1** (Scheme 28).¹⁰⁵ The reaction with symmetrical 1,3-dienes proceeds selectively, whereas in the case of nonsymmetrical substrates, formation of separable regioisomers is possible. The product yields are 11–80%.



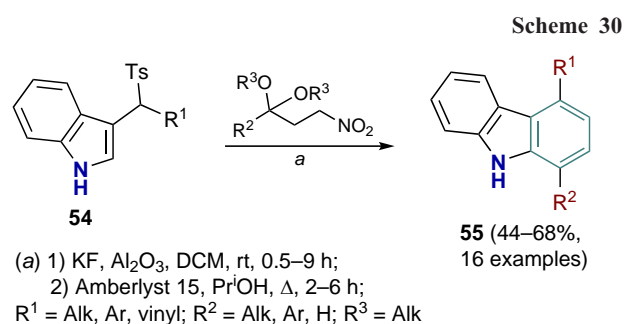
(a) [BrettPhosAu(MeCN)SbF₆] (0.05 equiv.), DCE, 80°C, 3–24 h;
R¹ = H, Alk, All; R², R³ = Alk, Ar; BrettPhos is 2-dicyclohexylphosphino-3,6-dimethoxy-2',4',6'-tri(isopropyl)biphenyl

Carbazoles **53**, functionalized with a C1-aryl group and a C4-alkyl group, were synthesized by a one-pot method starting from 2,3-unsubstituted indole **1** and β -nitro- β,γ -unsaturated ketone (Scheme 29).¹⁰⁶ Depending on the structure of the ketone, only C1-aryl-substituted carbazoles can be obtained, with yields ranging from 55% to 78%.

The two-step reaction of sulfonyl-substituted indoles **54** with protected β -nitro ketones gives C4-substituted and C1,C4-

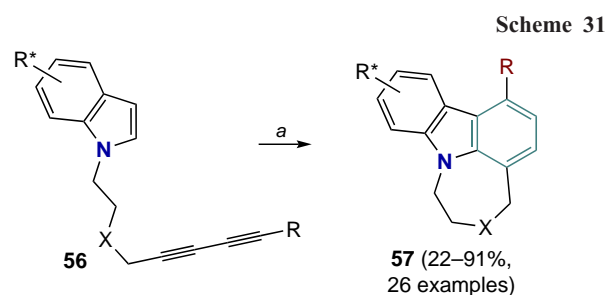


(a) 1) TFE (0.05 equiv.), rt, 8 h;
2) Amberlyst 15, 2-MeTHF, 100°C, MW, 3 h;
R = H, Alk; TFE is trifluoroethanol



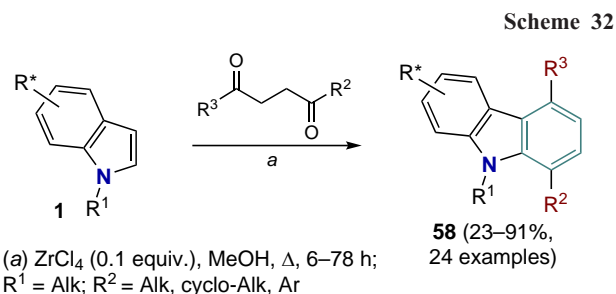
disubstituted carbazoles **55** (Scheme 30).¹⁰⁷ The yields of these products are 44–68%.

Another approach to carbazoles **57** with alkyl and aryl substituents at the C1 and C4 positions involves the palladium(II)-catalyzed intramolecular annulation of indolyl-substituted diynes **56** (Scheme 31).¹⁰⁸ A cycloalkyl moiety attaches to the C1 position of the products, while the alkyl and aryl substituents can be at the C4 atom, which is determined by the nature of the substrate. The yields of the target carbazoles range from 22% to 91%. One of the products was obtained in gram quantities.



(a) [(dppp)Pd(OTf)₂] (0.05 equiv.), DCE, BzOH or PivOH, air, 70 or 150°C, 21 h; X = NSO₂R, NTs, O, (CH₂)₂;
R = Alk, cyclo-Alk, Ar; Piv is pivaloyl

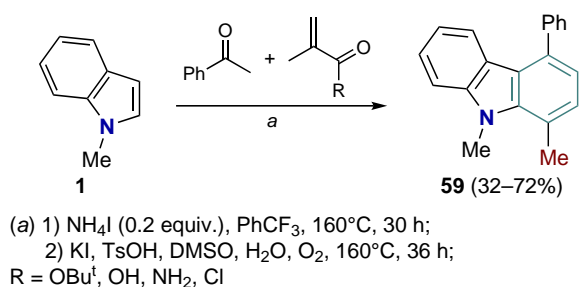
Alkylaryl-substituted carbazoles **58** can also be obtained *via* benzoannulation of 2,3-unsubstituted indoles **1** with diketones, in the presence of a Lewis acid (Scheme 32).⁷⁰ The resulting products always have the C1-alkyl substituent and can have both alkyl and aryl groups at the C4 position, depending on the diketone structure. The yields of the target carbazoles range from 23% to 91%. Synthesis of 1,4,5,8-tetramethyl-9*H*-carbazole demonstrates the applicability of this method to pyrrole derivatives.



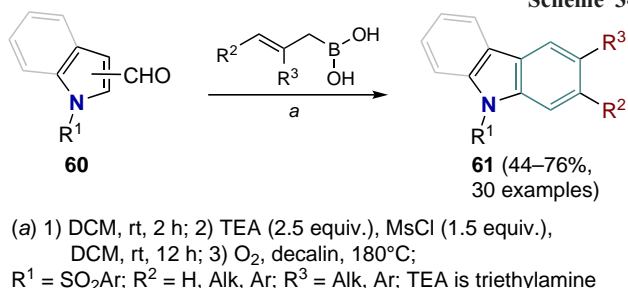
The annulation of 2,3-unsubstituted 1-methylindole with acetophenone and methacrylic acid derivatives produced C1,C4-disubstituted carbazole **59** (Scheme 33).¹⁰⁹

A three-step, one-pot method has been developed for the benzoannulation of pyrrole and indolecarboxaldehydes **60** with allylboronic acids to furnish C2- or C3-substituted products **61** (Scheme 34).¹¹⁰ This method provides an access to target

Scheme 33



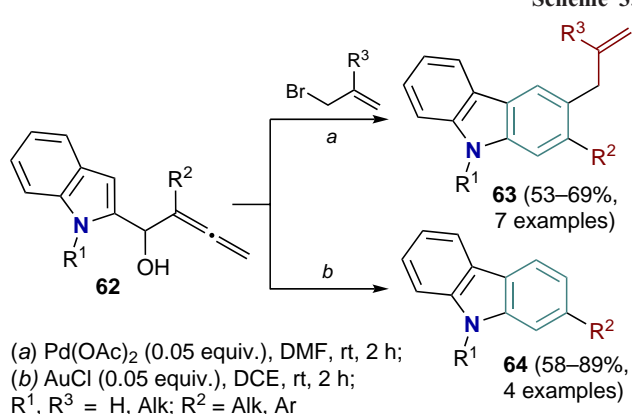
Scheme 34



compounds with both alkyl (or cycloalkyl) and aromatic (benzene, thiophene, and indole derivatives) substituents. The possibility of preparative (gram-scale) synthesis has also been demonstrated for this protocol, with alkyl- and aryl-substituted indoles and carbazoles being isolated in yields ranging from 44% to 76%.

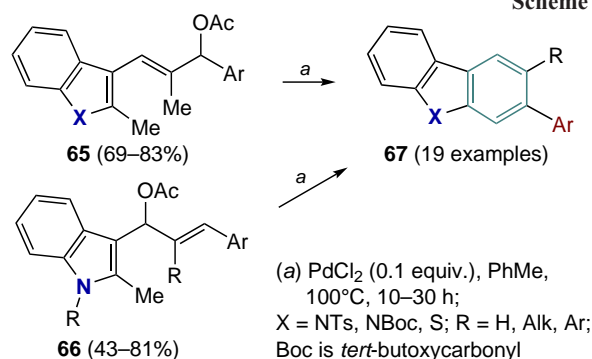
Carbazoles with a C2-substituted or C2,C3-disubstituted benzene ring are accessible *via* annulation of indole-tethered allenols **62** (Scheme 35).¹¹¹ The palladium(II)-catalyzed reaction using allyl bromides gives carbazoles **63** with C2 and C3 substituents in 53–69% yields. The substituents at the C2 atom of the products are methyl or phenyl groups and the substituents at the C3 atom are alkyl moieties. When gold(I) chloride is used as a catalyst instead of a palladium(II) catalyst in the absence of the bromine derivative, C2-substituted carbazoles **64** are obtained (yields 58–89%).

Scheme 35



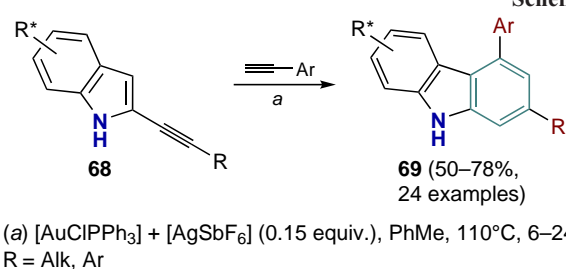
C2- and C3-substituted carbazoles can be obtained in two ways. Both routes involve palladium(II) chloride-catalyzed cyclization using different substrates **65** ($\text{X} = \text{NTs}$, NBoc) and **66**, which contain a vinyl or allyl group, and an acylated hydroxy group in the α -position to the double bond (Scheme 36).¹¹² Benzothiophenes **65** ($\text{X} = \text{S}$) are also involved in this reaction. The yields of carbazoles and dibenzothiophenes **67** are 43–83%.

Scheme 36



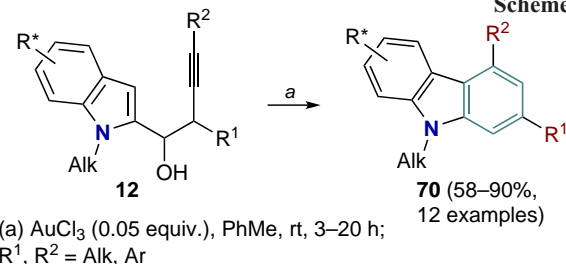
A simple and effective synthetic approach to carbazoles **69** involves treating 2-alkynylindoles with arylacetylenes in the presence of a bimetallic gold(I)–silver(I) catalyst (Scheme 37).¹¹³ The fundamental feature of this strategy is the sequential activation of terminal and internal triple bonds. The versatility of this method was demonstrated on the example of the reaction between various 2-alkynylindoles and arylacetylenes.

Scheme 37

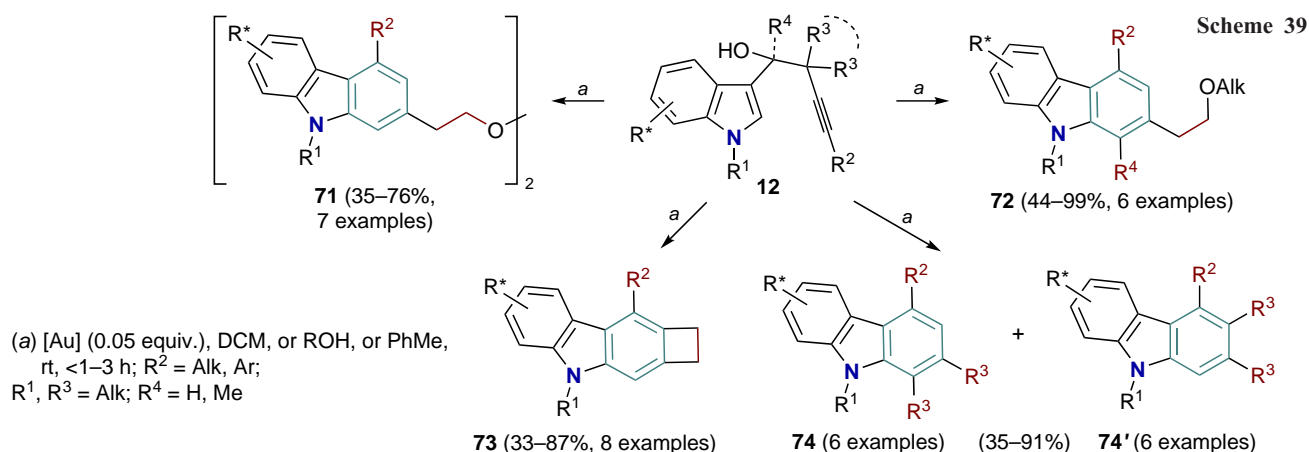


Various methods for the synthesis of carbazoles bearing alkyl and aryl substituents in different positions from 1-(indol-2-yl)but-3-in-1-ols **12** have been reported. For example, the gold(III) chloride-catalyzed cyclization of 1-(indol-2-yl)but-3-in-1-ols is described (Scheme 38).¹¹⁴ This process gives rise to C2,C4-disubstituted carbazoles **70** with alkyl and aryl substituents in 58–90% yields.

Scheme 38

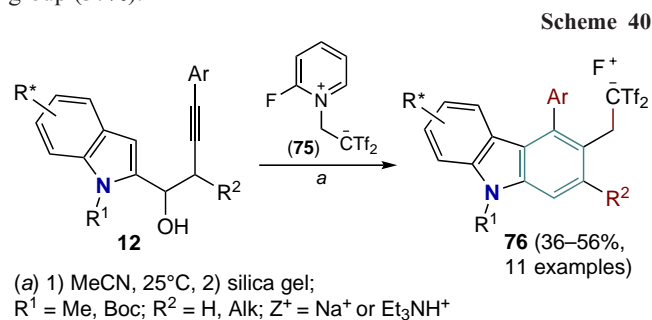


The cyclization of 1-(indol-2-yl)but-3-in-1-ols **12**, catalyzed by gold(I) complexes (such as $[\text{Au}(\text{MeCN})(\text{JackiePhos})][\text{SbF}_6]$ and two specially obtained analogues), can give various carbazoles (Scheme 39).¹¹⁵ Depending on the substrate, catalyst and solvent used, carbazoles **71–74** were obtained, with the benzene ring being substituted at positions C1, C2, C4 or C2, C3 and C4. In the presence of a specially synthesized gold(I) complex based on a sterically hindered phosphine in dichloromethane, cyclobutene-annulated carbazoles **73** were obtained (33–87%). When dichloromethane is replaced with toluene, the main products are dimeric ethers **71** (yields 35–76%), while replacing it with alcohol, the main products are derivatives **72** (yields 44–99%). Using another gold(I) complex as a catalyst produces an inseparable mixture of isomers (yields ranging from 35 to 91%), with the main product being carbazoles

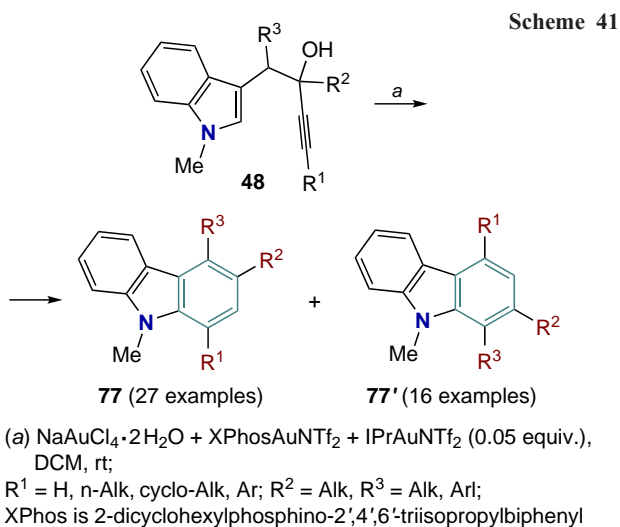


74 and the by-product being carbazoles 74'. C4-aryl groups act as additional substituents in products 71–74.

The bis(triflyl)ethylation of the indole ring of indole-2-ylbut-3-in-1-ol 12 with salts 75, followed by treatment with silica gel affords bis(triflyl)ethyl salts of carbazoles 76, in 36–56% yields (Scheme 40).¹¹⁶ The resulting products contain alkyl and aryl groups at the C2, C3 and C4 positions. One example demonstrates that introducing an iodine atom into the 3-position of indole gives rise to a carbazole containing a C1-hydroxy group (37%).

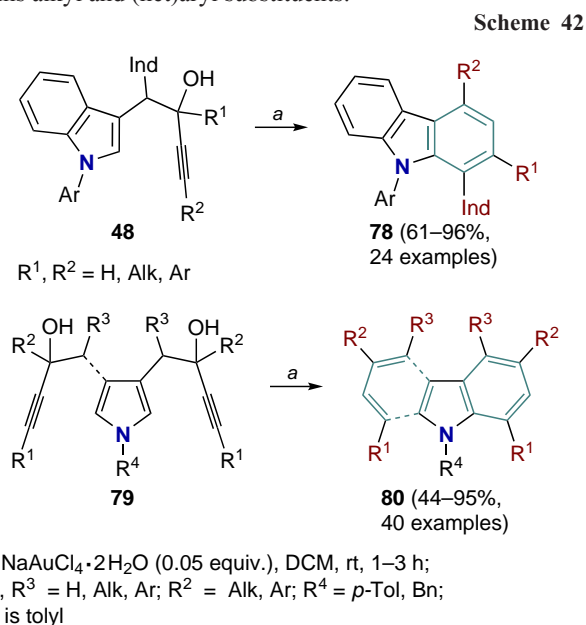


Trialkylaryl-substituted carbazoles are obtained by the cyclization of 1-(indol-3-yl)but-3-in-2-ols 48 in the presence of a gold(I or III) catalyst (Scheme 41).¹¹⁷ This method produces mixtures of isomers 77 and 77'. The selectivity of the reaction depends on the nature of both the catalyst and substrate. In some



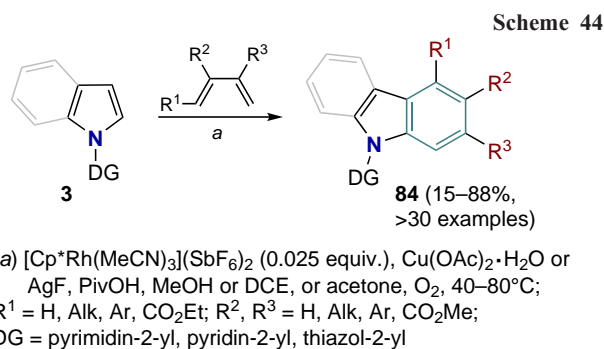
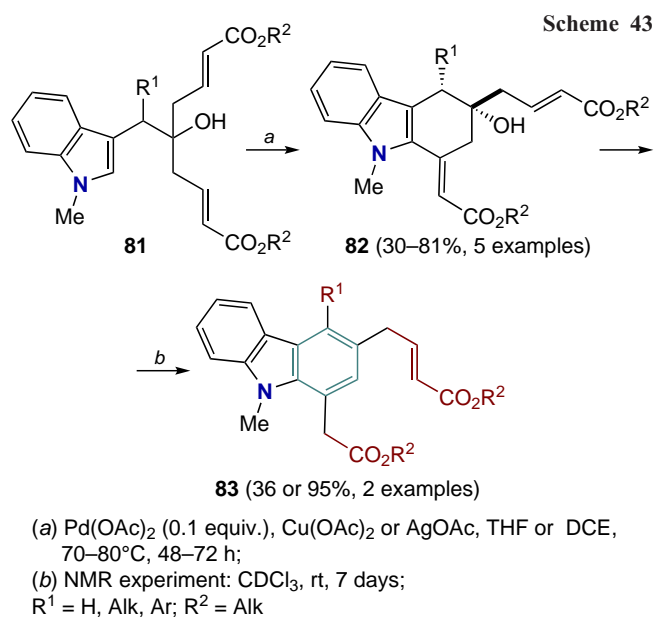
cases, it is possible to obtain one of the isomers with good selectivity. The product yields vary from 48% to 90%.

The gold(III)-catalyzed cycloisomerization of related indoles 48 and pyrroles 79 delivers carbazoles 78 (Refs 118 and 119) and indoles 80 (see Ref. 119) (Scheme 42). Introducing two propargyl alcohol moieties into the pyrrole molecule enabled a rare example of double benzoannulation to afford carbazoles. Using indoles as substrates, carbazoles 78 with substituents at positions C1, C2 and C4 are obtained (61–96%), whereas the reaction with pyrroles gives products 80 with substituents at positions C1, C3 and C4 (44–95%). The resulting benzene ring contains alkyl and (het)aryl substituents.



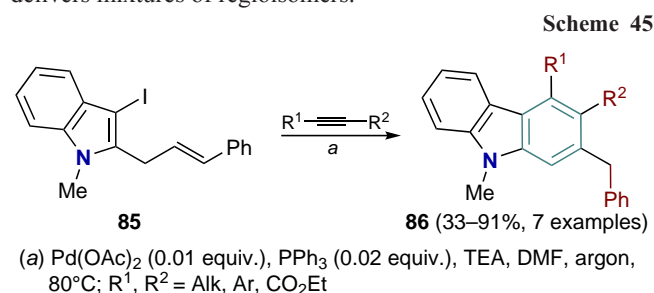
Catalysis with palladium(II) acetate promotes cyclization of indole derivatives 81 tethered with a bis(homoallylic) alcohol moiety. This produces compounds 82 (yields up to 81%), which undergo aromatization to give carbazoles 83 with substituents at positions C1, C3 and C4 (R¹ ≠ H) (Scheme 43).¹²⁰ Two carbazoles were obtained using this method, but it was primarily used for the synthesis of naphthalene derivatives.

Carbazoles and indoles 84, which have different alkyl and aryl substituents at the C2, C3 and C4 positions, can be obtained via a rhodium(III)-catalyzed reaction between 2,3-unsubstituted substrates and 1,3-dienes (Scheme 44).¹²¹ This method requires the use of nitrogen-containing directing groups at the nitrogen



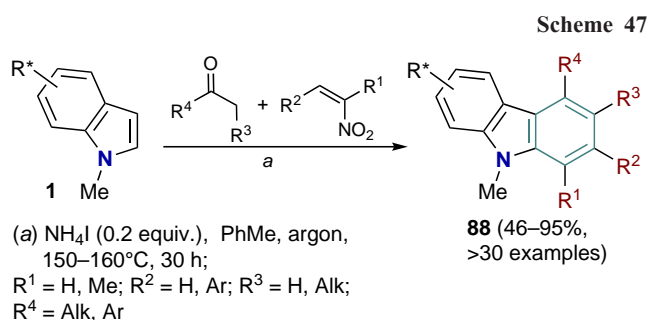
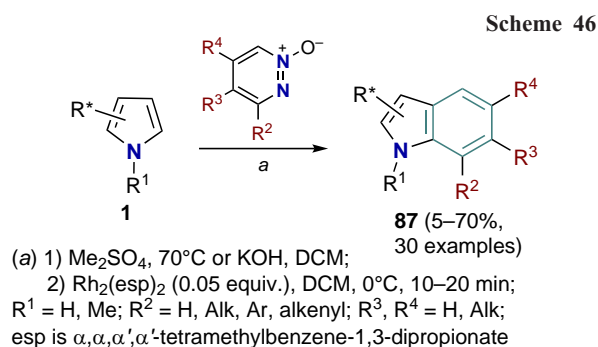
atom of the annulated *N*-heterocycle. Product yields range from 15% to 88%.

A method based on the annulation of 3-iodo-2-cinnamylindole **85** with alkynes, in the presence of palladium(II) acetate, has been proposed (Scheme 45).¹²² Carbazoles **86**, substituted with alkyl and aryl groups at positions C2, C3 and C4, were obtained in 33–91% yields. The reaction with unsymmetrical alkynes delivers mixtures of regioisomers.



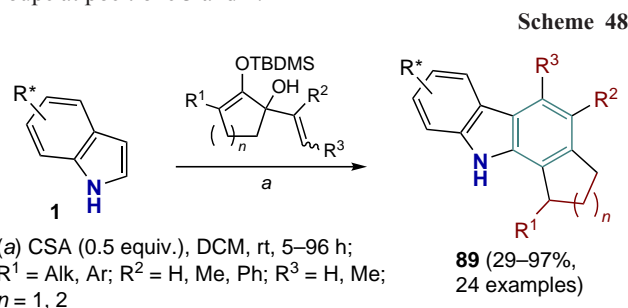
Indoles **87** were synthesized by the rhodium(II)-catalyzed reaction of 2,3-unsubstituted pyrroles with pyridazine *N*-oxides (Scheme 46).⁷⁶ The use of various pyridazine *N*-oxides gave rise to a series of C1-substituted aryl- and alkylindoles, as well as products bearing additional alkyl groups at the C2 and C3 positions.

Several examples of the synthesis of BAHs containing the substituted benzene ring have been reported. For example, the annulation of 2,3-unsubstituted indoles **1** with ketones and nitroalkynes, promoted by NH_4I , gives selectively poly-



substituted carbazoles **88** (Scheme 47).¹²³ The reaction products bear alkyl and aryl groups in various positions, and are isolated in yields ranging from 46 to 95%.

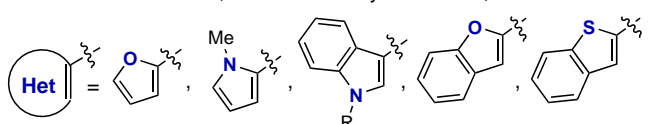
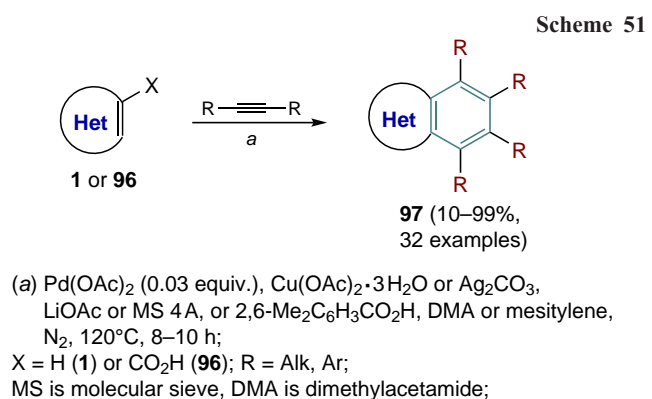
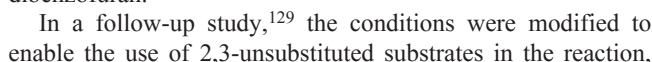
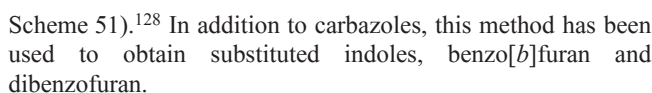
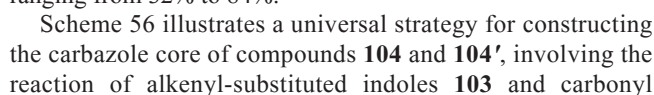
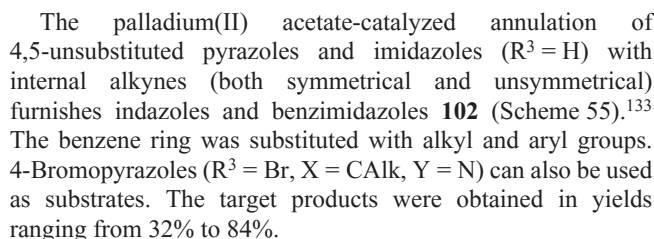
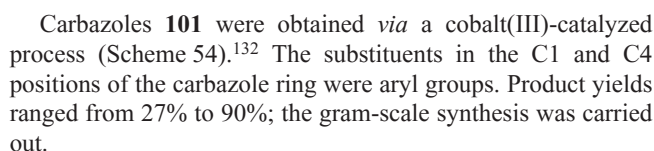
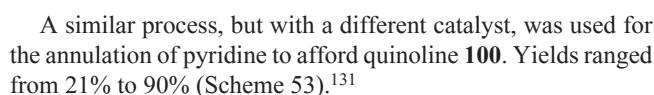
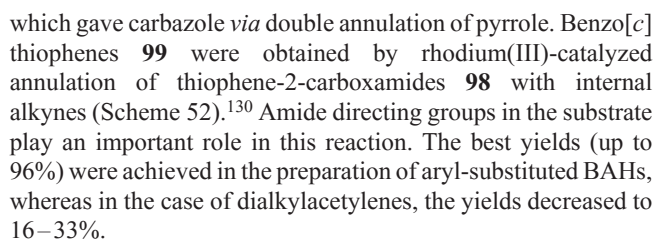
Carbazoles **89** were obtained from 2,3-unsubstituted indoles **1** and cyclic α -hydroxysilylenol ethers tethered with a vinyl group at the α -position, *via* treatment with camphorsulfonic acid (CSA) (Scheme 48).¹²⁴ Such products are annulated with carbocycles at positions 1 and 2 and may contain alkyl or aryl groups at positions 3 and 4.

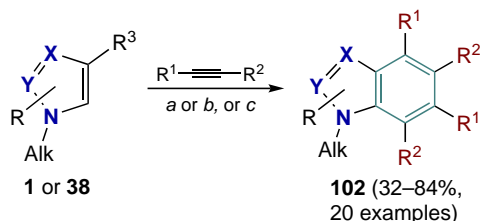


Carbazoles bearing two, three or four substituents were obtained by the reaction of pyrroles¹²⁵ or indoles¹²⁶ with 1-vinylpropargyl alcohols **90**, in the presence of a ruthenium(0) complex **II** and trifluoroacetic acid (Scheme 49). The reaction produces a mixture of two isomeric products (**91** and **91'**), which can be separated by column chromatography. The regioselective formation of substituted carbazoles was demonstrated in a number of cases, with yields ranging from low to high (5–92%).

Alkyl- and aryl-substituted indoles **93** with one to three substituents were obtained by metathesis of substrates **92** in the presence of Grubbs second-generation catalyst (yields 66–99%, Scheme 50).¹²⁷ This method can also be implemented to synthesize carbazole **95** *via* double benzoannulation of pyrrole **94**.

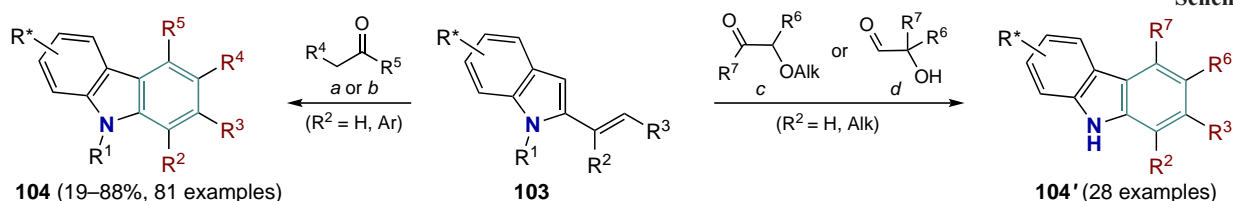
Disubstituted alkynes are widely used in the synthesis of BAHs with four aryl and/or alkyl substituents in the resulting benzene ring. For example, the palladium(II) acetate-catalyzed oxidative annulation of indole-3-carboxylic acids **96** with internal alkynes gives carbazoles **97** in 21–98% yields (see





- (a) $\text{Pd}(\text{OAc})_2$ (0.05 equiv.), $\text{Cu}(\text{OAc})_2 \cdot \text{H}_2\text{O}$, 1,4-dioxane, 120°C, 16 h;
 (b) $\text{Pd}(\text{OAc})_2$ (0.05 equiv.), NaOAc, AcOH, DMA, O_2 , 120°C, 16 h;
 (c) $\text{Pd}(\text{OAc})_2$ (0.05 equiv.), $\text{P}(\text{tBu})_3 \cdot \text{BF}_4$ (0.075 equiv.), K_2CO_3 (2.5 equiv.), PivOH (0.3 equiv.), PhMe, 120°C, 16 h;
 $\text{R}^3 = \text{H}$ (**1**) or Br (**38**); X = CAlk, Y = N; X = N: Y = CMe, CPh;
 $\text{R}^1, \text{R}^2 = \text{Alk, Ar}$

Scheme 55



- (a) 1) PhSO_2H (1.2 equiv.), $\text{TsOH} \cdot \text{H}_2\text{O}$ (0.1 equiv.), THF, 60°C, 1.2–1.5 h; 2) NaH (2.5 equiv.), DBU (0.1 equiv.), THF, rt, 0.3 h; 3) Pd/C (0.1 equiv.) or O_2 , decalin, 180°C, 24–72 h;
 (b) $(\text{PhO})_2\text{P}(\text{O})\text{OH}$ (0.15 equiv.), EtOAc, decalin, O_2 , 180°C, 48 h;
 (c) $\text{TsOH} \cdot \text{H}_2\text{O}$ (0.1 equiv.), EtOAc, 120°C, 2–96 h (yield 36–85%);
 (d) 1) $\text{TsOH} \cdot \text{H}_2\text{O}$ (0.2 equiv.), PhMe, 120°C, 12–48 h; 2) KOH, 90°C, 3 h (44–69%);
 $\text{R}^1 = \text{Alk, Ar}$; $\text{R}^3 = \text{H, Alk, Ar, alkenyl}$; $\text{R}^4 = \text{H, Alk, Ar, alkenyl}$; $\text{R}^5, \text{R}^6, \text{R}^7 = \text{H, Alk, Ar}$; DBU is 1,8-diazabicyclo[5.4.0]undec-7-ene

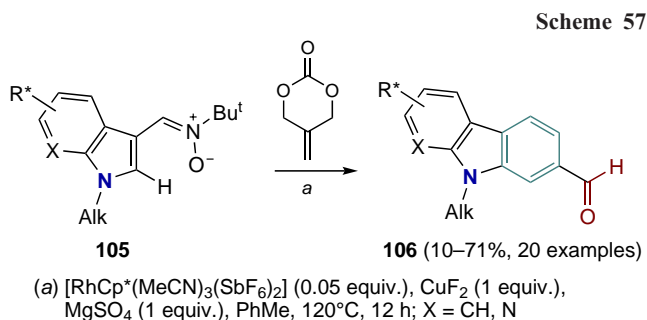
Scheme 56

compounds in the presence of Brønsted acids.^{78,134–137} This approach employs diversely substituted ketones and aldehydes, as well as their α -hydroxy derivatives. The target carbazoles contained one to four alkyl or aryl substituents on the benzene ring, and were isolated in yields ranging from 19% to 88%.

3. Benzoheterocycles with electron-withdrawing substituents

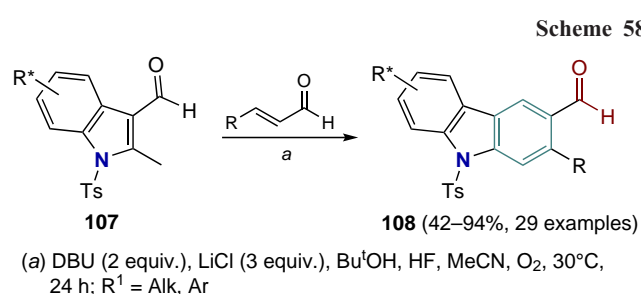
3.1. Formyl-substituted heterocycles

Two synthetic approaches to BAH-based aldehydes have been reported. The rhodium(III)-catalyzed reaction of indolyl nitrones (**105**, X = CH) with a cyclic, 2-methylidene-substituted carbonate gives carbazoles (**106**, X = CH) containing a C2-formyl group in yields ranging from 10% to 71% (Scheme 57).¹³⁸ In addition to carbazoles, this method delivers pyrido[2,3-*b*]indole-7-carbaldehyde **106** (X = N) in 52% yield. For one of the carbazoles, the possibility of scaling up the synthesis to produce gram quantities of the substrate was demonstrated.



Scheme 57

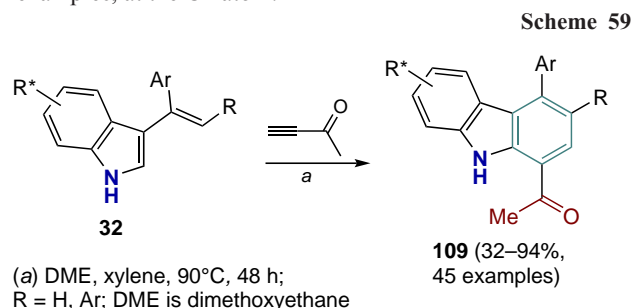
Carbazoles **108** with a C3-formyl group are formed by the reaction between *N*-tosylated 2-methylindole-3-carbaldehydes **107** and substituted enals under the action of the DBU–lithium chloride system (Scheme 58).¹³⁹ The C2 position of the products can be substituted with additional alkyl and aryl groups. The yields of carbazoles range from 42–94%.



Scheme 58

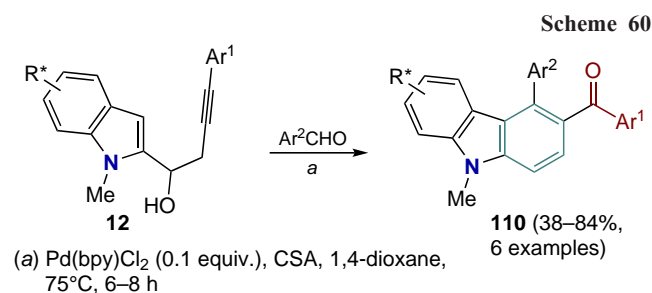
3.2. Acyl-substituted heterocycles

Several examples of the synthesis of BAHs containing one acyl group in the benzene ring have been reported. A method based on the Diels–Alder reaction between 3-alkenylindole derivatives **32** and acetylacetylene (but-3-yn-2-one) has been developed to prepare C1-substituted carbazoles **109** in 32–94% yields (Scheme 59).¹⁴⁰ Additional substituents in the benzene ring of such products are aryl groups at the C3 position and, in a number of examples, at the C4 atom.

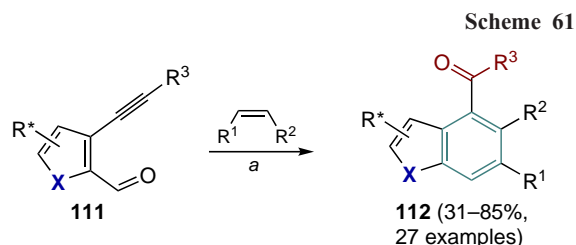


Scheme 59

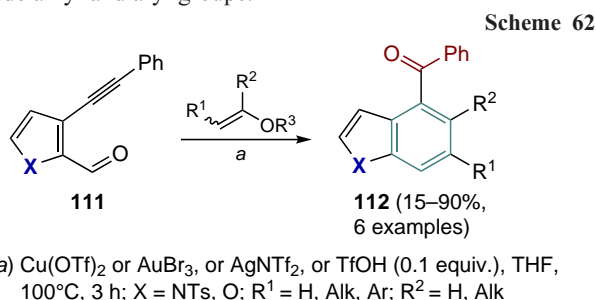
Carbazoles **110**, functionalized with a C3-aroil group, were obtained from 1-(indol-3-yl)but-3-en-1-ols **12** and aromatic aldehydes using $\text{Pd}(\text{bpy})\text{Cl}_2$ (bpy = 2,2'-bipyridine) as a catalyst (Scheme 60).¹⁴¹ The reaction products with yields ranging from 38% to 84% also contain aryl substituents at the C4 position of the benzene ring. It should be noted that in the presence of the $[\text{Pd}(\text{OAc})_2\text{bpy}]$ catalyst, the said substrates deliver C4-arylcarbazoles in the absence of aldehydes in 58–80% yields.



There are three synthetic approaches to C4-acyl-substituted BAHs. The synthesis of *N*-tosylindoles, benzo[*b*]thiophenes and benzo[*b*]furans **112** has been developed based on the reaction of 3-ethynylazole-2-carbaldehydes **111** with alkenes in the presence of bis(pyridine)iodonium(I) tetrafluoroborate (Scheme 61).^{142,143} The substituents at the C2 and C3 positions of the products can be alkyl or aryl groups, and the yields range from 31% to 85%.



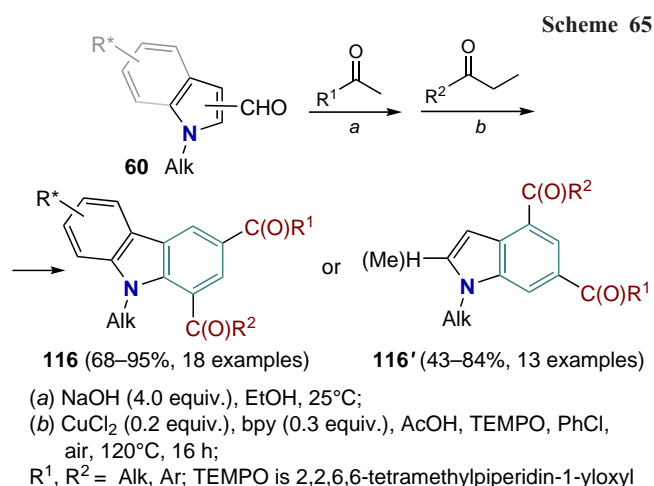
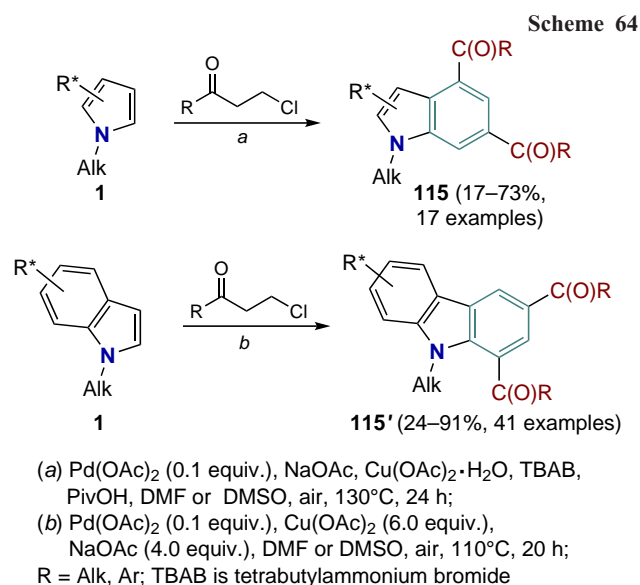
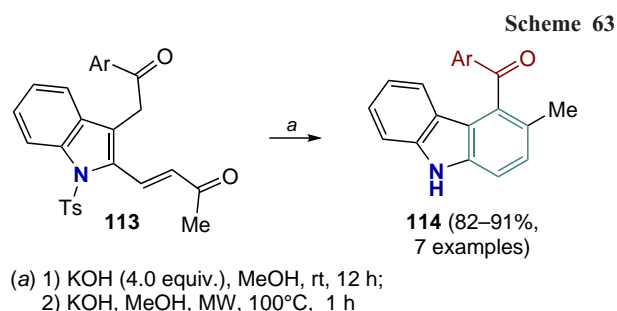
A similar method has been proposed for obtaining indole- and benzo[*b*]thiophene-based ketones from related substrates (Scheme 62).¹⁴⁴ The difference lies in the use of enol ethers as reagents and various Lewis or Brønsted acids (e.g., silver(I), gold(III) or copper(II) salts, or TsOH) as catalysts. The yields of the products depends on the nature of the catalyst and can reach 90%. Additional substituents at the C2 and/or C3 positions include alkyl and aryl groups.



5-Aroylcarbazoles **114** were obtained from 2-(2-acetylvinyl)-3-aroylethylindoles **113** by intramolecular aldol condensation, followed by the deprotection (Scheme 63).¹⁴⁵ The yields of these carbazoles are quite high, ranging from 82% to 91%.

Methods for obtaining diacylated BAHs have been developed. For example, the palladium(II) acetate-catalyzed reaction of 2,3-unsubstituted pyrroles or indoles **1** with β-chloroalkyl ketones gives products **115**, diacylated indoles¹⁴⁶ or carbazoles¹⁴⁷ (C2- and C4-substituted, 17–73%) and **115'** (C1- and C3-substituted, 24–91%) (Scheme 64).

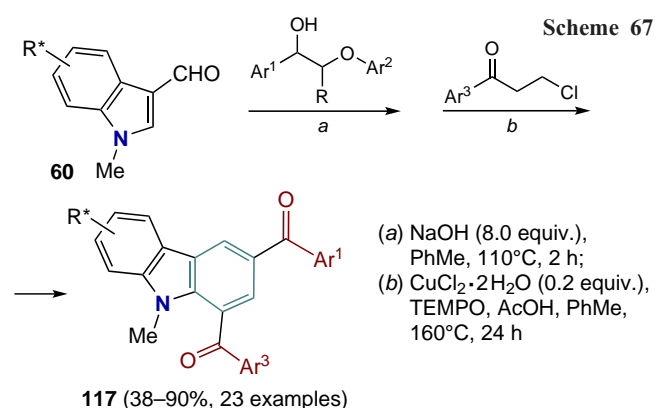
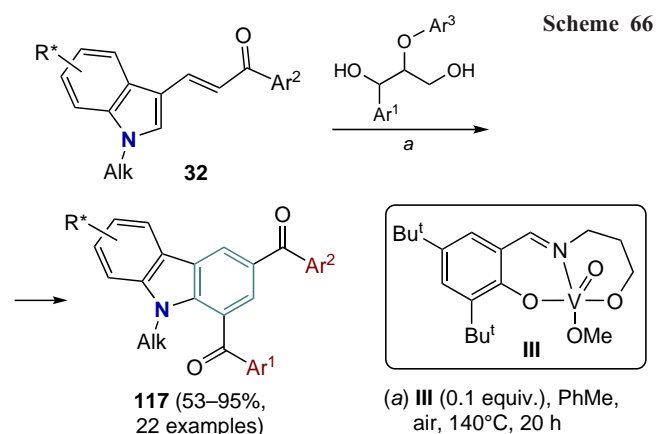
A similar approach is based on the copper(II) chloride-catalyzed annulation of indole-3-carbaldehydes or pyrrole-2-carbaldehydes by a sequential reaction with two different ketones (Scheme 65).¹⁴⁸ Using this method, C1,C3- or C2,C4-



diacylated carbazoles **116** (68–95%) and indoles **116'** (43–84%) were obtained, respectively.

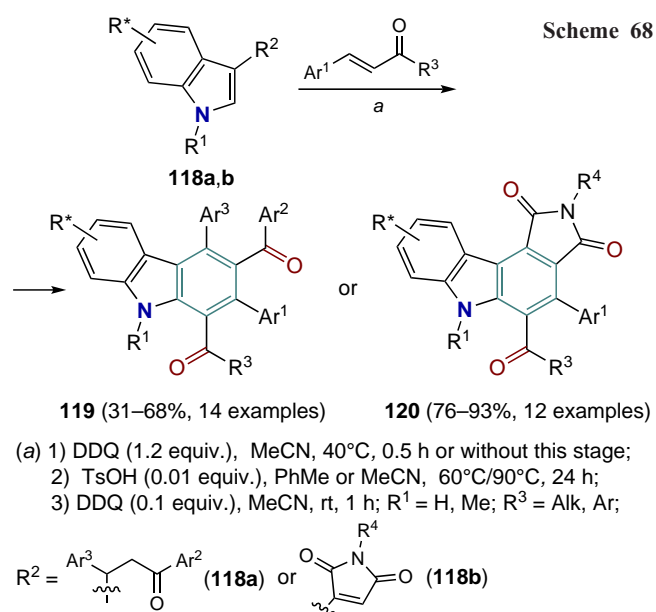
Carbazoles **117** bearing two different aroyl groups at the C1 and C3 positions were synthesized by benzoannulation of 3-alkenylindoles **32** with 2-aryloxypropane-1,3-diols in yields ranging from 53% to 95% (Scheme 66).¹⁴⁹ The reaction is promoted by a vanadium(V) complex **III** based on a tridentate Schiff base and occurs when refluxing the reagents in toluene in air for a long period of time.

Similar C1,C3-diaroyl-substituted carbazoles **117** are formed via a two-step procedure from indole-3-carbaldehydes **60** (Scheme 67).¹⁵⁰ In the first step, the reaction mixture is treated with ethylene glycol derivatives in the presence of an alkali.



Then, 3-chloropropiophenones and copper(II) chloride are added.

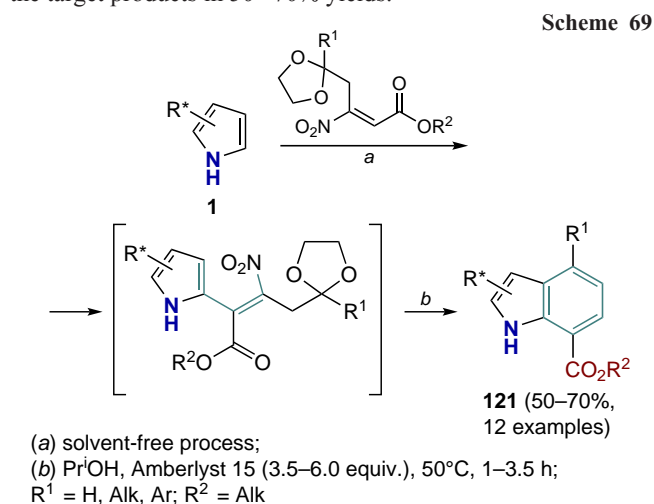
Another synthetic approach to carbazoles containing two or three acyl substituents on the benzene ring involves the acid-catalyzed Diels–Alder reaction of functionalized indoles **118a,b** with chalcones, followed by oxidation (Scheme 68).¹⁵¹ Using 3-(indol-3-yl)diarylpropanones **118a** as substrates gives C1,C3-diacetylcarbazoles **119** in 31–68% yields. When 3-(indol-3-yl)-maleimides **118b** are used as substrates, pyrrolo[3,4-*c*]carbazoles **120** are isolated in 76–93% yields.



Examples of the synthesis of BAHs containing the acyl-substituted benzene ring are also considered in Subsection 3.4, which covers the production of ketones and heterocyclic esters.

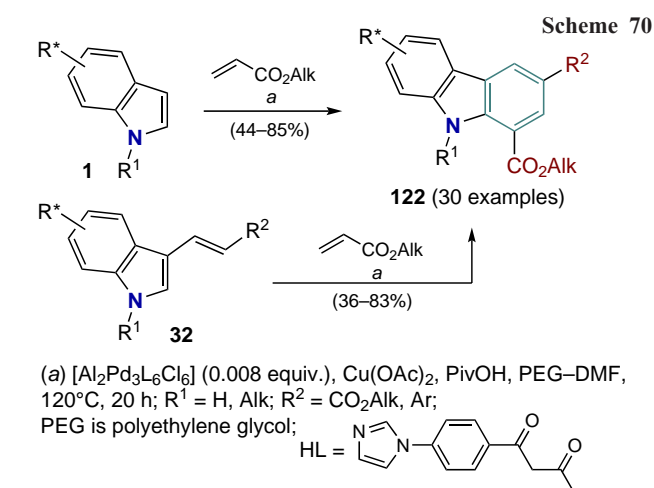
3.3. Ester-substituted heterocycles

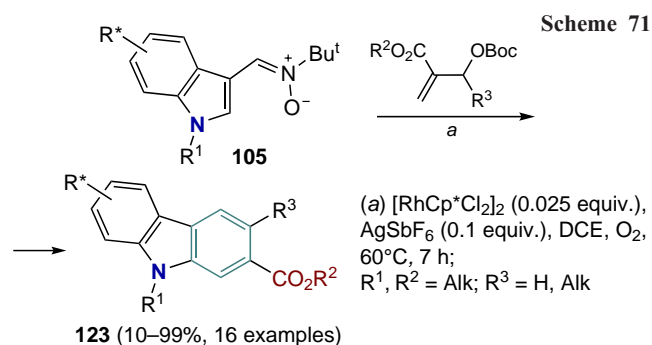
This section discusses methods for obtaining benzoannulated heterocycles containing one or two ester functionalities in the benzene ring, which is assembled during annulation. A one-pot synthesis of indoles **121** with a C1-alkoxycarbonyl group starting from 2,3-unsubstituted pyrroles and β-nitroacrylate derivatives has been developed (Scheme 69).¹⁵² The reaction runs in the absence of a solvent or catalyst to afford the corresponding Friedel–Crafts adducts. These adducts can then be treated with an acid catalyst (Amberlyst 15 resin) to produce the target products in 50–70% yields.



Carbazoles **122** with an ester moiety at the C1 and C3 positions were synthesized by the regioselective cascade oxidative Heck reaction using an Al(III) and Pd(II) supramolecular cage and the HL ligand (Scheme 70).¹⁵³ 2,3-Unsubstituted indoles **1** react with alkyl acrylates to give the target products in 44–85% yields. This reaction can also employ 3-alkenylindoles **32** as starting compounds, providing access to carbazoles containing both an ester functionality and an aryl group at the C3 position.

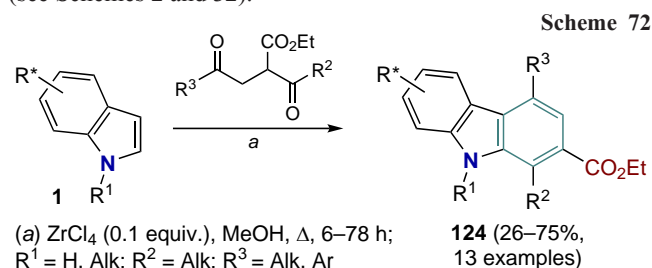
Methods have been developed for the synthesis of carbazoles with an ester group at the position C2. Rhodium(III)-catalyzed cross-coupling of indole nitrones **105** with hydroxy-substituted



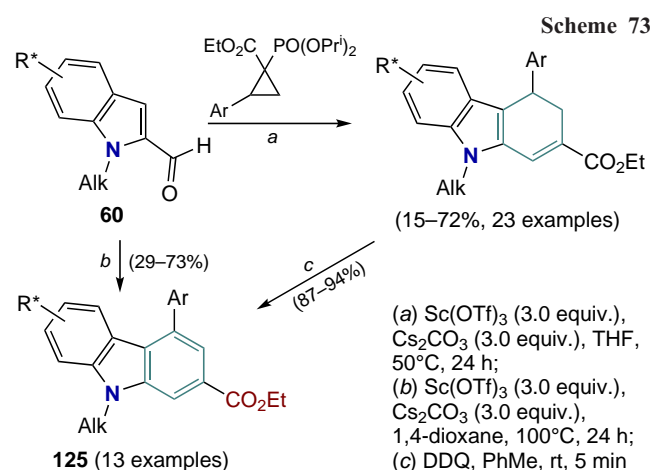


acrylate Michael acceptors affords the desired alkoxy-carbonylcarbazoles **123** in yields of up to 99% (Scheme 71).¹⁵⁴ A similar method was used to obtain the formyl-substituted BAHs (see Scheme 57).¹³⁸

C2-Ethoxycarbonylated carbazoles **124** were synthesized from 2,3-unsubstituted indoles and diketones containing an ester moiety in the presence of zirconium(IV) chloride in 26–75% yields (Scheme 72).⁷⁰ This method is a modification of that used to obtain BAHs with alkyl- and aryl-substituted benzene ring (see Schemes 2 and 32).

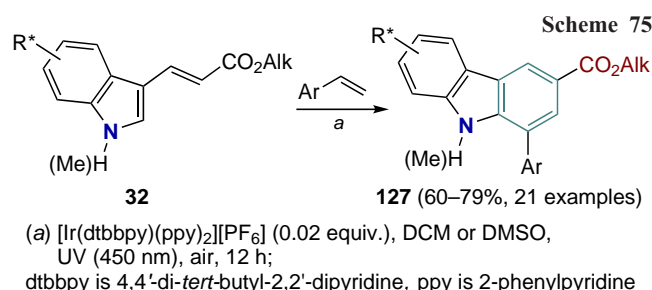
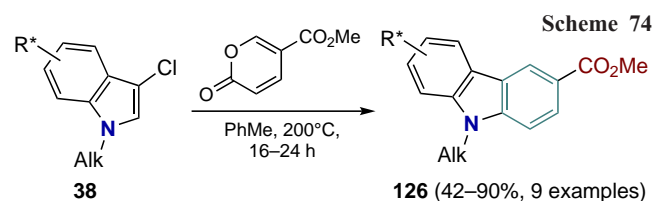


Carbazoles **125** were obtained from indole-2-carboxaldehydes and phosphonate-functionalized donor-acceptor cyclopropanes, using Lewis acids in dioxane (Scheme 73).¹⁵⁵ When carrying out the reaction in tetrahydrofuran, intermediate dihydrocarbazoles are formed which can be smoothly oxidized to the target products in rather high yields (from 87% to 94%).



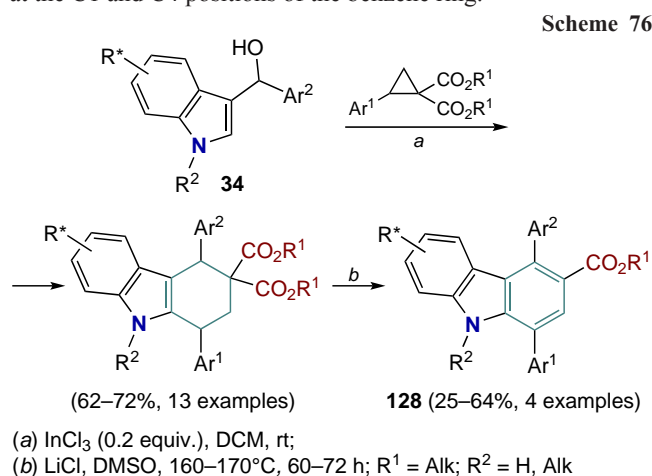
Several transformations are described yielding BAHs with a C3-ester moiety. Among them, mention should be made of the synthesis of carbazoles **126** via the Diels–Alder reaction between 3-chloroindoles **38** and 2-pyrone derivatives (42–90%) (Scheme 74).¹⁵⁶

Carbazole-3-carboxylates **127** were obtained by the reaction of 3-alkenylindoles **32** with styrene derivatives under blue light



irradiation and an iridium(III) complex as the photocatalyst (Scheme 75).¹⁵⁷ The tandem [2+2] cycloaddition–rearrangement affords the target products in 60–79% yields. The aryl group of styrene is an additional substituent at the C1 position of the benzene ring.

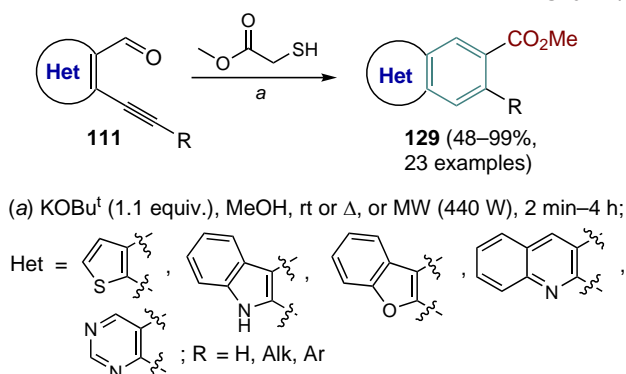
The Lewis acid-catalyzed annulation of indolyl alcohols **34** with donor-acceptor cyclopropanes produces intermediate dihydrocarbazoles (62–72%), which are then aromatized to give the target carbazole-3-carboxylates **128** (25–64%) (Scheme 76).¹⁵⁸ The resulting products also bear aryl substituents at the C1 and C4 positions of the benzene ring.



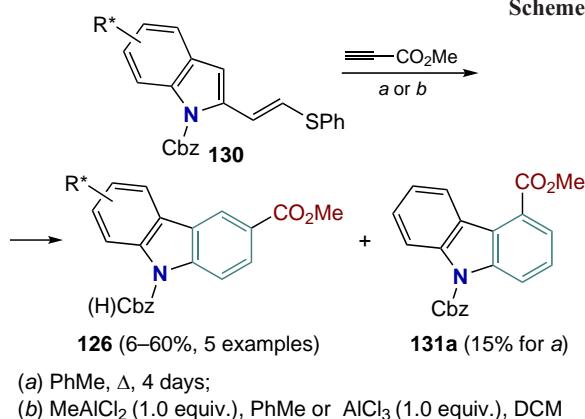
A wide range of C3-alkoxycarbonylated BAHs **129** (including carbazoles, quinazolines, acridines, benzo[*b*]thiophenes and dibenzofurans) were obtained by the microwave-assisted reaction of heterocycles **111**, which contain formyl and alkynyl groups in adjacent positions with methyl mercaptopropionate (Scheme 77).¹⁵⁹ Product yields ranged from 48% to 99%, and additional substituents at the C2 position could be alkyl or aryl groups.

Carbazoles **126** bearing a methoxycarbonyl group at the C3 position of the benzene ring were obtained by the reaction of *N*-benzyloxycarbonyl (Cbz)-protected (phenylthio)vinyl-substituted indoles **130** with methyl propionate in the presence of a Lewis acid (Scheme 78).¹⁶⁰ When the reaction was carried out without a catalyst using indole unsubstituted in the benzene ring ($\text{R} = \text{H}$), the formation of the by-product carbazole **131a**, which has a C4-ester group, was observed. The yields in this transformation reached 60%.

Схема 77

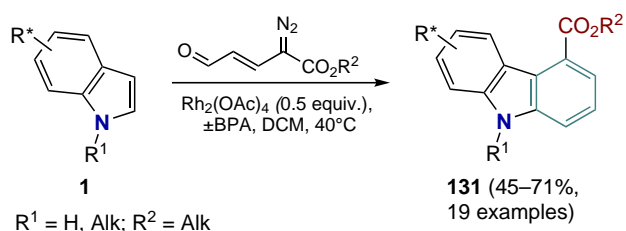


Scheme 78



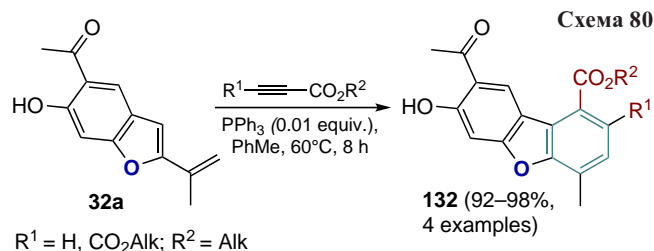
C4-alkoxycarbonyl-substituted carbazoles **131** were synthesized from 2,3-unsubstituted indoles **1** and diazoenals in the presence of rhodium(II) carboxylates and the Brønsted acid, the racemic 1,1'-binaphthyl-2,2'-dihydrophosphate (\pm BPA), in moderate yields (Scheme 79).¹⁶¹

Scheme 79



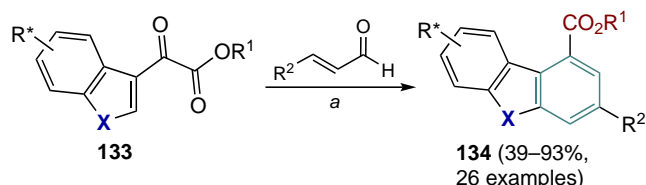
Benzofurans **132** with a C4-ester group are accessible through the Diels–Alder reaction of euparin (**32a**) with acetylenecarboxylic acid esters in high yields (Scheme 80).¹⁶² When acetylenedicarboxylic acid is used as a dienophile, this method enables the preparation of carbazoles containing two ester groups at the C3 and C4 positions. It should be noted that these products also have functional groups in the parent benzene ring.

Схема 80

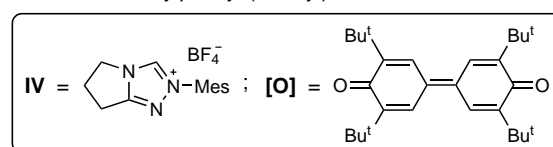


Catalyzed by an *N*-heterocyclic carbene (NHC) derived from precursor **IV**, the annulation of alkyl 2-heptyl-2-oxoacetates **133** with enals gives carbazoles and dibenzothiophenes **134** containing a C4-ester moiety (Scheme 81).¹⁶³ The yields of these products range from 39% to 93%, with alkyl or aryl groups being additional substituents at the C2 position of the newly-formed benzene ring.

Scheme 81

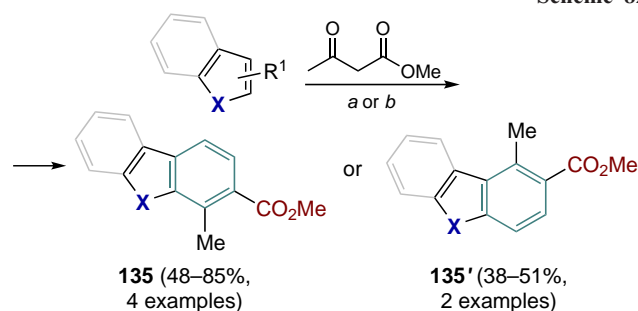


(a) **IV** (0.2 equiv.), **[O]**, DBU, THF, N₂, 30 °C, 24 h;
X = NTs, S; R¹ = Alk, alkenyl, Ar; R² = Alk;
Mes is 2,4,6-trimethylphenyl (mesityl)

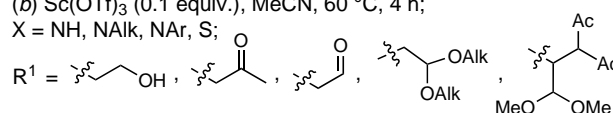


Isomeric methyl esters of indole, carbazole and benzothiophene (**135** and **135'**) were obtained from the appropriate heterocycles containing acetaldehyde or acetone functionalities or their precursors (Scheme 82).¹⁶⁴ Annulation of such substrates is promoted by methyl acetoacetate in the presence of a 2-iodobenzoic acid (IBX)—scandium(III) triflate system, which acts as a Lewis acid. The position (C3 or C2) of the ester moiety depends on the position of the substituent relative to the heteroatom in the substrate (α or β). Product yields are 48–85% for methyl 2-carboxylates **135** and 38–51% for methyl 3-carboxylates **135'**.

Scheme 82

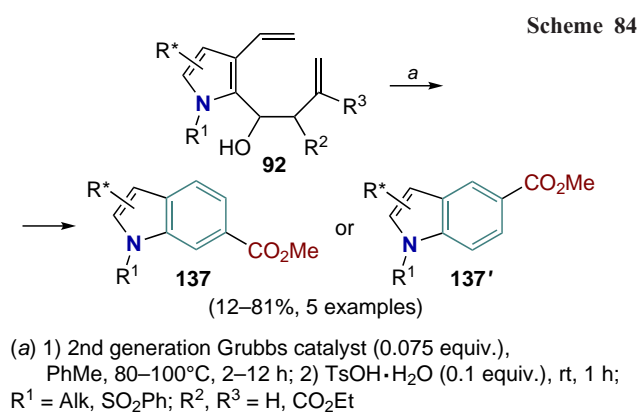
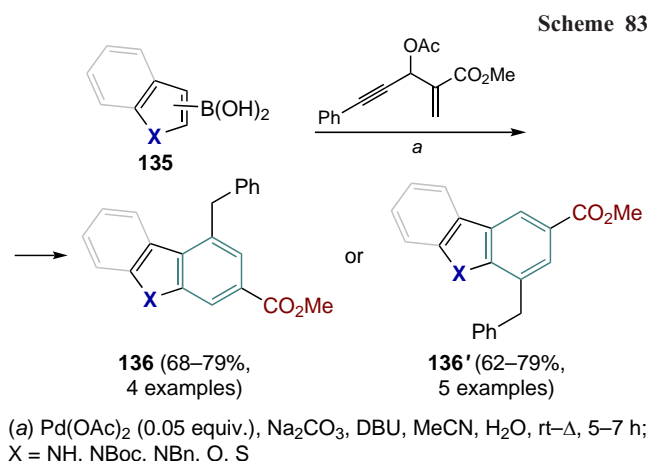


(a) IBX, then $\text{Sc}(\text{OTf})_3$ (0.1 equiv.), MeCN, 60°C, 2+4 h;
(b) $\text{Sc}(\text{OTf})_3$ (0.1 equiv.), MeCN, 60 °C, 4 h;
X = NH, NAlk, NAr, S;

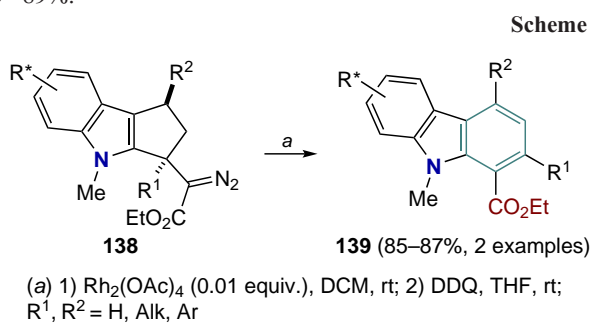


Another approach to C2- (**136**) or C3-methoxycarbonylated (**136'**) heterocycles involves the palladium(II) acetate-catalyzed annulation of arylboronic acids **135** with propargyl-substituted acrylates (Scheme 83).¹⁶⁵ The position of the ester functionality depends on the position of the boron-containing unit in the substrate. The yields of such benzoheterocycles are moderate (62–79%).

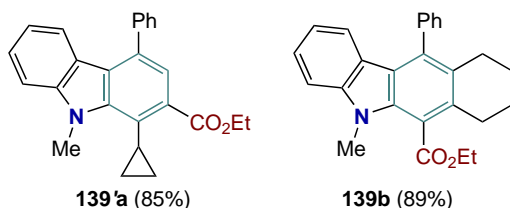
Indoles **137** and **137'** with an ester group at positions C2 or C3 were also obtained by metathesis of substituted pyrroles **92** in the presence of a Grubbs second-generation catalyst in 12–81% yields (Scheme 84).¹²⁷



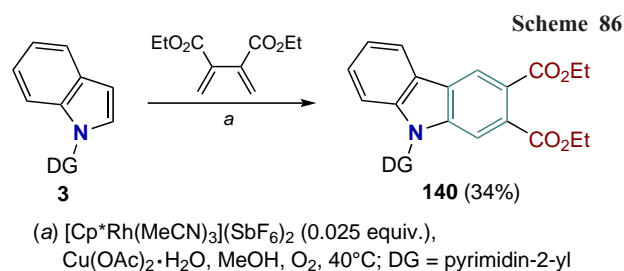
The possibility to obtain carbazoles **139** bearing an ethoxycarbonyl group at the C1 or C2 atom from tetrahydrocyclopenta[*b*]indole-substituted diazoesters was demonstrated (Scheme 85).¹⁶⁶ The position of the ester moiety depends on the nature of the substrate; in some cases, a mixture of isomers was formed. The product yields are in the range of 85–89%.



Other examples

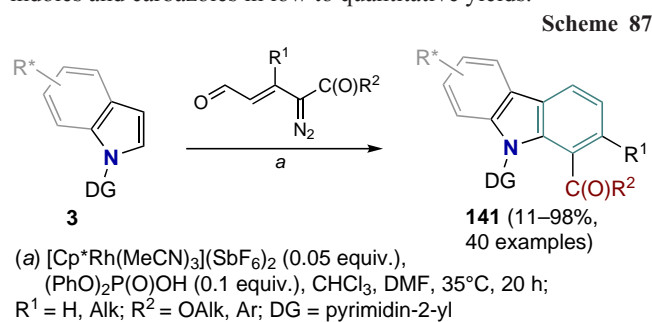


The synthesis of carbazole **140**, which has two ester groups at the C2 and C3 atoms, was accomplished through the annulation of 2,3-unsubstituted indole with 1,3-diene in the presence of a rhodium(III) complex (Scheme 86).¹²¹

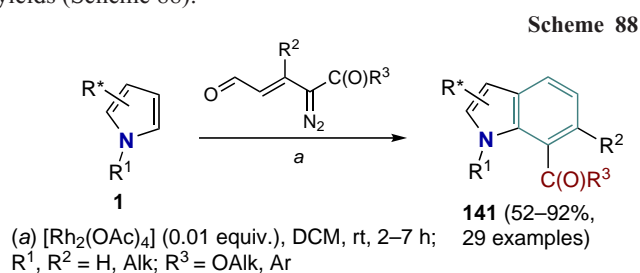


3.4. Ester- or acyl-substituted heterocycles

This Section describes the synthesis of benzoheterocycles containing an ester or acyl functionalities. Several synthetic approaches are known to BAHs with a single C(O)R group at the C1 position, where R = Ar, Alk or OAlk. One example is the rhodium(III)-catalyzed annulation of 2,3-unsubstituted heterocycles with diazoenals in the presence of a Brønsted acid, which affords indoles and carbazoles **141** (Scheme 87).¹⁶⁷ This method requires a pyrimidine directing group to be present at the nitrogen atom of the heterocycle. Additional alkyl groups can be present at the C2 position. The authors obtained forty different indoles and carbazoles in low to quantitative yields.

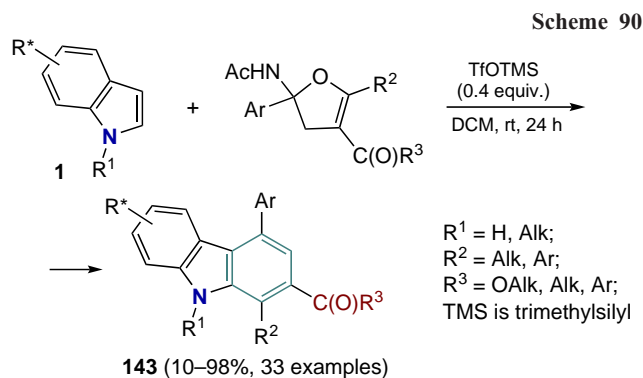
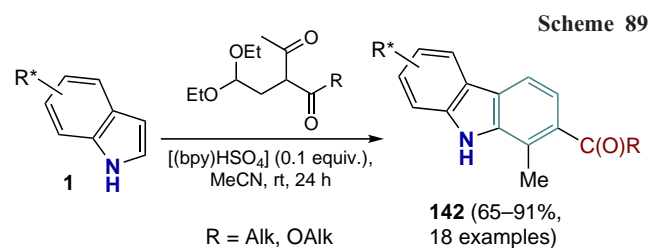


Using $[\text{Rh}_2(\text{OAc})_4]$ as a catalyst in this reaction gave rise to a series of C1-alkoxycarbonyl-substituted indoles **141** in 52–92% yields (Scheme 88).¹⁶⁸

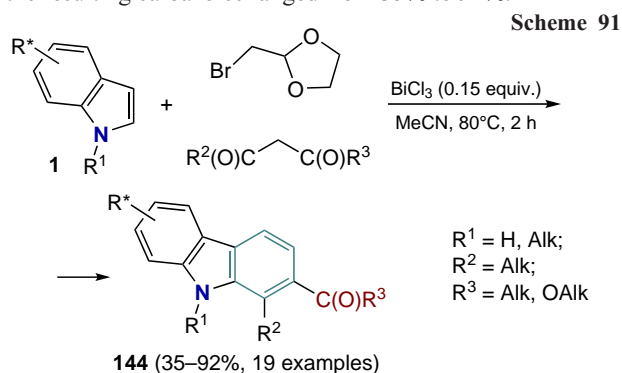


C2-carbonylated carbazoles can be synthesized from 2,3-unsubstituted indoles in several ways. For example, 1,3-dicarbonyl compounds containing a protected aldehyde group react with indoles in the presence of Brønsted acid ionic liquids (e.g., [bpy]SO₄) to give carbazoles **142** in yields ranging from 65 to 91% (Scheme 89).¹⁶⁹ The second substituent in the resulting benzene ring is a C1-methyl group.

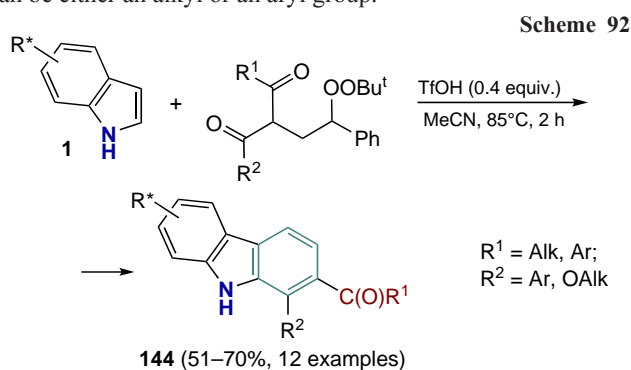
Trisubstituted carbazoles **143**, which contain a carbonyl group at the C2 position in addition to hydrocarbon substituents, were obtained from indoles and dihydrofuran derivatives (Scheme 90).¹⁷⁰ Trimethylsilyl trifluoromethanesulfonate was used as a catalyst for this reaction, and the yields of the product ranged from low to quantitative, depending on the nature of the substituents in the starting dihydrofuran.



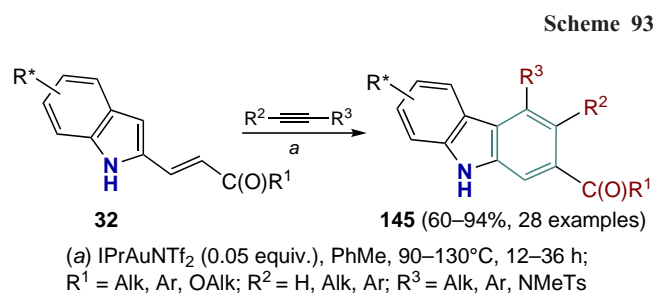
C2-carbonyl-substituted carbazoles **144** were obtained *via* a three-component reaction involving 2,3-unsubstituted indoles, protected bromoacetaldehyde and 1,3-dicarbonyl compounds in the presence of bismuth(III) chloride (Scheme 91).¹⁷¹ The yields of the resulting carbazoles ranged from 35% to 92%.



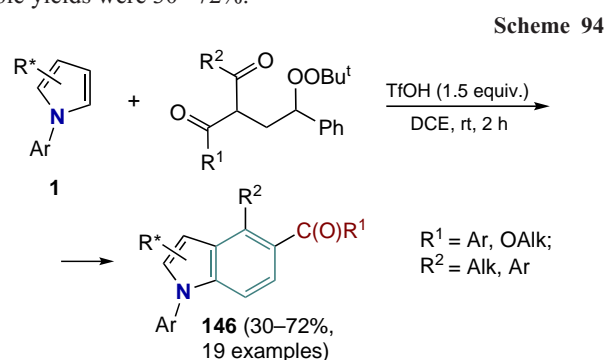
Annulation of 2,3-unsubstituted pyrroles and indoles with γ -carbonyl-*tert*-butyl peroxides in the presence of TfOH affords the corresponding NH-carbazoles (Scheme 92).¹⁷² The yields of products **144** are 51–70%, and the substituent at the C1 position can be either an alkyl or an aryl group.



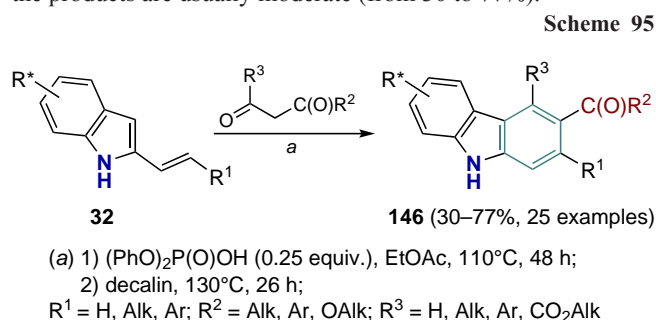
Finally, a method has been developed for synthesizing carbazoles **145** containing different C(O)R groups at the C2 atom of the benzene ring. Gold(I)-catalyzed annulation of 2-alkenylindoles **32** with mono- and disubstituted alkynes provides products in 60–94% yields (Scheme 93).¹⁷³



Methods for obtaining carbazoles and indoles with an acyl or ester group at the C3 position have been proposed. The above method, which is intended for C2-substituted carbazoles (see Scheme 92), was also used to synthesize C3-carbonylindoles **146** from 2,3-unsubstituted *N*-arylpyrroles (Scheme 94).¹⁷⁴ The indole yields were 30–72%.



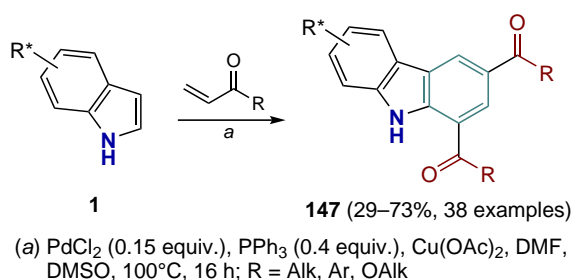
C3-carbonyl-substituted carbazoles **146** are accessible *via* a one-pot annulation protocol involving 2-alkenylindoles **32** and ethyl acetoacetate derivatives in the presence of a Brønsted acid catalyst (Scheme 95).¹⁷⁵ Varying the structure of the carbonyl-substituted agent opens access to carbazoles with electron-withdrawing moieties at the C3 and C4 positions. The yields of the products are usually moderate (from 30 to 77%).



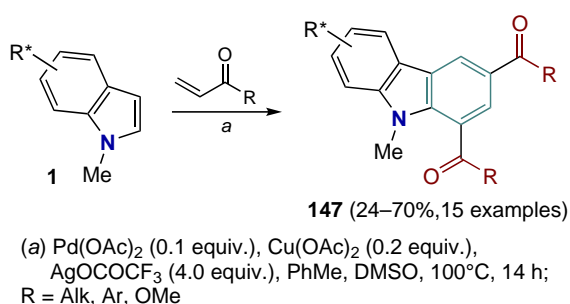
Methods are known for producing BAHs with several acyl and/or ester functionalities on the benzene ring. One such method involves obtaining carbazoles with two carbonyl groups at the C1 and C3 positions of the newly-formed benzene ring, based on 2,3-unsubstituted indoles and unsaturated ketones (Scheme 96).¹⁷⁶ The intermediate compounds in this reaction are styrylindoles, which can also serve as starting materials.¹⁷⁷ The scalability of the method was demonstrated in gram-scale experiments.

C1,C3-dicarbonyl-substituted *N*-methylcarbazoles **147** were also obtained in 24–70% yields from 2,3-unsubstituted *N*-methylindoles and alkenes, similar to those used in the previous reaction, in the presence of a trimetallic system of palladium(II) acetate–copper(II) acetate–silver trifluoroacetate (Scheme 97).¹⁷⁸

Scheme 96



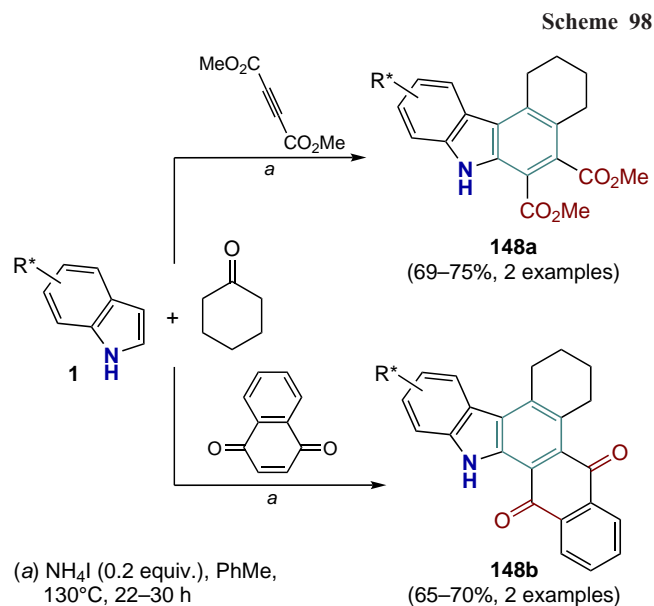
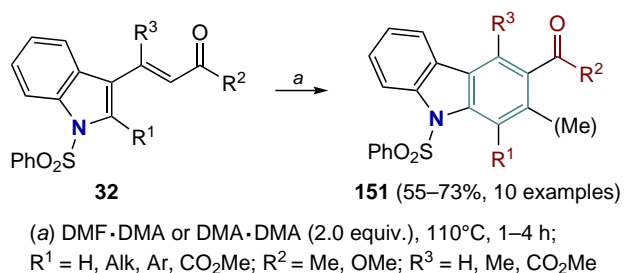
Scheme 97



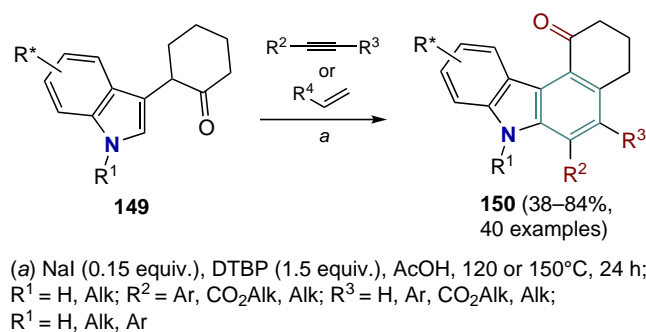
The reaction of 2,3-unsubstituted indoles with cyclohexanone and various dienophiles produces C1,C2-dicarbonyl-substituted carbazoles **148a,b** in 69–75% yields (Scheme 98).¹⁷⁹ The dienophiles used were 1,4-naphthoquinone and dimethyl acetylenedicarboxylate.

A method has been developed for the synthesis of carbonyl-substituted fused carbazoles **150** based on derivatives of (indol-3-yl)cyclohexanone **149** and alkynes or alkenes, using sodium iodide and di-*tert*-butyl peroxide (DTBP) as an oxidant for iodine generation (Scheme 99).¹⁸⁰ When alkynes were used, mainly C4-carbonylcarbazoles (38–84%) and only one C1,C4-dicarbonylcarbazole (65%) were obtained. Conversely, the use of alkenes gave C1,C4-dicarbonylcarbazoles in yields ranging from 44% to 61%.

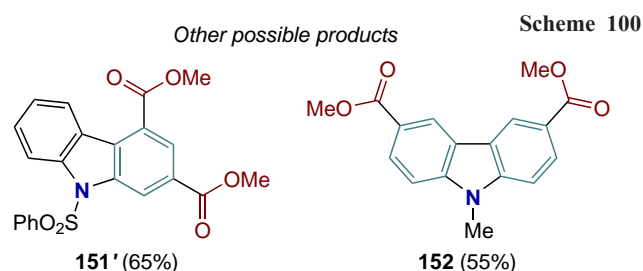
Carbazoles **151**, bearing one or two carbonyl groups, are synthesized from vinylindoles **32**, which form enamines when treated with *N,N*-dimethylformamide dimethyl acetal



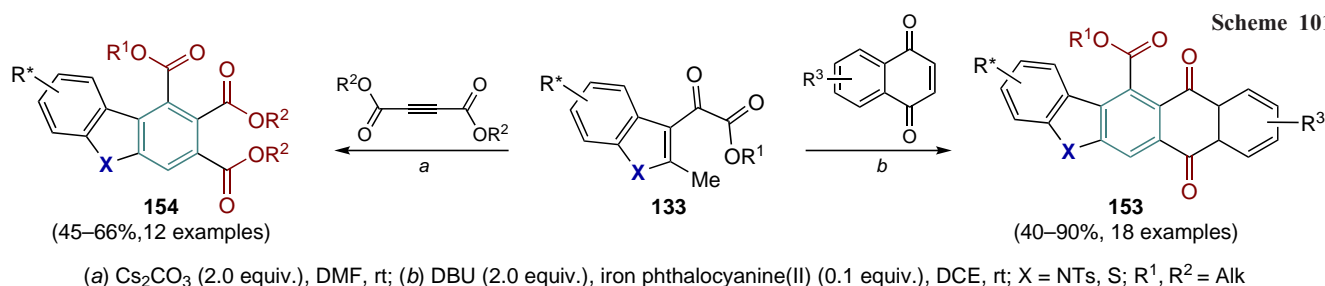
Scheme 99



(DMF·DMA) (Scheme 100).¹⁸¹ Further thermal electrocyclization and aromatization furnish the target products in 55–73% yields. It should be noted that using *N,N*-dimethylacetamide dimethyl acetal gives carbazoles with an additional C2-methyl group. The position of the carbonyl group depends on the substrate's structure, particularly the position of



Scheme 101

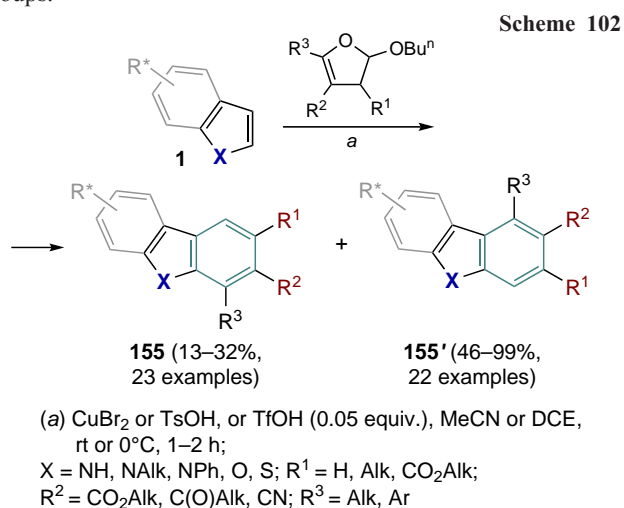


the acrylic moiety. For example, in the case of indole-3-yl acrylate, product **151'** is formed in a moderate yield. The dibenzoannulation of pyrrole to carbazole **152** with two methoxycarbonyl groups is also demonstrated (55%).

Another example of the application of dienophiles in the synthesis of C2,C3,C4-tricarbonyl-substituted carbazoles and dibenzothiophenes is shown in Scheme 101.¹⁸² 2-Hexyl-2-oxoacetates **133** react with 1,4-naphthoquinone in the presence of DBU to give heterocycles **153** annulated with naphthalene-1,4-dione (40–90%). Adding an iron(II) phthalocyanine complex to the reaction mixture significantly increases the product yield. The reaction with dialkyl acetylenedicarboxylates gives benzoheterocycles **154** with three carboxy groups at the C2, C3 and C4 positions of the benzene ring (45–66%).

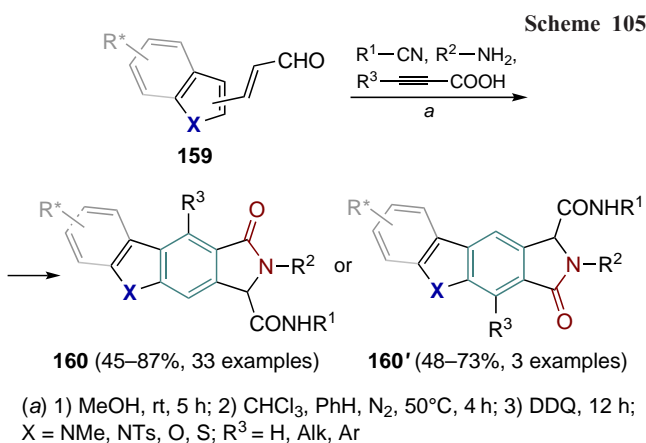
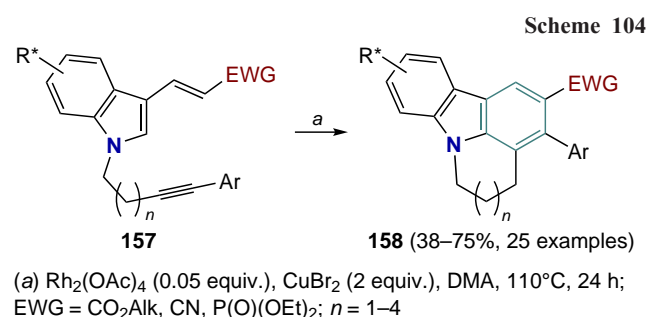
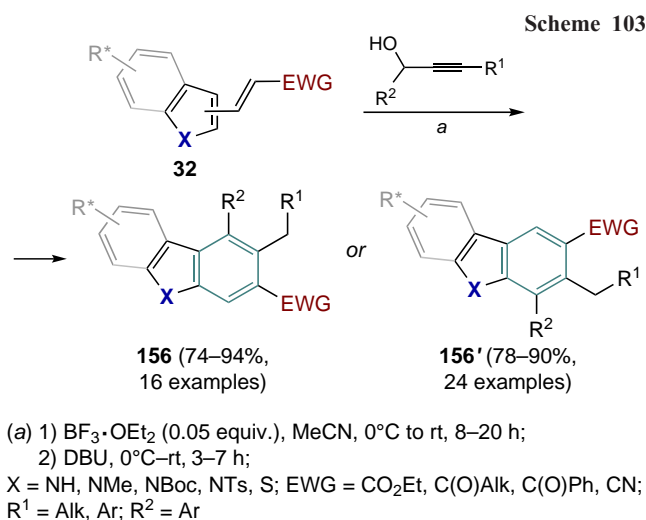
3.5. Heterocycles with different electron-withdrawing substituents

This Section discusses synthetic approaches to benzoheterocycles, where the resulting benzene ring is substituted with various types of electron-withdrawing groups (nitrile, amide, sulfur- or phosphorus-containing). Indoles, carbazoles, benzo[*b*]furans and benzo[*b*]thiophenes with different combinations of substituents were obtained by the benzoannulation of 2,3-unsubstituted heterocycles with 2-butoxy-2,3-dihydrofuran (Scheme 102).¹⁸³ Depending on the type of heterocycle, the EWG substituent ($R^2 = \text{CO}_2\text{Alk}$, C(O)Alk or CN) can be located at position either C2 or C3 of the newly-formed benzene ring. Product yields range from 13% to 99%. In addition, the benzene ring can contain alkyl and aryl groups.



A method has been developed to obtain compounds **156** and **156'**, including carbazoles,¹⁸⁴ indoles and benzo[*b*]thiophene,¹⁸⁵ with a carbonyl or nitrile group. This method is based on alkenyl-substituted benzene heterocycles **32** (Scheme 103). The reaction involves the acid-catalyzed annulation of heterocycles with propargyl alcohols. The position of the C2 or C3 functional group depends on the position of the alkenyl substituent in the substrate molecule.

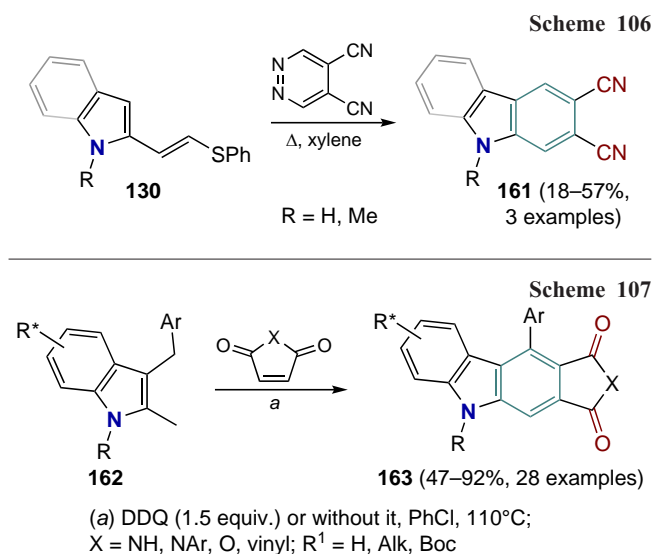
The rhodium(III)-catalyzed benzoannulation of indolyl acrylate derivatives **157** with an alkynyl-tethered nitrogen atom gives carbazoles **158** containing an alkylcarbonyl, nitrile or phosphonate group at the C3 position of the benzene ring (Scheme 104).¹⁸⁶ The yields of carbazoles vary from 38% to 75%.



Indoles, carbazoles and benzo[*b*]thiophene and benzo[*b*]furane (**160** and **160'**) with an amide group at the C2 or C3 position can be obtained *via* the sequential Ugi and Diels–Alder reactions (Scheme 105).¹⁸⁷ Heterocyclic acrolein derivatives **159** react with arylamines, carboxylic acids and isonitriles to give bisamides, which are then cyclized and oxidized to produce the target benzoannulated heterocycles in 45–87% yields.

2,3-Dicyano-substituted indoles and carbazoles **161** were obtained by thermal cycloaddition of dicyanopyridazine to 2-thioalkenyl-tethered pyrroles and indoles **130** (Scheme 106).¹⁸⁸ Three products **161** were isolated in 18–57% yields.

The Diels–Alder oxidative reaction between alkylindoles **162** and dienophiles afforded carbazoles **163** that were trapped by a dicarbonyl-containing cycle (Scheme 107).¹⁸⁹ For heterocyclic dienophiles (*e.g.*, maleimide and furanedione), the method involves using DDQ as an oxidant. In the case of

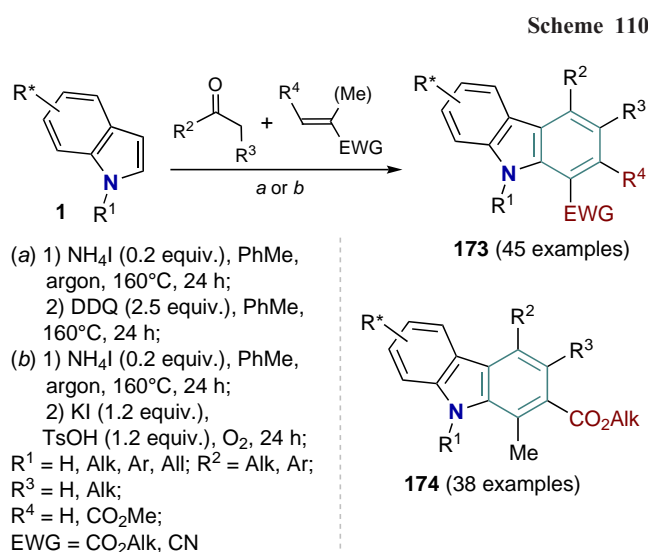
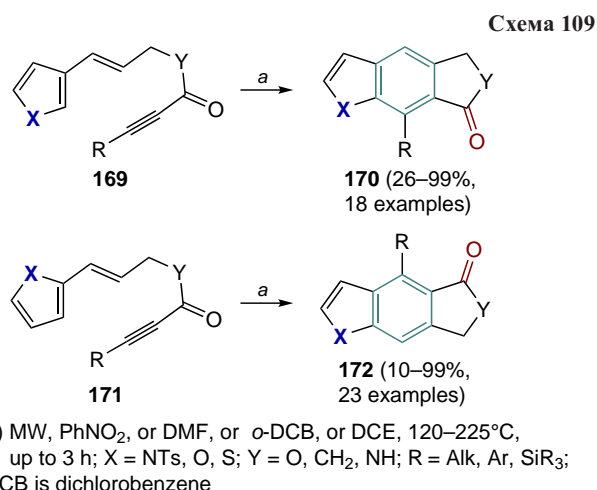


benzoquinone, however, the reaction proceeds without an additional oxidant. Product yields range from 47% to 92%.

The palladium(II) acetate-catalyzed Heck reaction of 2,3-dibromoheterocycles **164** with acrylates gives three adducts: isomeric dihydrocarbazoles **165**, **166** or 2,3-(dialkenyl)-heterocycles **167** in 67–95% yields (Scheme 108). Subsequent dehydrogenation of compounds **165**, **166** on Pd/C, or their thermal 6 π -electrocyclization furnishes products **168** — indoles,¹⁹⁰ carbazoles,¹⁹¹ benzofurans¹⁹² and dibenzofurans^{193,194} bearing two ester, nitrile or aryl moieties in 69–99% yields.

A synthetic approach to keto-substituted indoles, benzo[*b*]-thiophenes and benzo[*b*]furans **170** and **172**, based on heterocyclic derivatives **169** and **171** containing a complex enyne unit at the positions 2 or 3, is described (Scheme 109).¹⁹⁵ Depending on the position of the substituent in the heterocycle, C3- or C2-carbonyl-substituted products were isolated. Additional substituents at the positions C1 or C4 of the heterocycle benzene ring are phenyl, trimethylsilyl, and triisopropylsilyl groups. Product yields depend on the substrate structure and range from 10% to 99%.

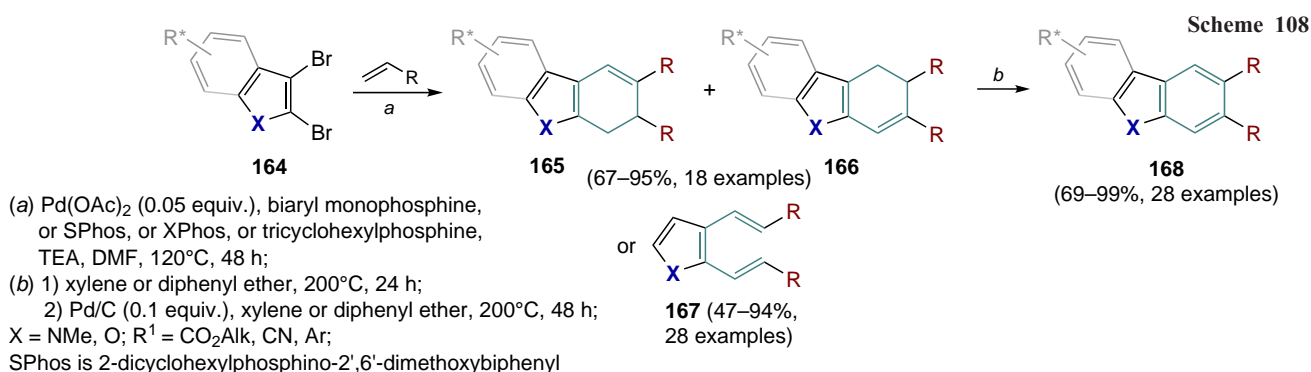
The following Schemes illustrate methods for the synthesis of benzoheterocycles bearing one to three electron-withdrawing groups, depending on the substrate's structure. For example, carbazoles **173** and **174**, which contain an EWG at the C1 and C2 positions, are formed *via* the ammonium iodide-promoted benzoannulation of 2,3-unsubstituted indoles with ketones and alkenes in 20–83% yields (Scheme 110).^{109,196,197} The structure of the alkene governs the position of the carbonyl group in carbazoles (C1 or C2) and also the possibility of obtaining

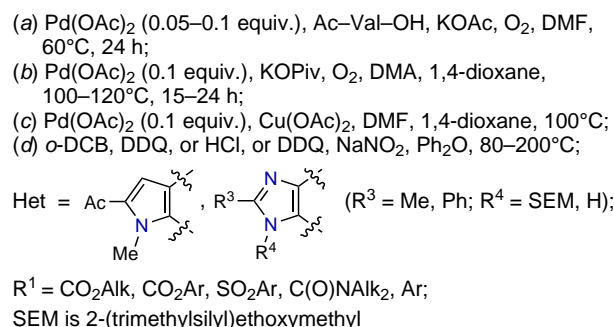
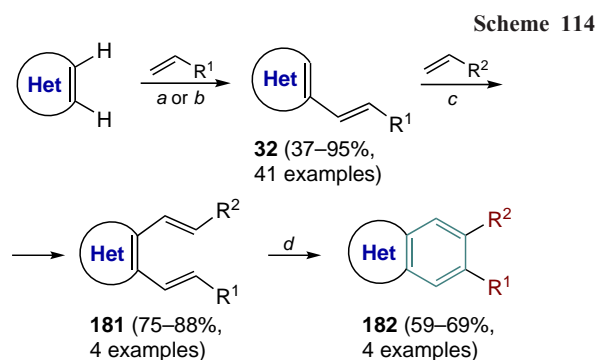


C1,C2-dicarbonylated products. It should be noted that both alkyl- and aryl-substituted carbazoles are accessible by this method.¹²³

Carbazoles and α -carbolines **175**, which are functionalized by EWGs at the positions C2 and C3 of the benzene ring, are obtained from 2- or 3-methoxycarbonylvinyl derivatives **32** and the corresponding alkenes in 60–92% yields (Scheme 111).¹⁹⁸ C1,C3-disubstituted carbazoles were also isolated as minor products in this reaction

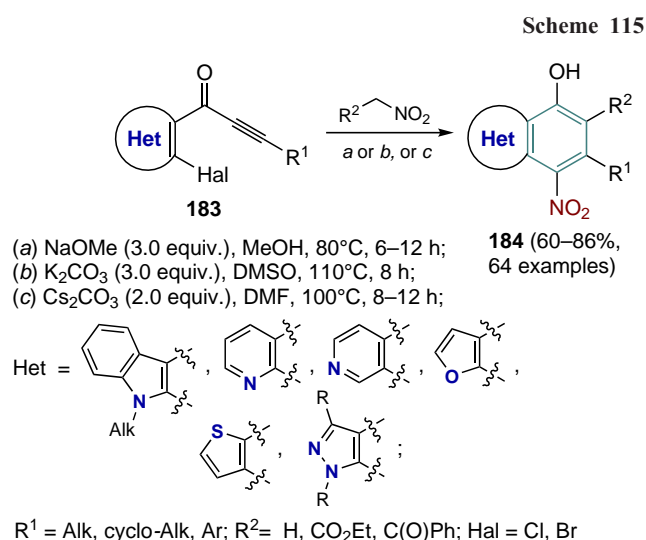
Quinolines, acridines and phenanthridines **177** and **178** were obtained from functionalized heterocycles **176** containing both an O-substituted allyl group and a chlorine atom



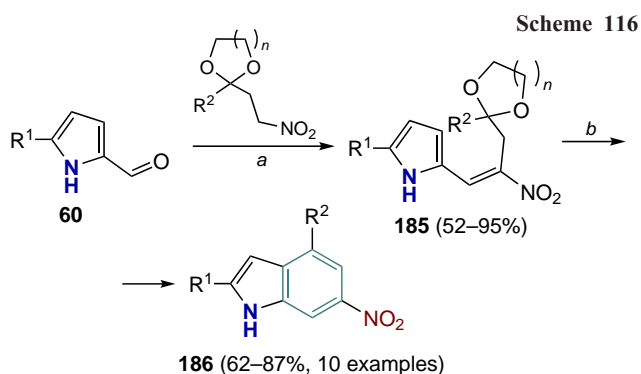


3.6. Nitro-substituted heterocycles

Benzo[*a*]carbazole derivatives **180** were synthesized from 3-substituted indoles **179** and benzoquinone, with the latter acting both as an oxidant and a dienophile (Scheme 113).²⁰¹ It should be noted that, in addition to carbonyl groups, nitro and alkyl or aryl groups can be introduced into the C3 and C4 positions, respectively. The product yields are 70–98%.



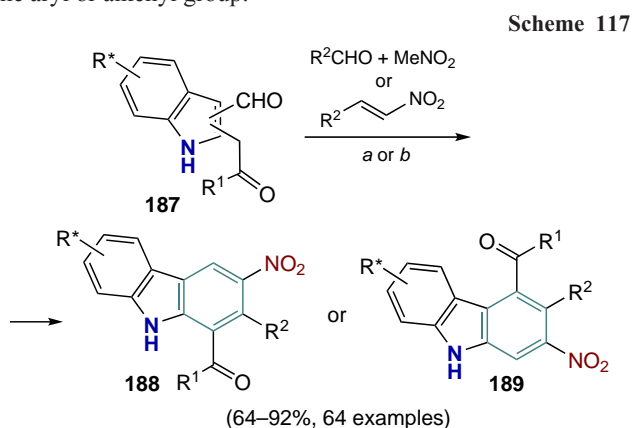
The synthesis of C2-nitroindoles **186** was carried out in two steps starting from pyrrole-2-carboxaldehydes (Scheme 116).²⁰⁷ At the initial step, the substrates were annulated with protected



(a) DBU (1.0 equiv.), 1,4-dioxane, reaction in flow mode, rt;
(b) Amberlyst 15 (0.05 equiv.), 2-MeTHF, MW, 100°C, 45 min;
R¹ = H, Alk; R² = H, Alk, Ar; n = 1, 2

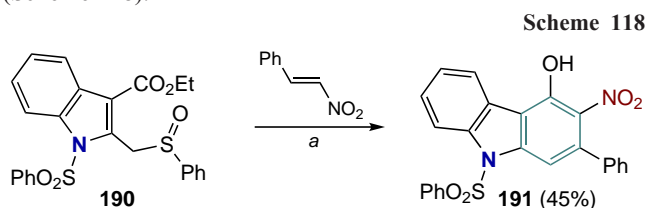
β-nitroketones to form intermediates **185**. This was followed by microwave-assisted cyclization in the presence of Amberlyst 15 ion-exchange resin to give nitroindoles **186** in yields ranging from 62% to 87%.

Carbazoles functionalized with a nitro group at the C3 position of the benzene ring can be produced in several ways. For example, a large series of C2- and C3-substituted nitrocarbazoles **188** and **189** were prepared by condensation of NH-indole-3(2)-carbaldehyde derivatives **187** with nitroalkenes (including those generated *in situ*) in 64–92% yields (Scheme 117).^{77,208} These carbazoles contain a carbonyl group in the C1 or C4 position of the resulting benzene ring, as well as one aryl or alkenyl group.

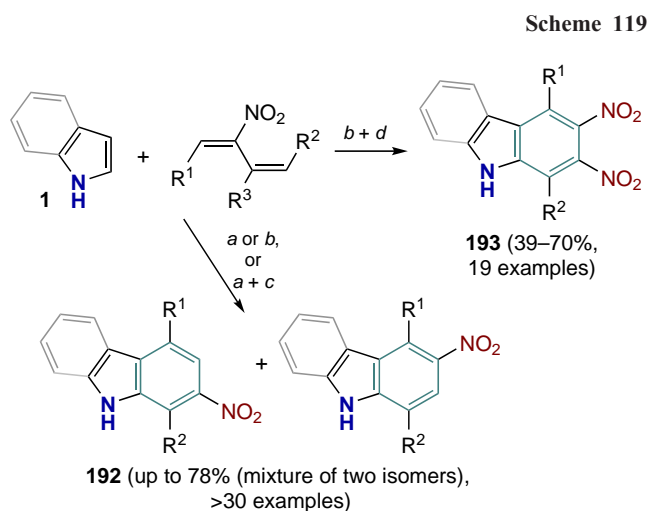


(a) DABCO (0.3 equiv.), H₂O, 70°C, 10–20 h;
(b) Al₂O₃, 65°C, 3–6 h;
R¹ = Ar, Het, OAlk; R² = Ar, Het, alkenyl;
DABCO is 1,4-diazabicyclo[2.2.2]octane

The synthesis of C3-nitro-substituted carbazole **191** by the reaction of indolylmethylsulfoxide **190** and 2-nitrostyrene in the presence of a strong base in 45% yield is described (Scheme 118).²⁰⁹



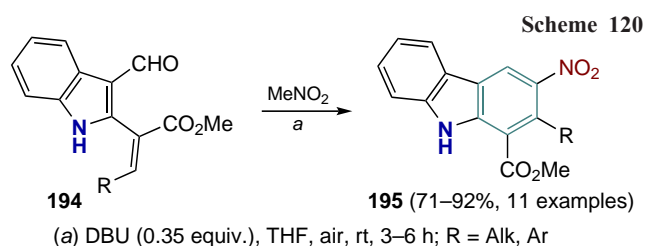
(a) Bu^tOK (1.0 equiv.), THF, MeCN, 0°C, 2–5 h



(a) Zn(OAc)₂ (0.05 equiv.), EtOH, 50°C; (b) TFE, 50°C;
(c) DBU (1.1 equiv.), DCM, rt; (d) DDQ (2.0 equiv.), PhMe, rt;
R¹ = Alk, Ar; R² = Alk, Ar, SO₂Me; R³ = NO₂, SO₂Ph

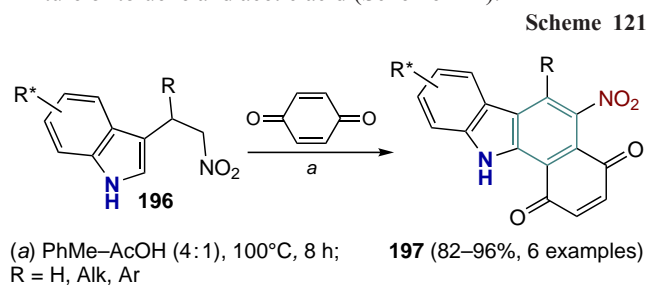
Indoles²¹⁰ and carbazoles containing a nitro group at the positions C2 and/or C3 are accessible by the reaction of 2,3-unsubstituted heterocycles with mono- or dinitrobutadienes (Scheme 119).^{211–213} In general, this reaction proceeds non-selectively to afford three functionalized BAHs **192** and **193**, but there are some examples where the products can be isolated in good yields. For example, dinitro-substituted indoles were obtained in 39–70% yields when the process was carried out in trifluoroethanol, followed by oxidation with DDQ.²¹⁰ C3-nitro-substituted carbazoles were obtained in yields of up to 58% when the reaction was carried out in the presence of zinc(II) acetate in ethanol.²¹²

Indole-3-carbaldehyde derivatives **194** react with nitromethane in the presence of DBU to give C3-substituted nitrocarbazoles **195** in 71–92% yields (Scheme 120).²¹⁴



(a) DBU (0.35 equiv.), THF, air, rt, 3–6 h; R = Alk, Ar

Six benzo[*a*]carbazoles **197** bearing a C3-nitro group were obtained in high yields (82–96%) by reacting 3-(2-nitroethyl)-indole derivatives **196** with benzoquinone on heating in a mixture of toluene and acetic acid (Scheme 121).²⁰¹

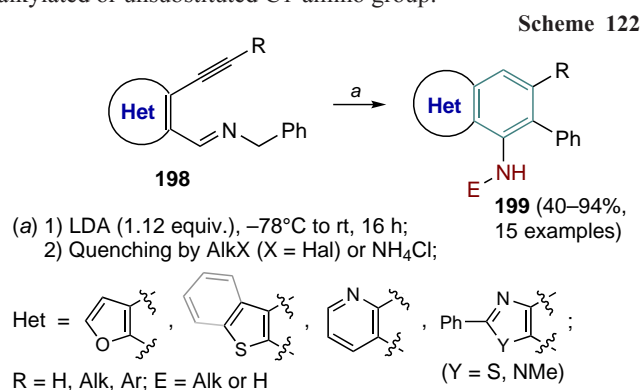


(a) PhMe–AcOH (4:1), 100°C, 8 h;
R = H, Alk, Ar

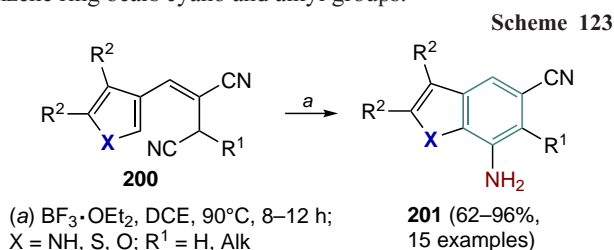
4. Benzoheterocycles with electron-donating substituents

4.1. Amino-substituted heterocycles

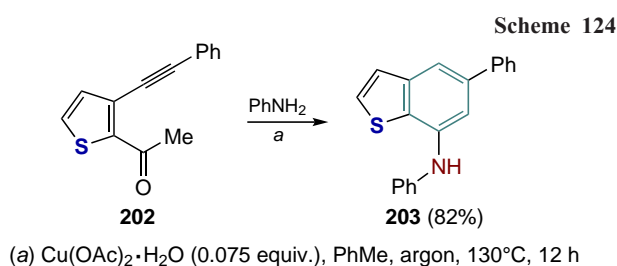
Methods have been developed for the synthesis of BAHs that are functionalized with an amino group at the C1 position of the benzene ring newly-formed in the result of annulation with the heterocycle. Depending on the method used, products containing both free and substituted NH_2 groups can be obtained. One of the most effective methods in terms of the number of the resulting benzoheterocycles is the cyclization of *N*-benzylhetarylmethanimines (Scheme 122).^{215,216} Deprotonation of the latter with lithium diisopropylamide (LDA) and further treatment with alkyl halides or ammonium chloride affords BAHs **199** in yields ranging from 40% to 94%. This method delivers derivatives of benzo[*b*]thiophene, benzo[*b*]furan, benzo[*d*]imidazole, benzothiazole, quinoline and dibenzothiophene containing an alkylated or unsubstituted C1-amino group.



The synthesis of indoles, benzo[*b*]thiophenes and benzo[*b*]furans **201** with an unsubstituted amino group at the C1 position is reported (Scheme 123).²¹⁷ Acid-catalyzed cyclization of nitriles **200**, derived from the corresponding aldehydes according to the Hueben–Hoesch method, gives the annulated heterocycles in 62–96% yields. In addition to the amino group, the resulting benzene ring bears cyano and alkyl groups.

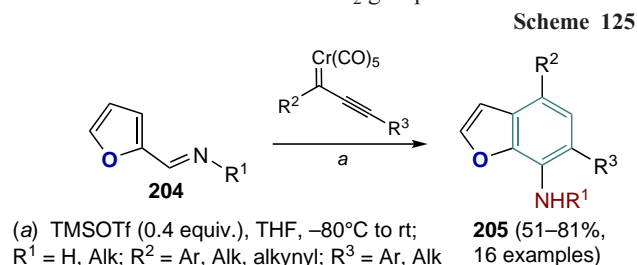


The next method was primarily tested for producing functionalized naphthalenes, but its principal applicability to benzo[*b*]thiophene was also demonstrated (Scheme 124). Copper(II) acetate-catalyzed aminobenzannulation of the

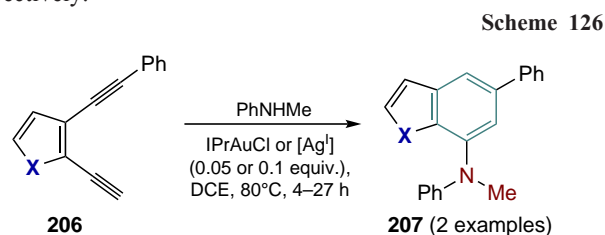


thiophene-based (*o*-alkynyl) aryl ketone **202** affords the product **203** with an arylated C1-amino group in a good yield.²¹⁸

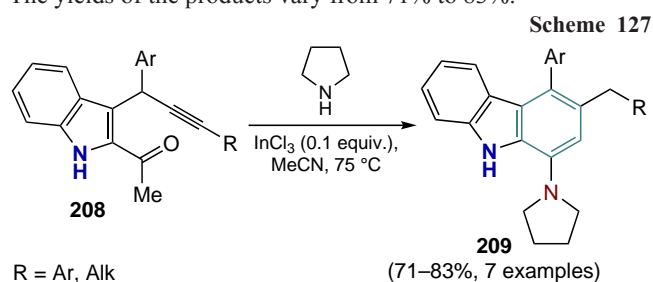
Benzo[*b*]furans **205** were obtained *via* the benzannulation of 2-furfural imines **204** with chromium alkynylcarbene complexes (Fischer carbene complexes) (Scheme 125).^{219,220} This approach enables the regioselective synthesis of the target products in 51–81% yields. Various alkyl substituents can be present as R^1 substituents at the NH_2 group.



The synthesis of BAHs with a tertiary amino group at the position C1 was carried out using gold(I)-catalyzed cyclization of diynes **206** (Scheme 126).²²¹ Benzo[*b*]thiophene and benzo[*b*]furan **207** are produced in 75 and 35% yields, respectively.



The following method based on the reaction of 2-acetyl-3-propargylindoles **208** with pyrrolidine using indium trichloride as a catalyst, gives rise to C1-aminocarbazoles **209** (Scheme 127).²²² In the reaction, pyrrolidine forms an enamine, which undergoes the subsequent cyclization, and the amino group enters the carbazole molecule as a pyrrolidine residue. The yields of the products vary from 71% to 83%.

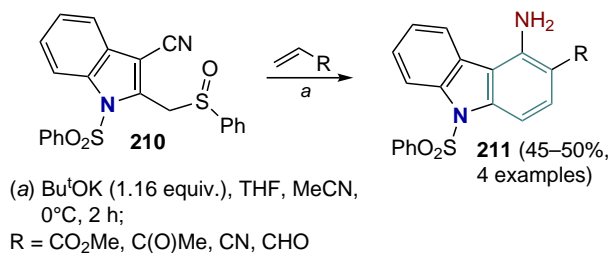


Synthetic approaches to BAHs with a C4-amino moiety have been developed. BAHs with an unsubstituted amino group can be obtained in two ways. In the first case, annulation of indole-2-methylsulfoxide **210** with Michael acceptors affords C4-aminocarbazoles **211** in 45–50% yields (Scheme 128).²⁰⁹ In addition to the amino group, the product molecule contains electron-withdrawing groups at the position C3.

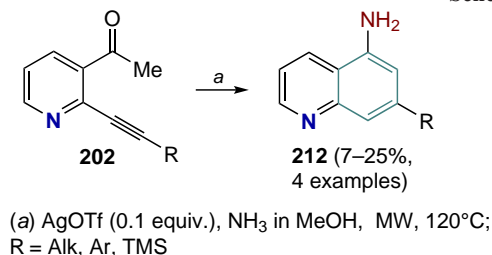
In the second case, silver(I) triflate-catalyzed cyclization of 3-acetyl-2-alkynylpyridines **202** is carried out in the presence of ammonia. The by-products of this reaction are quinolines **212** (7–25%), isolated together with the target 1,6-naphthyridines (Scheme 129).²²³

The use of dialkyl-substituted amines in the presence of molecular sieves provides more efficient cyclization of similar substrates to give C4-amino-substituted acridines and quinolines

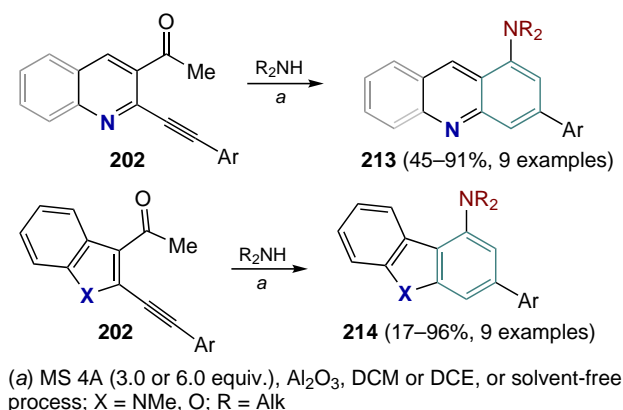
Scheme 128



Scheme 129



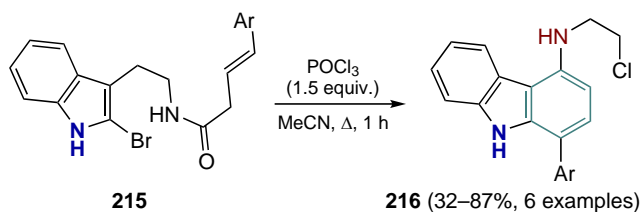
Scheme 130



213, as well as (di)benzofurans and carbazoles **214** (Scheme 130).^{224,225} The yields of acridines and quinolines range from 45% to 91%, dibenzofurans are formed in 17–55% yields, carbazoles are formed in 65–96% yields. Diethylamine and benzylamine can also be used to synthesize carbazoles.

C1-Aryl-C4-aminocarbazoles **216** were obtained from bromine-substituted tryptamines **215** (Scheme 131).²²⁶ The yields of the products varied from 32 to 87%. The use of bromine-free substrates affords carbazoles containing no amino moiety.

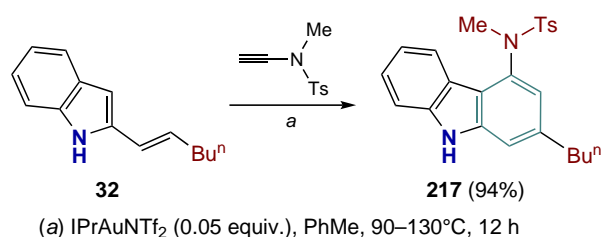
Scheme 131



One example of the preparation of C4-amino-substituted carbazole **217** from 2-alkenylindole **32** has been described (Scheme 132).¹⁷³ The annulation reaction runs in a very high yield using a gold(I) complex as a catalyst.

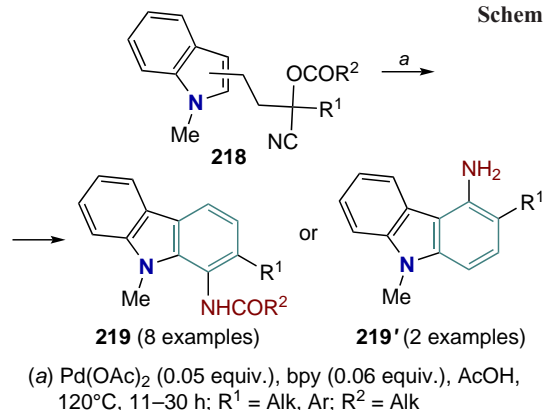
Other approaches to BAHs substituted with amino groups at positions C1 or C4 have also been reported. Thus, palladium(II) acetate-catalyzed cyclization of indoles **218**, which are

Scheme 132



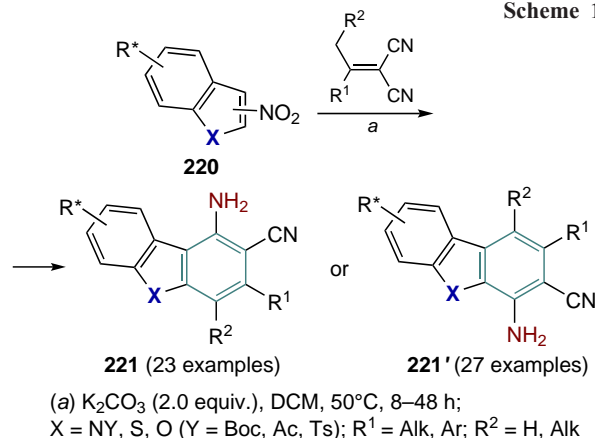
functionalized with a cyanohydrin unit attached to the heterocycle through an alkyl spacer, delivers carbazoles **219** and **219'** (Scheme 133).²²⁷ Depending on the position (C2 or C3) of the alkyl moiety in the substrate, C1-amino- or C4-aminocarbazoles are formed in 75–90% yields. In addition to the amino group, various alkyl or aryl substituents are also present in the products at the *ortho* position relative to it.

Scheme 133

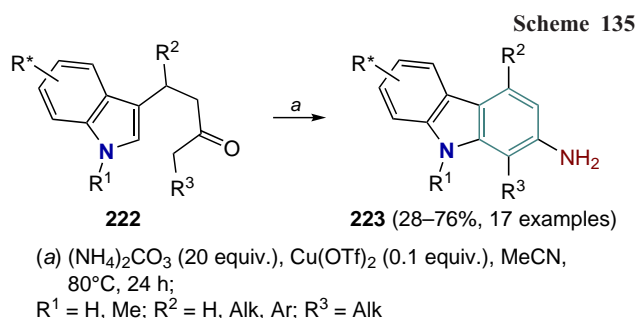


Another method, which is an annulation of nitroindoles **220** with alkylidene malononitriles, provides an access to amino-substituted carbazoles, dibenzothiophenes and dibenzofurans **221**, **221'** (Scheme 134).^{228,229} Depending on the position of the nitro group in the starting compound, products with amino groups at the position C1 or C4 are formed. The newly formed benzene ring is additionally functionalized by the cyano group, various alkyl and aryl moieties. Product yields vary in a wide range (from 5% to 95%).

Scheme 134

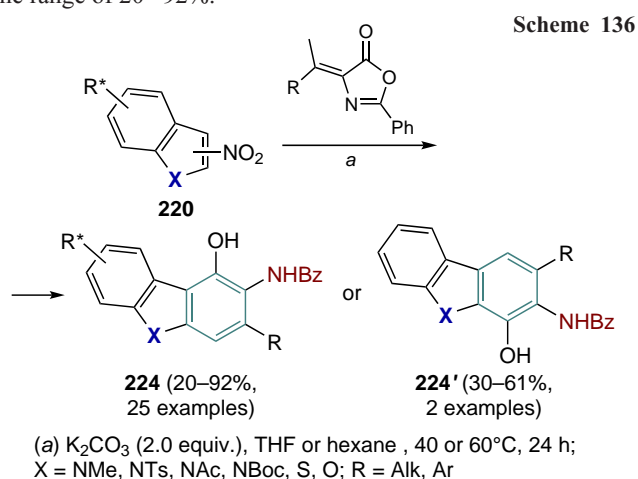


Carbazoles **223** bearing a C2-amino moiety were obtained by a one-pot copper(II) triflate-catalyzed reaction of 1-(1*H*-indol-3-yl)pentan-3-ones **222** with ammonium carbonate (Scheme 135).²³⁰ The process appears to be radical. In addition to the amino group, the benzene ring in the products contains

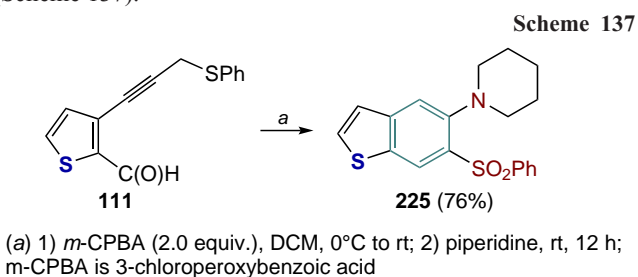


alkyl and aryl substituents in the positions C1 and C4. Resulting aminocarbazoles were obtained in 28–76% yields.

Compounds **224** and **224'**, substituted with a benzoylated amino group at the positions C2 and C3 of the benzene ring, respectively, were obtained by annulation of 3-nitroindoles **220** with alkylidene oxazolones (Scheme 136).²³¹ The position of the functional group in the product depends on the position of the nitro group in the substrate. Along with the amino group, the resulting benzene ring is substituted with a hydroxy group in the position C1 or C4. The yields of the reaction products were in the range of 20–92%.

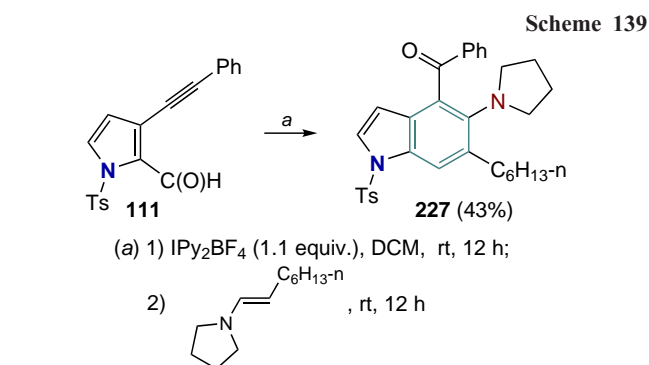
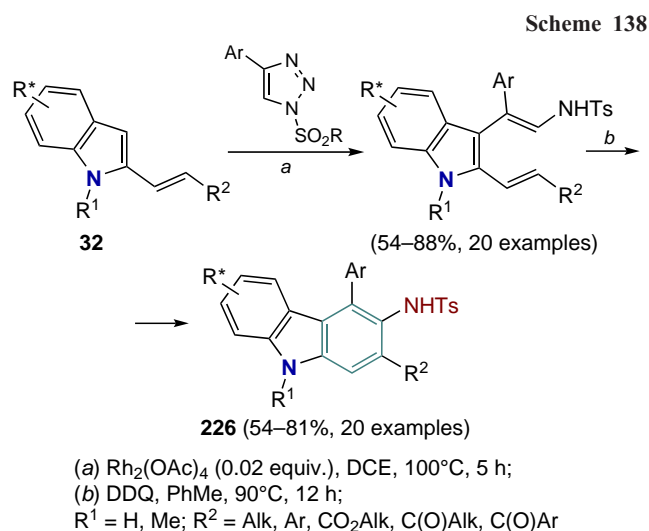


The possibility of obtaining benzo[*b*]thiophene **225** with a C3-piperidine moiety as an amino group from 3-alkynylthiophene-2-carbaldehyde **111** in 76% yield was demonstrated (Scheme 137).²³²



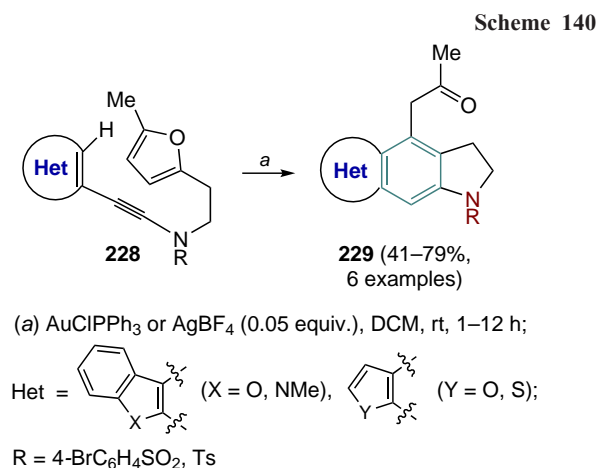
3-Aminocarbazoles **226** were obtained by thermal oxidative electrocyclization of appropriate 2,3-dialkenylindoles (Scheme 138).²³³ The latter compounds were derived from 2-alkenylindoles **32** and *N*-sulfonyl-4-aryltriazoles in the presence of rhodium(II) acetate (54–88%). The yields of the target carbazoles varied from 54% to 81%. Additional substituents in the benzene ring were the C4-aryl group, as well as alkyl, aryl, ester and keto groups at the position C2.

Iodocyclization of 3-alkynylpyrrole-2-carboxaldehyde **111** and enamine based on pyrrolidine and capryl aldehyde affords



C3-pyrrolidine-substituted indole **227** in 43% yield (Scheme 139).¹⁴² This reaction was demonstrated on just one example.

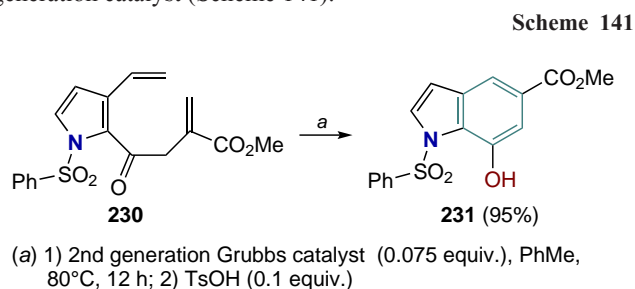
The synthesis of C2(C3)-amino-substituted benzo[*b*]furans, benzo[*b*]thiophenes, carbazoles, and dibenzofurans **229**, in which the nitrogen atom is part of a five-membered ring annulated with a benzene ring, is described. The method is based on benzoannulation of arylalkynylamide derivatives **228**, which contain a furan ring in the peripheral substituent, in the presence of a gold(I) catalyst (Scheme 140).²³⁴ A key factor for obtaining the target products is the length of the linker between the amino group and the furan moiety, which must include two methylene units. An increase in the number of methylene units in the linker resulted in the formation of



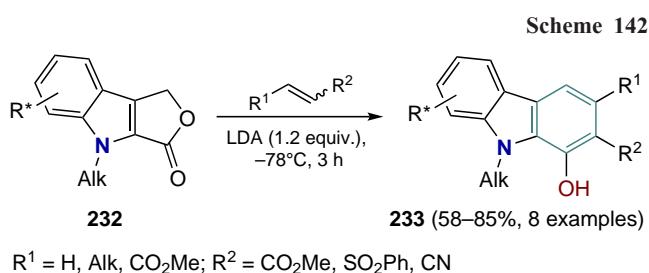
other products. Amino-substituted heterocycles were obtained in 41–79% yields.

4.2. Hydroxy-substituted heterocycles

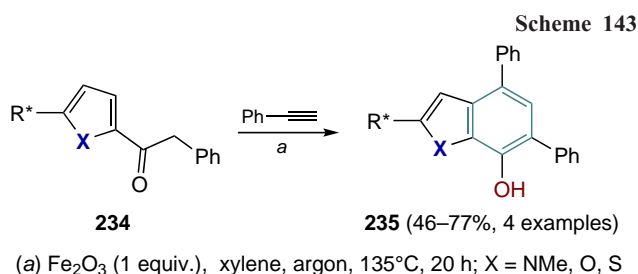
Synthetic approaches to benzoannulated heterocycles containing a C1-hydroxy group are presented below. The possibility of synthesizing hydroxyindoles was demonstrated using product **231** as an example, which was obtained in 95% yield by metathesis of olefin **230** in the presence of a Grubbs second-generation catalyst (Scheme 141).¹²⁷



C1-Hydroxycarbazoles **233** were synthesized *via* [4+2] cycloaddition of furoindoles **232** and various Michael acceptors in the presence of LDA (Scheme 142).²³⁵ Various electron-withdrawing groups acted as additional substituents at the C2 and C3 atoms. The yields of the products were 58–85%.

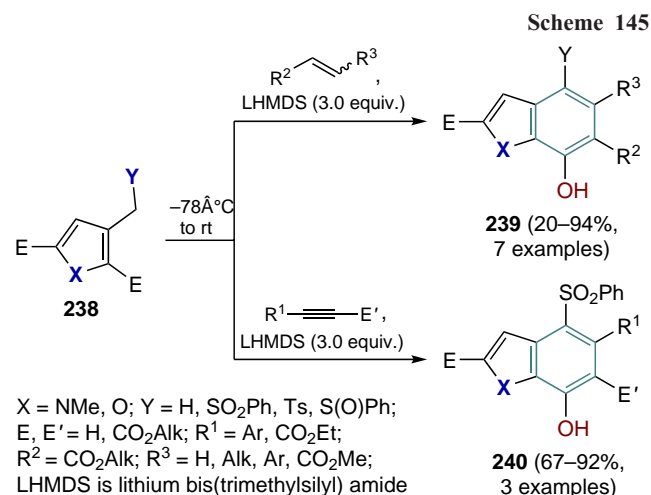
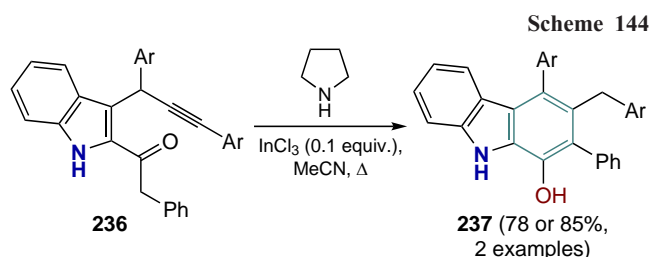


Several examples of syntheses of C1-hydroxy-C2,C4-diphenyl-substituted indoles, benzo[*b*]thiophenes and benzo[*b*]furans have been reported (Scheme 143).²³⁶ The iron(III) oxide-promoted cycloaddition of phenylacetylene to substrates **234** gives regioselectively compounds **235** in 46–77% yields.



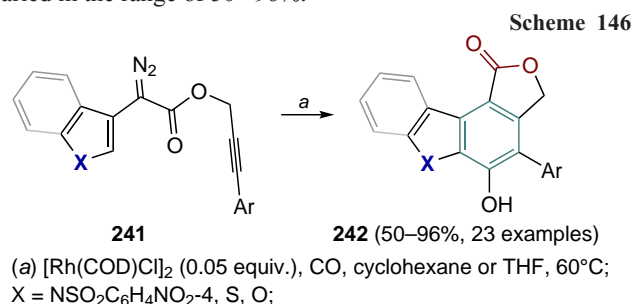
The synthesis of C1-hydroxycarbazoles **237** is illustrated by two examples of benzoannulation of 2-acyl-3-propargylindoles **236** with pyrrolidine using indium(III) chloride as a catalyst (Scheme 144).²²²

The synthesis of C1-hydroxyindole and benzo[*b*]furan derivatives by nucleophilic annulation of 2,3-disubstituted heterocycles **238** with Michael acceptors has been reported (Scheme 145).²³⁷ Various alkenes and alkynes were used as the latter; the yield of compounds **239** and **240** were 20–94% and 67–92%, respectively. The method is regioselective and, as will



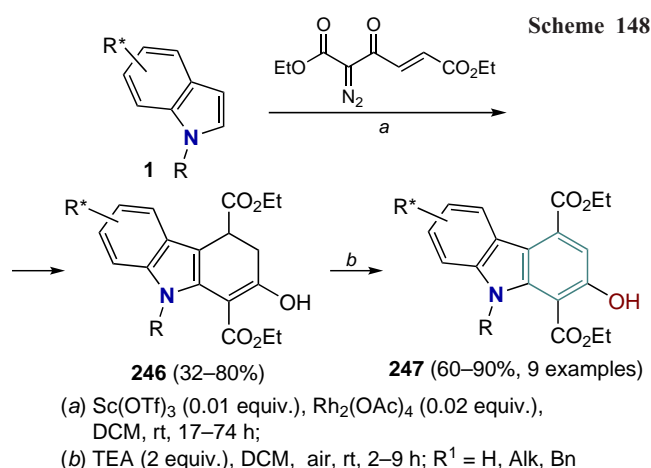
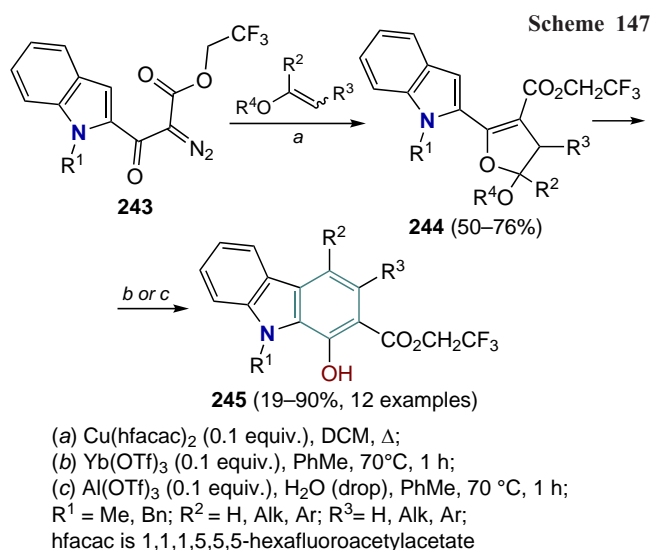
be shown below, is suitable for the synthesis of C4-hydroxy-substituted carbazoles (see Scheme 158).

Carbazoles, dibenzothiophene and dibenzofuran, annulated with furanone and containing an OH group at the C1 position of the benzene ring, were synthesized by rhodium(I)-catalyzed benzoannulation of heterocyclic propargyl derivatives of diazoacetate **241** (Scheme 146).²³⁸ The yields of products **242** varied in the range of 50–96%.

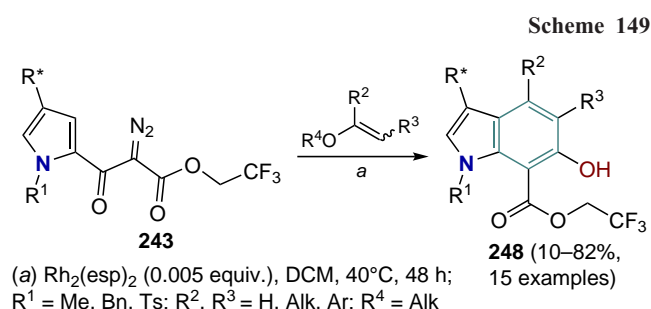


Lewis acid-catalyzed benzoannulation of 5-(indol-2-yl)-2,3-dihydrofuran acetals **244** delivers C1-substituted carbazoles **245** in 19–90% yields (Scheme 147).²³⁹ Substrates **244** were obtained by the copper(II)-catalyzed reaction of *N*-indolyl- α -diazo- β -ketoesters **243** with enol esters.

There are several approaches to BAHs with a hydroxyl group at the C2 position of the resulting benzene ring. Only three such examples have been described for indole and carbazole. Thus, C2-hydroxycarbazoles **247** were synthesized by a two-step process from 2,3-unsubstituted indoles and a diazo derivative of acetoacetic ester (Scheme 148).²⁴⁰ The reaction occurs with simultaneous tandem catalysis by scandium(III) triflate (Michael reaction) and rhodium(II) acetate (cyclization *via* a carbene intermediate) followed by treatment of the intermediate **246** with triethylamine. The transformation products are carbazoles bearing additional ester groups at positions C1 and C4 (60–90%).

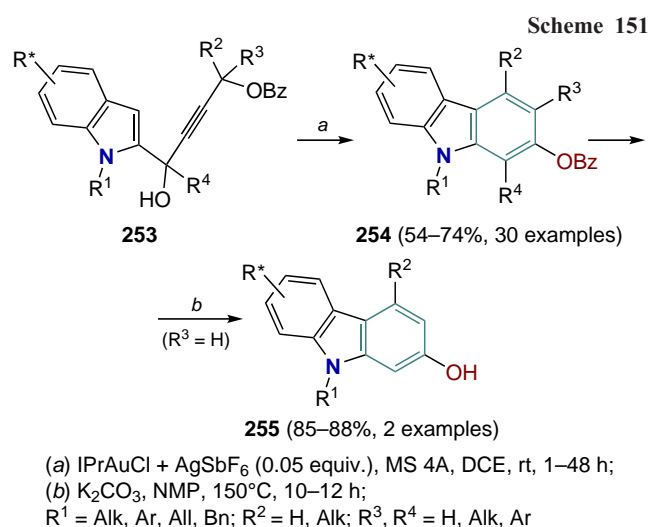
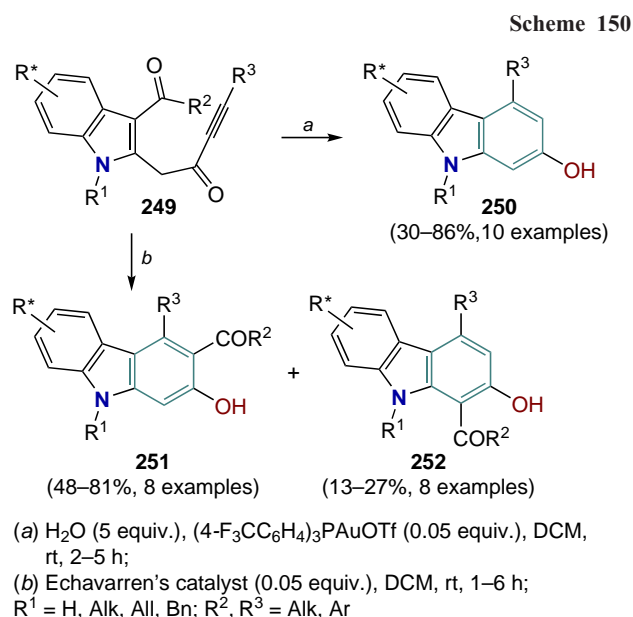


The rhodium(II) acetate-catalyzed annulation of pyrrolyl- α -diazo- β -ketoesters **243** with enol ethers affords C2-hydroxyindoles **248** in 10–82% yields (Scheme 149).²⁴¹



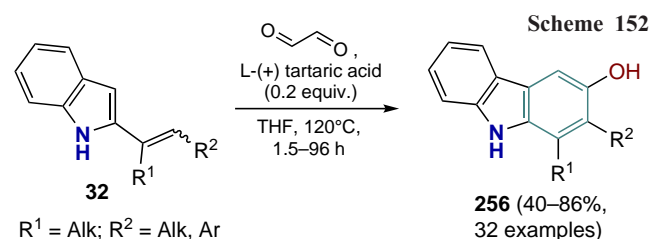
C2-hydroxy-substituted carbazoles can be obtained by gold(I)-catalyzed cycloisomerization of functionalized indoles **249** (Scheme 150).²⁴² Depending on the type of the catalytic complex used, products **250** (30–86%) or their analogues **251** and **252** with an additional keto group at position C1 or C3 (yields 48–81% for the main product) are formed.

For the synthesis of C2-hydroxy-substituted carbazoles **254**, gold(I)-catalyzed cyclization of substituted 4-benzoyloxy-1-(indol-2-yl)-2-alkynols **253** was used (Scheme 151).²⁴³ The yields of the products varied from 54% to 74%. It was shown that the debenzoylation can be effectively carried out in the



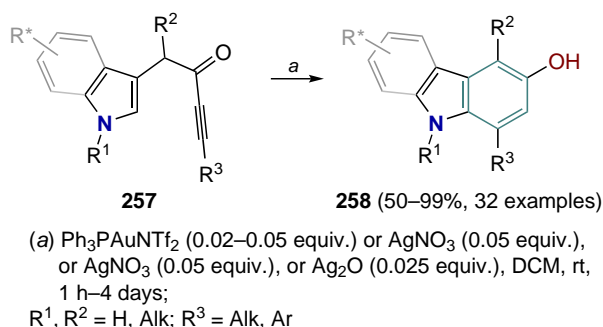
presence of potassium carbonate in *N*-methyl-2-pyrrolidone (NMP) to give hydroxy derivatives **255**.

Four methods are known for the preparation of BAHs (carbazoles and dibenzofuran) substituted with an OH group at the C3 position. A Brønsted acid-catalyzed cascade one-pot benzoannulation was reported for 2-alkenylindoles **32** (Scheme 152).²⁴⁴ The latter react with glyoxal to give C1,C2-disubstituted C3-hydroxycarbazoles **256** in 40–86% yields.



Based on indolynones **257**, the synthesis of indoles and carbazoles **258** was carried out in the presence of gold complexes^{245,246} and silver complexes⁴⁰ (Scheme 153). C3-Hydroxy-substituted indoles are formed in 64–95% yields under the action of silver(I) nitrate, whereas the corresponding

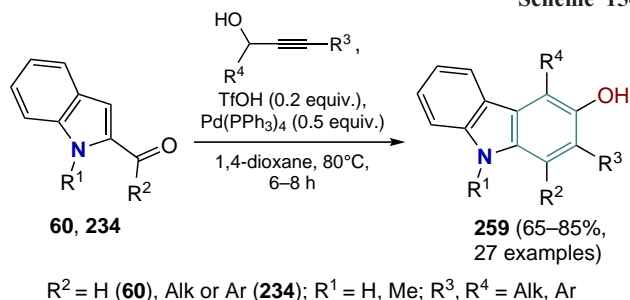
Scheme 153



carbazoles are formed in the presence of a gold(I) complex (50–99%).

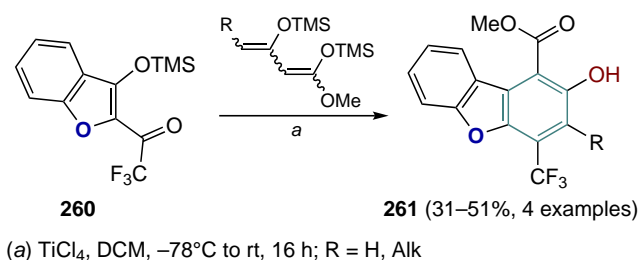
The assembly of C3-hydroxycarbazoles *via* a cascade of propargylation and palladium(0)-catalyzed annulation of 2-carbonyl-substituted indoles with propargyl alcohols has been described (Scheme 154).²⁴⁷ This method provides an access to carbazoles **259** bearing additional substituents (aryl and alkyl groups) at positions C1, C2 and C3. The yields of such products range from 65% to 85%.

Scheme 154



1,3-Bis(trimethylsiloxy)buta-1,3-dienes react with functionalized benzofuran **260** to give tetrasubstituted dibenzofurans **261** in moderate yields (31–51%) (Scheme 155).²⁴⁸ The benzene ring in the products contains, in addition to hydroxy group, trifluoromethyl, alkyl and ester moieties.

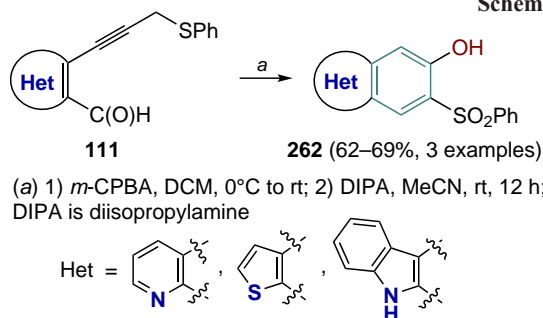
Scheme 155



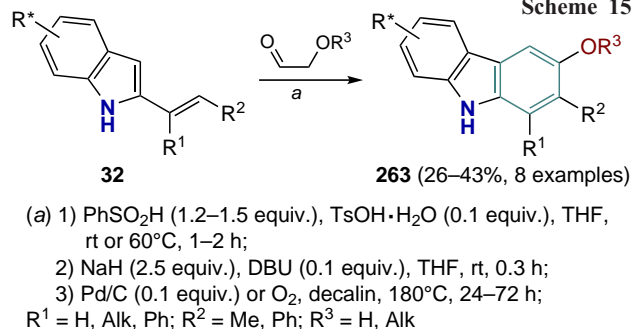
Carbazole, quinoline, and benzo[*b*]thiophene bearing a hydroxy group at the C3 atom of the benzene ring were obtained by a two-step procedure from the appropriate formylethynyl-substituted heterocycles **111** containing alkynyl and formyl groups in adjacent positions (Scheme 156).²³² The yields of products **262** ranged from 62–69%.

Another method for the synthesis of C3-hydroxy-substituted carbazoles **263** is based on a three-step reaction between alkenyl-substituted indoles **32** and hydroxyethanal (or its alkylated derivatives) (Scheme 157).⁷⁸ Carbazoles **263**, containing methyl and phenyl groups as additional substituents in the benzene ring, were obtained in 26–43% yields.

Scheme 156



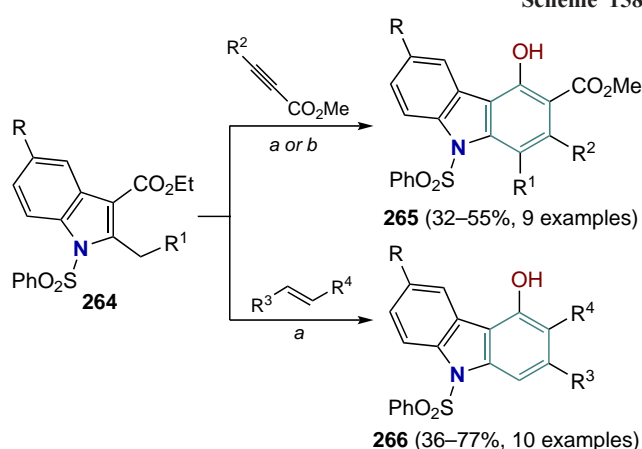
Scheme 157



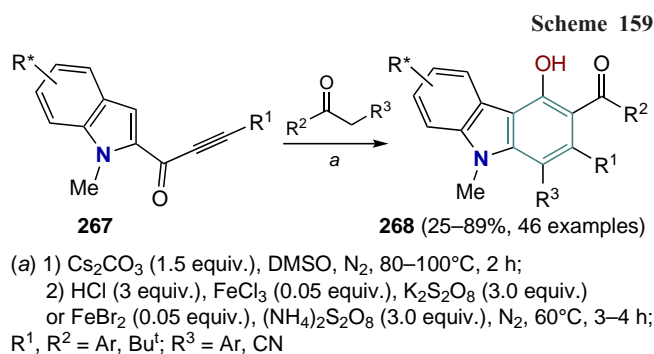
A large number of protocols have been developed for the preparation of C4-hydroxy-substituted BAHs, primarily carbazoles. The six following Schemes illustrate approaches to such BAHs bearing additional carbonyl groups in the resulting benzene ring. For example, the reaction of ester-substituted indole **264** with Michael acceptors in the presence of a base gives substituted carbazoles **265** and **266** (Scheme 158).²⁰⁹ When an alkyne is used in this reaction, the electron-withdrawing substituent (R^1) at the methyl group in the α -position of indole **265** is retained (32–55%), while in the case of alkenes, C4-hydroxycarbazoles **266** with the vacant C1 position are produced (36–77%). A similar method is also suitable for the preparation of indoles with a C1-hydroxy group (see Scheme 145).²³⁷

A method for the synthesis of tetrasubstituted carbazoles **268** from hetarylnones **267** and α -unsubstituted ketones in the

Scheme 158

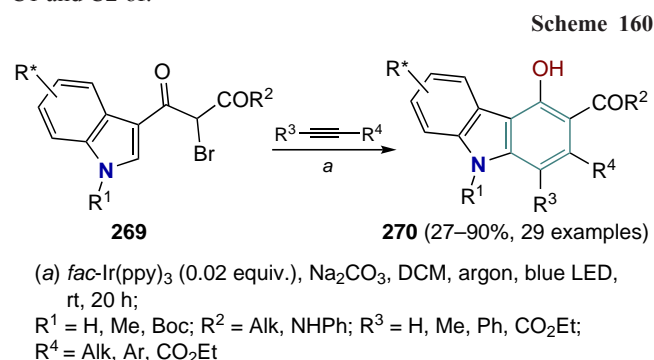


(a) Bu^tOK (1.0 equiv.), THF, 0°C , 2–5 h;
 (b) 1) LDA (3.0 equiv.), THF, -78°C ;
 2) DMAD (2.0 equiv.), -78°C to rt, 2 h;
 $\text{R}^1 = \text{H}, \text{CO}_2\text{Me}, \text{Ar}, \text{CN}; \text{R}^2 = \text{H}, \text{CO}_2\text{Me}; \text{R}^3 = \text{H}, \text{Alk}, \text{Ar};$
 $\text{R}^4 = \text{CHO}, \text{C(O)Me}, \text{CO}_2\text{Me}, \text{CN}, \text{NO}_2$;
 DMAD is dimethyl acetylenedicarboxylate

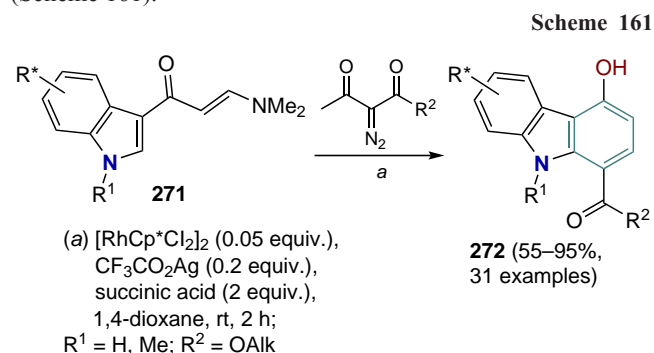


presence of a base followed by oxidation is described (Scheme 159).^{249,250} The one-pot protocol allows product yields of up to 89%. The position C3 of the benzene ring of such carbazoles contains the acyl group.

Scheme 160 illustrates an approach to carbazoles **270** based on the blue light-photocatalyzed reaction of indole-based bromoketones **269** with alkynes in the presence of an iridium catalyst.²⁵¹ This approach allows the preparation of three- and four-substituted carbazoles (27–90%). The benzene ring in such products can bear electron-withdrawing substituents at positions C1 and C2 of.

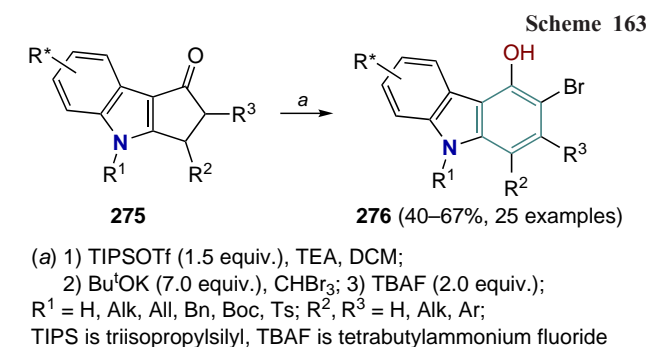
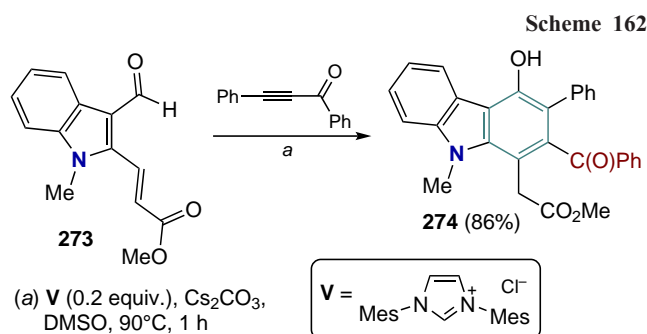


The rhodium(II)-catalyzed [5+1]-annulation of indole enaminones **271** with diazo compounds leads to C4-hydroxycarbazole carboxylates **272** in yields from 55% to 95% (Scheme 161).²⁵²



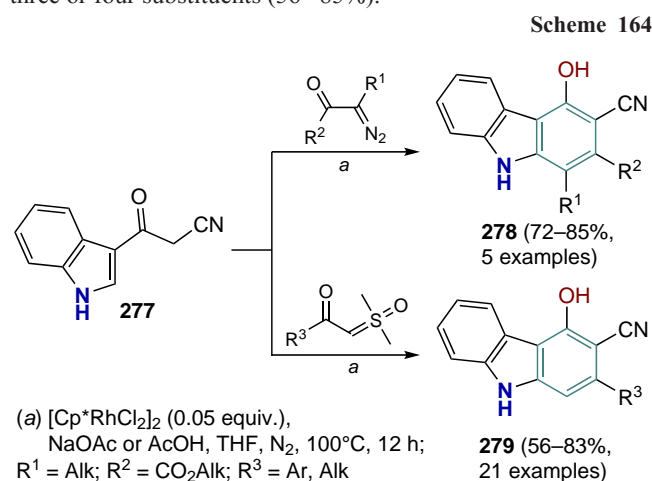
One example of the synthesis of C2-benzoyl-substituted carbazole **274** by the reaction of Michael acceptor-tethered 3-formylindole **273** and 1,3-diphenylprop-2-yn-1-one has been reported (Scheme 162).²⁵³ The intermolecular Stetter cyclization catalyzed by the *N*-heterocyclic carbene **V** affords the carbazole in 86% yield.

C3-bromosubstituted C4-hydroxycarbazoles **276** were obtained by benzoannulation of indolocyclopentanones **275**, during which the benzene ring is formed *via* the expansion of the



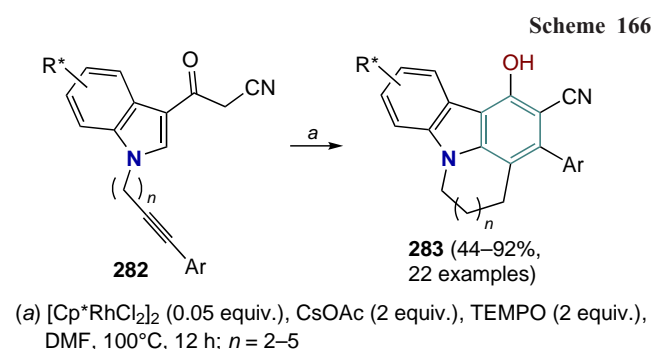
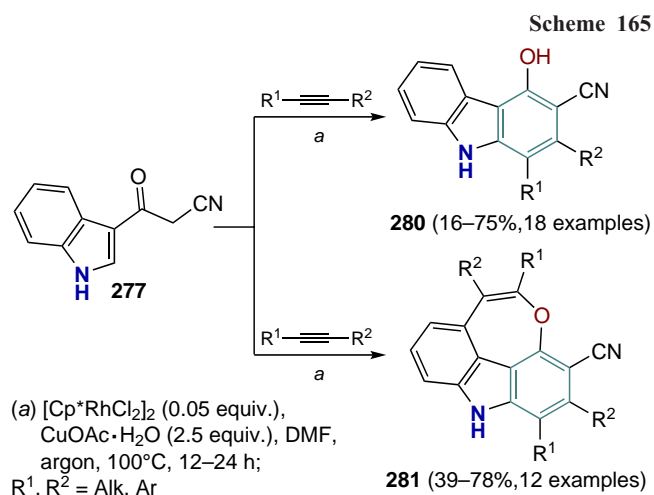
five-membered ring (Scheme 163).²⁵⁴ The yields of the products ranged from 40 to 67%.

A separate group of methods includes various variants of the rhodium(II)-catalyzed annulation of 3-(indol-3-yl)-3-oxopropanenitrile **277** to afford C4-hydroxycarbazoles with an additional nitrile group in the benzene ring. It has been demonstrated that the annulation of compound **277** with sulfoxonium ylides or α -diazo ketones furnishes functionalized carbazoles **278** and **279** (Scheme 164).²⁵⁵ Depending on the type of reagent, this method can be used to obtain products bearing three or four substituents (56–85%).

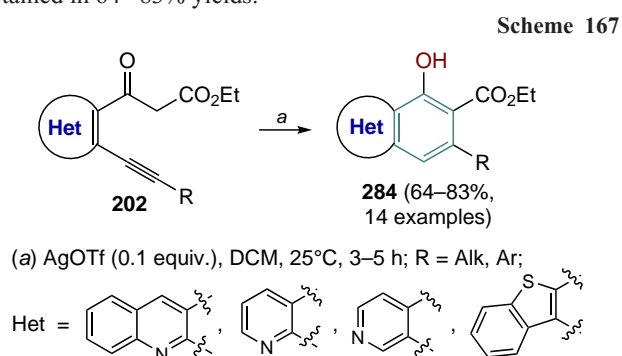


Alkynes are also used as reagents for the catalytic annulation of compound **277** (Scheme 165).²⁵⁶ Depending on the reaction time, carbazoles **280** (16–75%) or 4*H*-oxepino[2,3,4,5-*def*] carbazoles **281** (39–78%) are formed as a result of additional annulation *via* CH-activation of carbazole.

A similar method afforded six-, seven-, and eight-membered hydroazepino[3,2,1-*jk*]carbazoles **283** starting from substrates **282** containing a carbon–carbon triple bond in the substituent at the nitrogen atom (Scheme 166).²⁵⁷ The yields of the products in this case range from 44 to 92%.

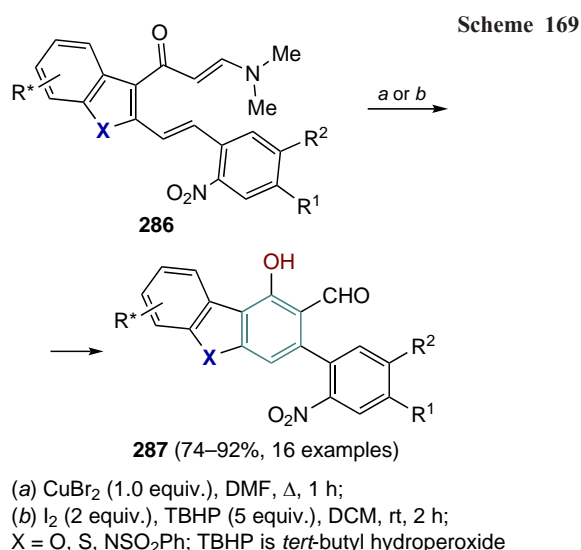
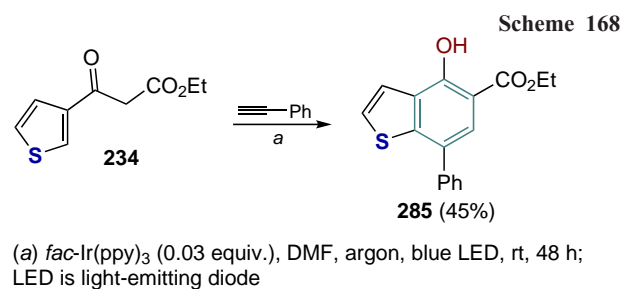


Among other approaches to C4-hydroxy-substituted BAHs, mention should be made of the silver(I) triflate-catalyzed cyclization of 3-(2-alkynyl)hetaryl- β -ketoesters **202**, which affords compounds **284**, C4-hydroxy-functionalized acridines and quinolines, as well as C1-substituted dibenzothiophene (Scheme 167).²⁵⁸ The C2 position of the benzene ring of these compounds may be substituted by alkyl and aryl groups, and the C3 atom may contain an ester functionality. The products were obtained in 64–83% yields.



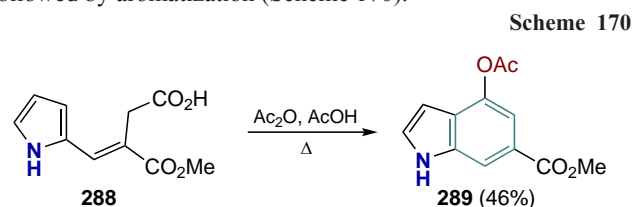
One example of the synthesis of C1-hydroxy-substituted benzothiophene **285** is known. It is formed by the reaction of β -ketoester **234** with phenylacetylene in 45% yield (Scheme 168).²⁵⁹

A one-pot method for the preparation of C4-hydroxy-substituted carbazoles,²⁶⁰ dibenzofurans and dibenzothiophenes²⁶¹ (**287**) based on the cyclization of functionalized heterocycles **286** is described (Scheme 169). The reaction is promoted by copper(II) bromide or iodine in the presence of an oxidant, followed by thermal electrocyclization and aromatization to give products in 74–92% yields. In addition to



the hydroxy group, the newly-formed benzene ring contains aryl and formyl moieties at positions C2 and C3, respectively.

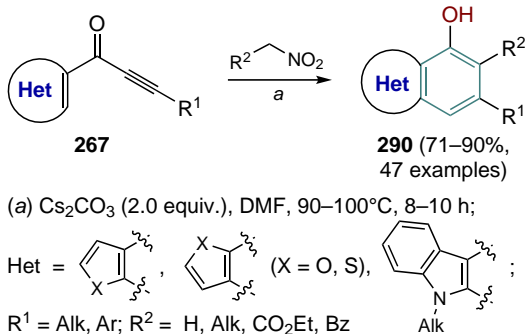
Methyl C4-acetoxy-substituted indole-5-carboxylate **289** was also obtained by cyclization of functionalized pyrrole **288** followed by aromatization (Scheme 170).²⁶²



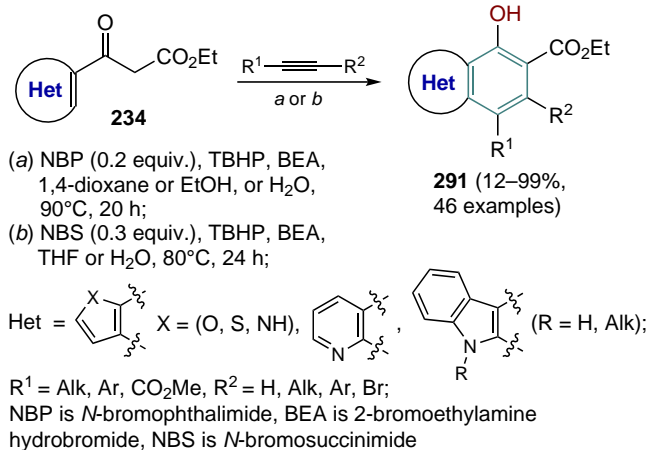
The methods presented below afford, depending on the structure of the substrate, benzoheterocycles bearing a hydroxy group at position C1 or C4. For example, the use of hetarylnones **267** and various nitroalkanes opens the way to hydroxy-substituted carbazoles, benzo[*b*]thiophenes, and benzo[*b*]furans (**290**) in yields from 71 to 90% (Scheme 171).^{205,206} This method is a modification of the one discussed above in Scheme 115.^{41,204} The structure of the nitroalkane determines the structure of the product and makes it possible to obtain heterocycles with additional alkyl, aryl or carbonyl (carboxyl) moieties in 71–90% yields.

The synthesis of C1(C4)-hydroxy-substituted BAHs *via* annulation of hetarylethanones **234** with alkynes has been reported (Scheme 172).^{263,264} This method provided regioselective synthesis of a variety of products **291**, *vis.*, derivatives of carbazole, indole, quinoline, benzo[*b*]thiophene and benzo[*b*]furan, in yields ranging from 12 to 99%. Phenyl and ester moieties were present as additional substituents at positions C1 and C3 of the benzene ring of such compounds.

Scheme 171

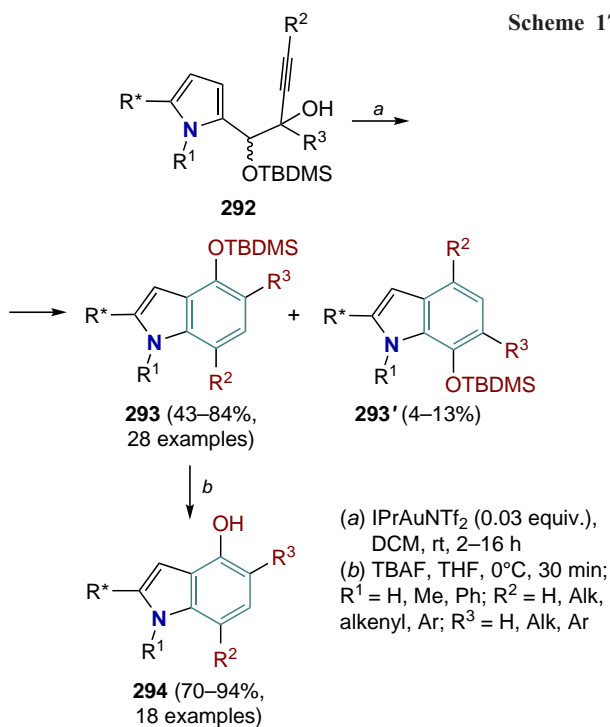


Scheme 172



A method for the synthesis of C4-silyloxyindoles **293** by gold(I)-catalyzed annulation of O-silylated pyrrolyl glycols **292** has been developed (Scheme 173).⁷⁹ The yields ranged from 43 to 84%, with C1-silyloxyindoles **293'** isolated as minor products (4–13%). The possibility of effective desilylation with tetrabutylammonium fluoride was demonstrated. The benzene

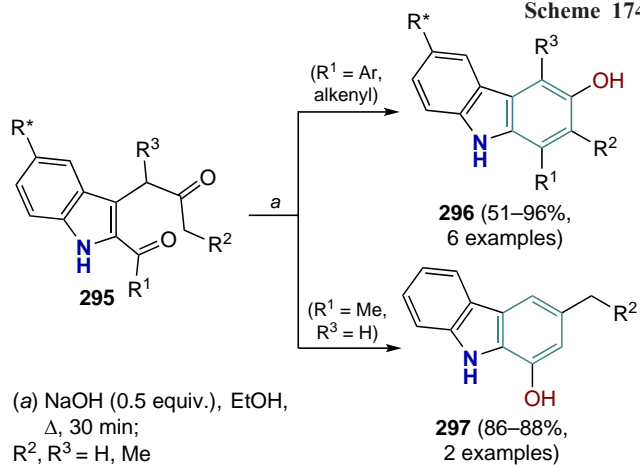
Scheme 173



ring of compounds **293** and **294** contains alkyl, alkenyl, and/or aryl substituents. Another approach to hydroxy-substituted carbazoles, dibenzothiophenes, and dibenzofurans with additional amino substituents was discussed above (see Scheme 136).²³¹

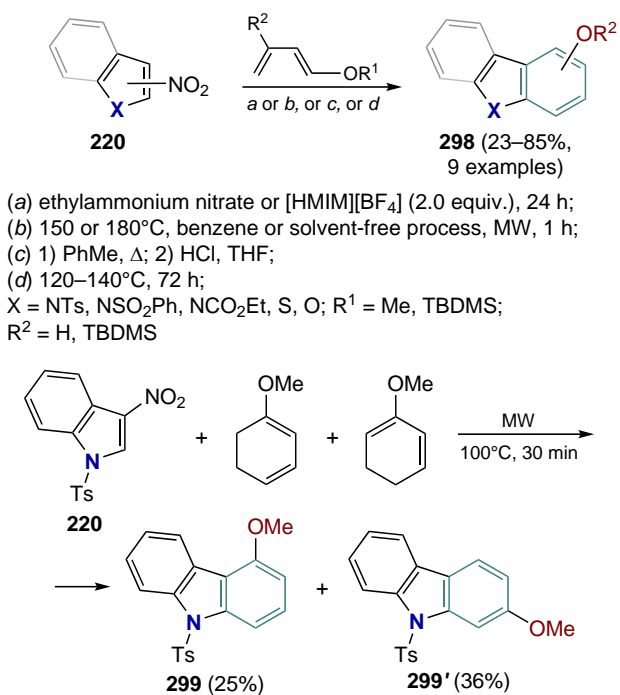
The following schemes show approaches that have been developed for the synthesis of BAHs substituted with a hydroxy group predominantly at the position C2 or C3. For example, cyclization of indoles **295** bearing two carbonyl groups affords carbazoles **296** containing an OH group at the position C3 in 51–96% yields (Scheme 174).⁸⁰ When the indole bears an acetyl group at the position 2, C1-hydroxycarbazoles **297** are formed (86–88%).

Scheme 174



The Diels–Alder reaction between Danishefsky’s diene or its analogues with heterocycles **220** containing a nitro group at the position 2 or 3 furnishes various BAHs **298** with a hydr(or silyl)-oxy group in the benzene ring. Using this method, C2- and C3-hydroxy-substituted indoles (43–47%),²⁶⁵ as well as carbazoles (23–85%),²⁶⁶ dibenzothiophenes (52–66%)²⁶⁷ with an OH group at the position C3 or C2, and C3-substituted dibenzofuran (85%)²⁶⁸ were obtained (Scheme 175). Depending on the

Scheme 175

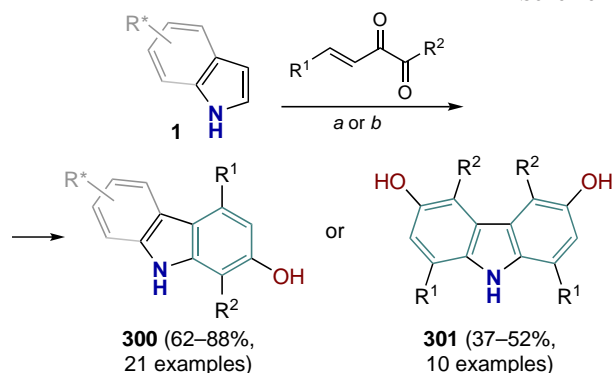


structure of the starting heterocycles, this reaction requires different conditions, *e.g.*, heating, microwave activation or the use of protic ionic liquids.

This strategy was used to obtain methoxy-substituted carbazoles.²⁶⁹ For example, the reaction of 3-nitro-1-(*p*-tolylsulfonyl)indole **220** with a mixture of 1-methoxy- and 2-methoxycyclohexa-1,3-dienes gave C4- and C2-methoxycarbazoles **299** and **299'** in 25 and 36% yields, respectively.

Annulation of 2,3-unsubstituted indoles with but-3-ene-1,2-diones in the presence of copper(II) triflate leads to C2-hydroxycarbazoles **300** in 62–88% yields (Scheme 176).²⁷⁰ The substrate in this reaction can also be pyrrole, which forms carbazoles **301** bearing two OH moieties (37–52%).

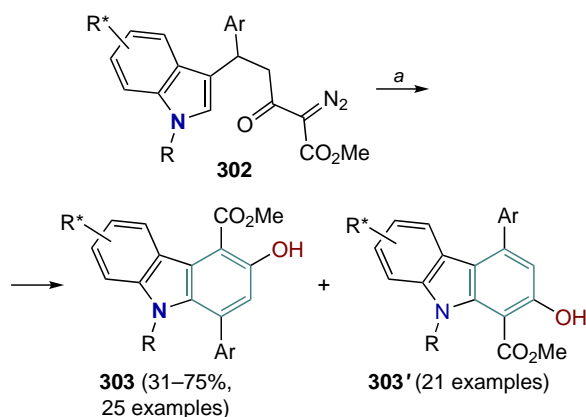
Scheme 176



(a) Cu(OTf)₂ (0.1 equiv.), 1,4-dioxane, 80°C, 3–5 h;
(b) Cu(OTf)₂ (0.2 equiv.), TsOH (0.1 equiv.), 1,4-dioxane, 70°C, 4 h;
R¹, R² = Alk, Ar

The following method allows the preparation of carbazoles **303** with hydroxy groups at the C3 position *via* copper(II) triflate-catalyzed annulation of indolyl- α -diazocarbonyl compound **302** in 31–75% yields (Scheme 177).²⁷¹ The minor products are C2-hydroxy-substituted carbazoles **303'**. Additional substituents on the benzene ring of the products are ester and aryl groups.

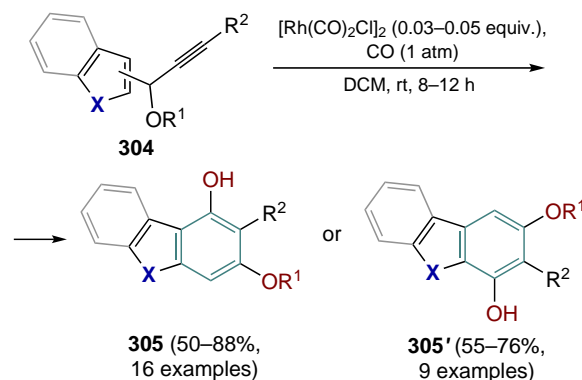
Scheme 177



(a) Cu(OTf)₂ (0.2 equiv.), Bu₄NClO₄ (1.0 equiv.), PhMe, N₂, 80°C;
R = H, Alk, Ph

Methods for the synthesis of carbazoles with two carboxyl groups in the resulting benzene ring are represented by two examples. A versatile approach to various BAHs (indoles, carbazoles, benzothiophenes, benzofurans and dibenzofurans) containing one free and one protected hydroxy group was developed on the basis of a rhodium(II)-catalyzed

Scheme 178

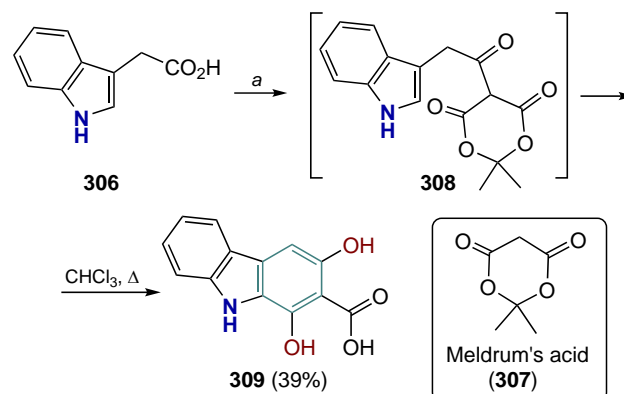


X = NSO₂Ar, NBn, NAc, O, S; R¹ = Piv, Ac, Bn; R² = H, Me, CO₂Et

benzoannulation (Scheme 178).^{272,273} Hetaryl-substituted but-3-yn-1-ols **304** give compounds **305** or **305'** in 50–88% yields. Depending on the position of the functional group in the substrate (2 or 3), both 1,3- and 2,4-dihydroxy-substituted products are formed.

Carbazole **309** with two hydroxy substituents at positions C1 and C3 was obtained by a two-step reaction of indol-3-ylacetic acid **306** with Meldrum's acid (**307**) *via* the intermediate **308** (Scheme 179).²⁷⁴ The yield of the product was 39%.

Scheme 179



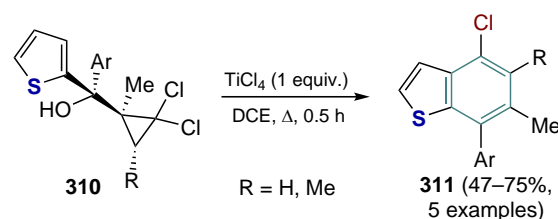
(a) **307** (1.2 equiv.), EDC, TEA, CHCl₃, Δ;
EDC is 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide

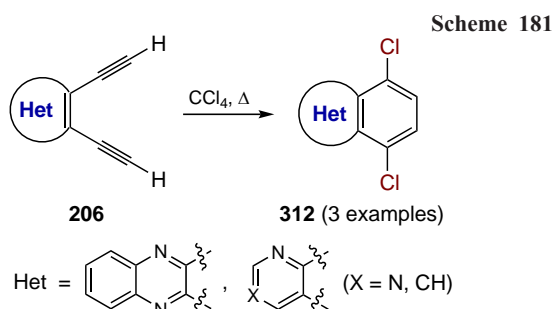
4.3. Halogenated heterocycles

This Section considers the methods for the benzoannulation of aromatic heterocycles to give products containing halogen atoms in the resulting benzene ring. The possibility of obtaining C4-chlorobenzo[*b*]thiophenes **311** by the reaction of (2,2-dichlorocyclopropyl)(2-thienyl)methanols **310** in the presence of a Lewis acid is demonstrated (Scheme 180).²⁷⁵ The yields of the products vary from 47 to 75%.

Thermal Bergman cyclization of 1,2-diethynylhetarenes **206** in tetrachloromethane gives chloro-substituted quinazoline and

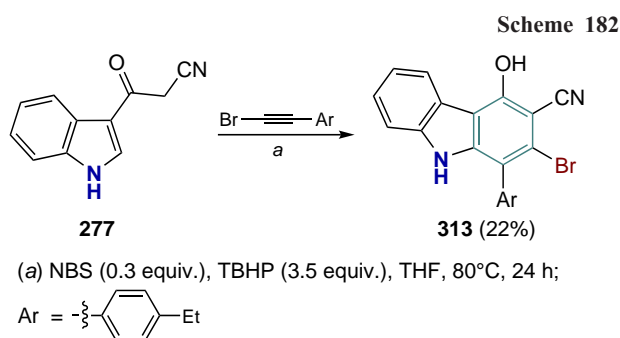
Scheme 180



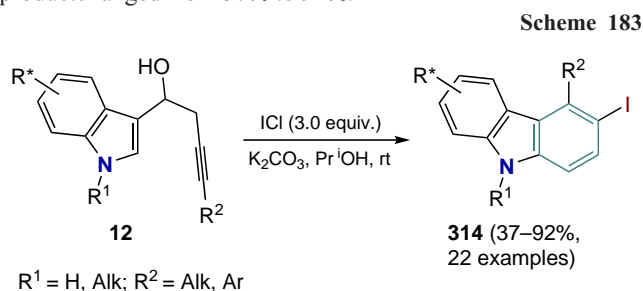


phenazine **312** in 15 and 81% yields, respectively (Scheme 181).²⁷⁶ Using quinoline as a substrate, a difficult-to-separate mixture of products is obtained, containing the target C1,C4-dichloro-substituted heterocycle.

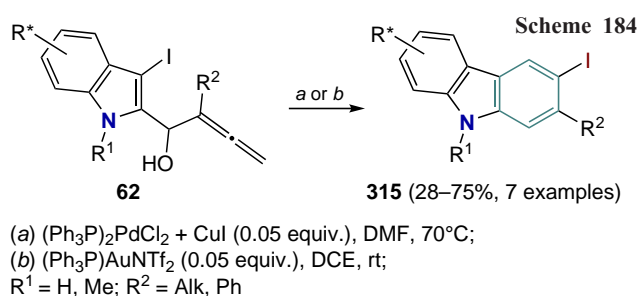
The reaction of the cyano-substituted acetyl indole **277** with a bromo-substituted alkyne in the presence of *N*-bromosuccinimide afforded carbazole **313** in 22% yield (Scheme 182).²⁶⁴ The synthesis of carbazoles containing a bromine atom in the benzene ring was also accomplished using the methods presented above in Scheme 163 (C3-bromo).²⁵⁴



A series of methods have been developed for the preparation of carbazoles bearing C2 and/or C3 iodine atoms. To get access to C3-iodocarbazoles **314**, iodocyclization of 1-(indol-3-yl)but-3-yn-1-ols **12** was carried out (Scheme 183).²⁷⁷ The yields of the products ranged from 37% to 92%.

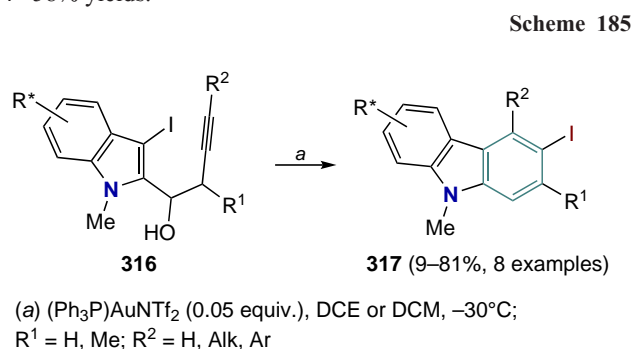


C3-Iodocarbazoles **315** can be assembled from 3-iodoindol-2-yl-containing allenes **62** using transition metal catalysis

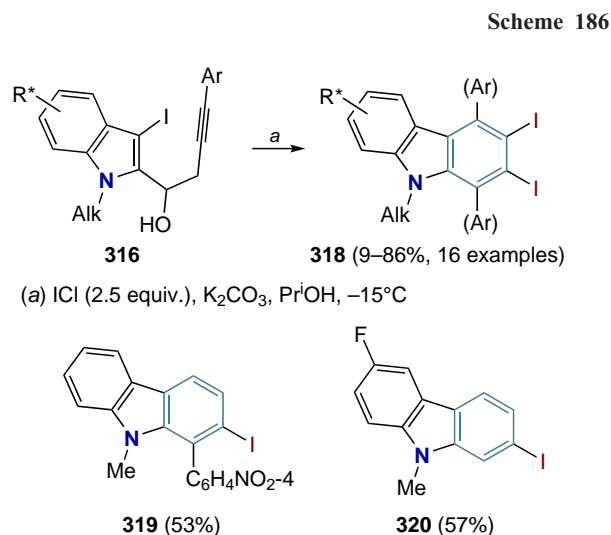


(Scheme 184).²⁷⁸ In the case of the palladium(II) complex, selective formation of the target iodocarbazoles in 50–75% yields was observed. In the presence of the gold(I) complex, along with the major product (28–69%), iodine-free carbazoles were isolated (7–38%).

A similar method for the preparation of C3-iodocarbazoles **317** was demonstrated for (3-iodoindolyl)butynols **316** (Scheme 185).²⁷⁹ The reaction is catalyzed by a gold(I) complex to give the target products in yields ranging from 9% to 71%. It is worth noting that when using bis(indolyl)-1,3-diyne, sequential cyclization occurs to give bis(iodocarbazoles) in 24–58% yields.²⁸⁰

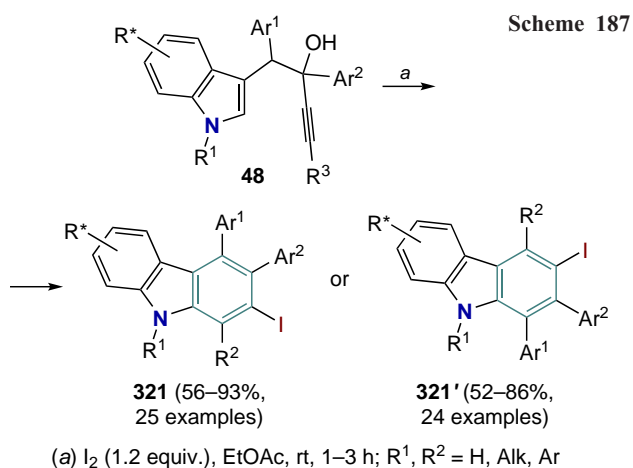


2,3-Diiodocarbazoles **318** were obtained from substrates **316** in 16–86% yields in the presence of iodine(I) chloride and potassium carbonate (Scheme 186).²⁸¹ In the presence of an electron-deficient aryl substituent at the triple bond or a fluorine atom in the indole ring, only C2-substituted iodocarbazoles **319** and **320** are formed.



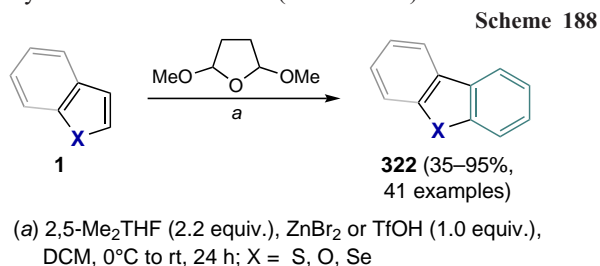
The next method allows for the synthesis of C2- (**321**, Ref. 282) or C3-iodocarbazoles (**321'**, Ref. 283) by cycloisomerization of aryl(indol-3-yl)methyl-substituted propargyl alcohols **48** (Scheme 187).

In the case where the Ar¹ substituent in the substrate is a benzene derivative, the reaction affords C2-substituted iodocarbazoles **321** (56–93%). The use of substrates substituted with 3-indolyl or trimethoxyphenyl groups in the reaction opens the way to C3-substituted iodocarbazoles **321'** (52–86%). Copper(II) triflate-promoted cyclization of the same substrates gives carbazoles containing alkyl and aryl substituents in the benzene ring (60–96%).²⁸⁴

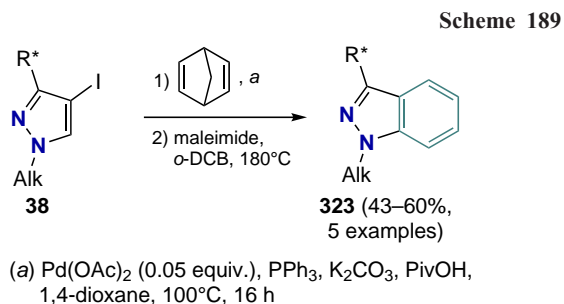


5. Unsubstituted benzoheterocycles and benzoheterocycles with variously substituted benzene ring

This Section highlights methods for obtaining BAHs with different substituents in the benzene ring, not covered in the previous Sections. The products unsubstituted on the benzene ring are accessible by several approaches. Acid-catalyzed annulation of 2,3-unsubstituted heterocycles with 2,5-dimethoxytetrahydrofuran affords derivatives of benzo[*b*]-thiophene, dibenzothiophene, dibenzofuran, dibenzoselenophene **322** in yields from 35% to 95% (Scheme 188).²⁸⁵

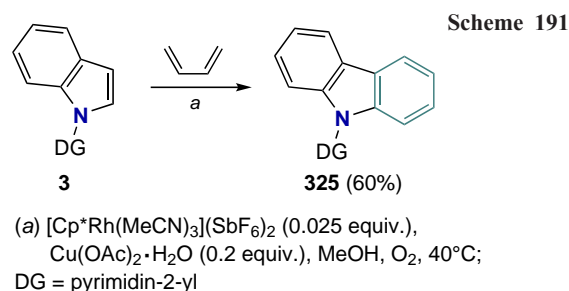
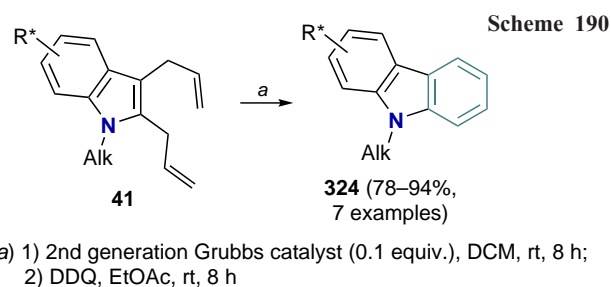


Indazoles **323** with an unsubstituted benzene ring were obtained by annulation of 4-iodopyrazole **38** with two norbornadiene molecules in the presence of palladium(II) acetate followed by a retro-Diels–Alder reaction (75–84% yields in the second step) (Scheme 189).²⁸⁶ The study⁹⁹ showed the applicability of this method to obtain other annulated heterocycles (carbazoles, benzothiophenes, dibenzothiophenes and benzofurans) without performing the second step, the retro-Diels–Alder reaction.

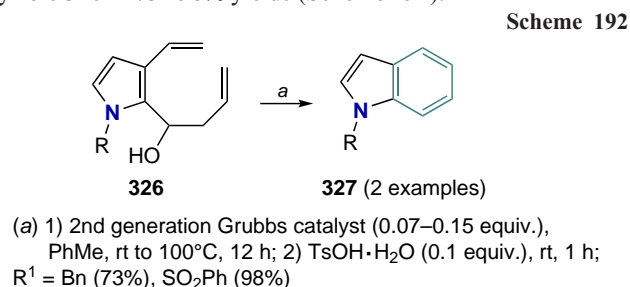


A series of carbazoles **324** containing unsubstituted benzene ring were synthesized by metathesis of 2,3-diallylindoles **41** followed by oxidation (Scheme 190).

Rhodium(II)-catalyzed annulation of 2,3-unsubstituted indole with buta-1,3-diene gives carbazole **325** (Scheme 191).¹²¹

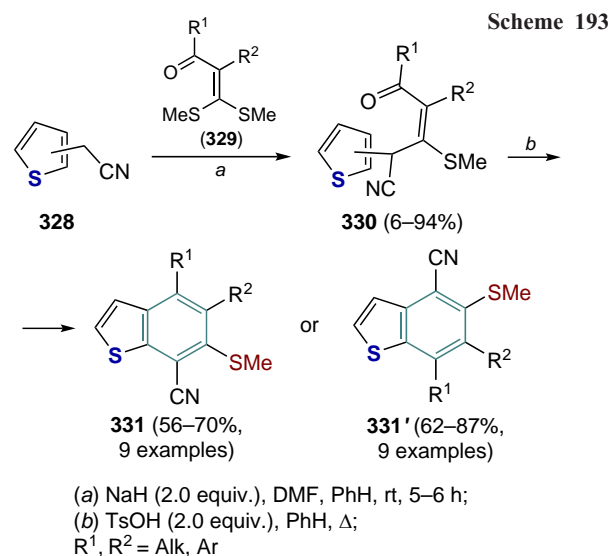


Two indoles **327**, containing unsubstituted carbon atoms, were synthesized by intramolecular metathesis of substituted pyrrole **326** in 73–98% yields (Scheme 192).¹²⁷



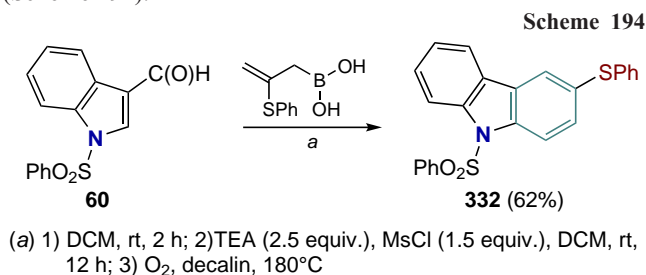
Unsubstituted BAHs were also isolated by annulation of nitro derivatives of heterocycles with some dienes (see Scheme 175).^{265–268}

The approaches to sulfur-containing BAHs are discussed below. A regioselective synthesis of polysubstituted benzo[*b*]-thiophenes has been developed based on annulation of 2(or 3)-cyanomethylthiophenes **328** with α -oxoketene dithioacetals **329** (Scheme 193).²⁸⁷ Adducts **330** are initially formed, which

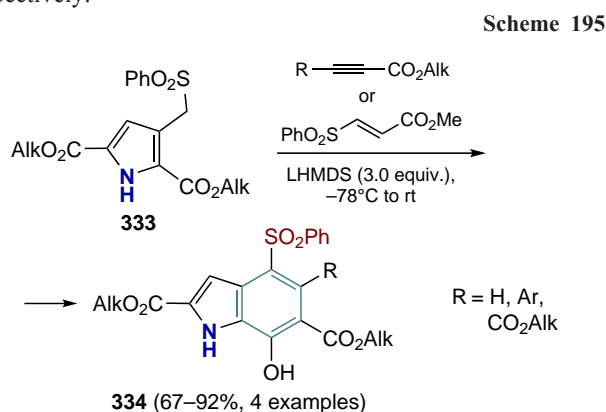


are converted in the second step of acid-induced cyclization into benzothiophenes **331** and **331'**, containing a methylthio group at position C2 or C3, in 56–87% yields.

An example of the assembly of 3-(phenylthio)carbazole **332** by annulation of 1-(phenylsulfonyl)indole-3-carbaldehyde with [2-(phenylthio)allyl]boronic acid in 62% yield is described (Scheme 194).¹¹⁰



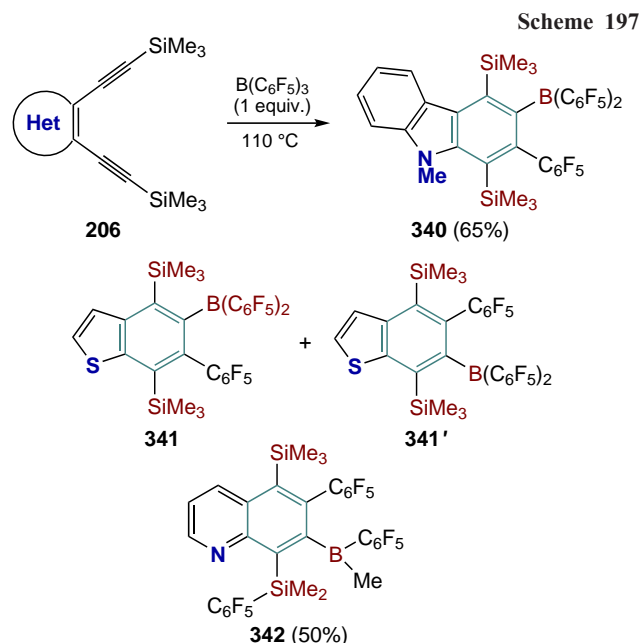
Indoles **334** with a C4-phenylsulfonyl moiety were obtained by nucleophilic annulation of ester- and (phenylsulfonyl) methyl-substituted pyrroles **333** with Michael acceptors (Scheme 195).²³⁷ A sulfur-containing product was also formed using (*E*)-methyl 3-(phenylsulfonyl)acrylate as a Michael acceptor. Additional substituents in the benzene ring of the products are hydroxyl and ester moieties in positions C1 and C3, respectively.



Sulfonyl-substituted carbazoles **335** and **336** (Ref. 212) and indoles **337–339** (Ref. 213) can be obtained by the reaction of the appropriate 2,3-unsubstituted indoles and pyrroles with 2-nitrobuta-1,3-dienes bearing two additional sulfonyl groups (Scheme 196). This affords mixtures of products. Benzoannulation of indole yields C1-methylsulfonylcarbazole **335** as the main product (40–63%), and in the case of annulation of

pyrrole, a mixture of three different indoles with sulfonyl substituents in positions C1 and C3 was isolated.

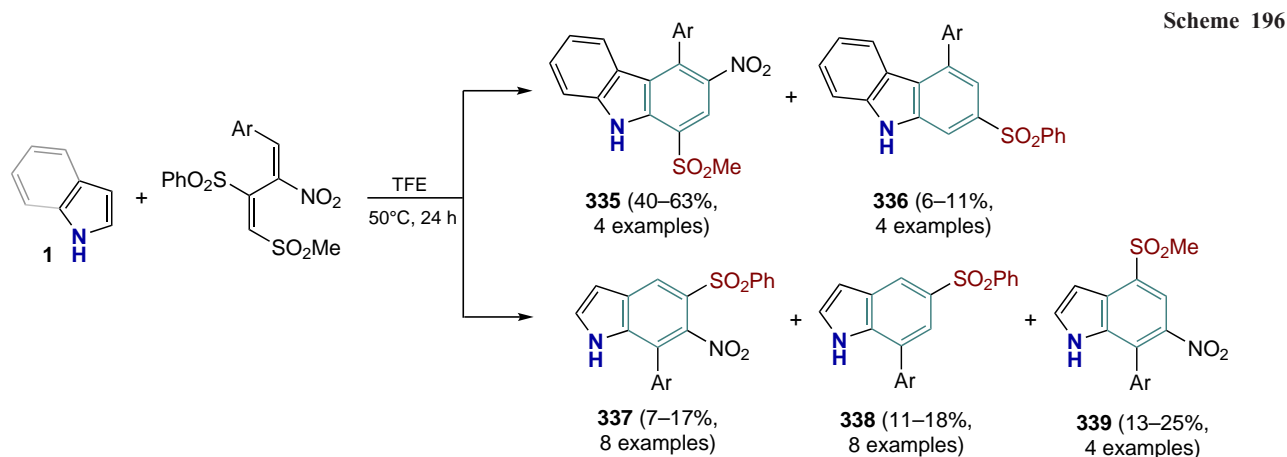
Examples of BAHs bearing other organoelement substituents are known. Carbazoles, benzothiophenes, and quinolines with silicon- and boron-containing groups were synthesized from 1,2-bis(trimethylsilyl)ethynylarenes **206** (Scheme 197).²⁸⁸

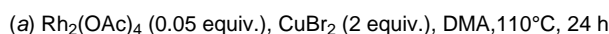
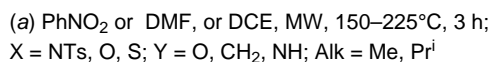


The latter react with tris(pentafluorophenyl)borane to give BAHs containing the tetrasubstituted benzene ring. In the case of indole annulation, the formation of a single product furnishes carbazole **340** (yield 65%), is observed, using thiophene gives a mixture of regioisomers **341** and **341'** (in a total yield of 75%), and the annulation of the pyridine derivative gives quinoline **342** (50%). The possibility of oxidative cleavage of the B(C₆F₅)₂ moiety to give hydroxy-substituted BAHs is shown.

Indoles, benzo[*b*]furans, benzothiophenes **344** and **345** with a trialkylsilyl substituent at position C1 or C4 were obtained by the microwave-assisted intramolecular Diels–Alder reaction of enyne-substituted heterocycles **343** (Scheme 198).¹⁹⁵ The position of the silicon substituent depends on the position of the enyne unit in the substrate (2 or 3). The yields of benzoannulated heterocycles vary from 41 to 86%.

The synthesis of carbazole **347** with a C3-diethylphosphoryl substituent was accomplished by the rhodium(II) acetate-





6. 'On-demand' functionalization of benzoheterocycles

positions of the resulting BAH to be varied depending on the nature of the reagents and the structure of the substrates should potentially be more valuable and versatile. Individual approaches to the ‘on-demand’ introduction of substituents of different nature at this stage of development of the benzoannulation strategy are discussed in this Section.

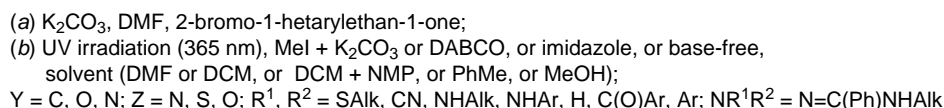
A two-step procedure for the benzannulation of hetarylacetic acids **306** has been developed, which yielded first dihetarylethenes **348** and then BAHs **349–352** bearing C-, N-, O-, or S-substituents at position C2 or C3 (Scheme 200).²⁸⁹

Dihetarylethenes **348** undergo photochemical rearrangement in the second step upon UV light irradiation. The substituent type in the resulting benzene ring depends on the nature of the heterocycle. Thus, the rearrangement of substrates bearing a thiophene, pyrrole or pyrazole ring gives C-substituted products **349** in 40–91% yields. A similar reaction of thiazole-based dihetarylethenes afforded S-substituted compounds **350** in 47–85% yields. Hydroxy-substituted heterocycles **351** were obtained using dihetarylethenes based on 5-substituted oxazole (29–54%). Finally, N-substituted BAHs **352** are formed during photorearrangement of oxazole- or imidazole-substituted substrates (35–94%). For each family of dihetarylethenes, optimal photolysis conditions were selected (e.g., addition of DABCO to eliminate side processes, use of alkylating agents to stabilize photolysis products).

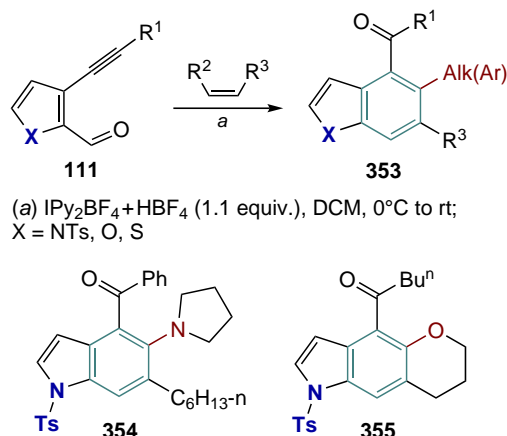
The above method^{142, 143} for obtaining BAHs based on the cyclization of 3-alkyne-substituted azole-2-carbaldehydes **111** with alkenes in the presence of bis(pyridine)iodonium tetrafluoroborate also provides ‘on-demand’ functionalization of the benzene ring at position C3 (Scheme 201).

The use of ordinary alkenes gives rise to alkyl- or aryl-substituted indoles, benzothiophenes and benzofurans **353**. When carrying the reaction with a pyrrolidine enamine or vinyl ether, N-substituted indole **354** or 1,6,7,8-tetrahydropyrano-[2,3-*f*]indole **355** are obtained. The yields of the products in this reaction are 31–85%.

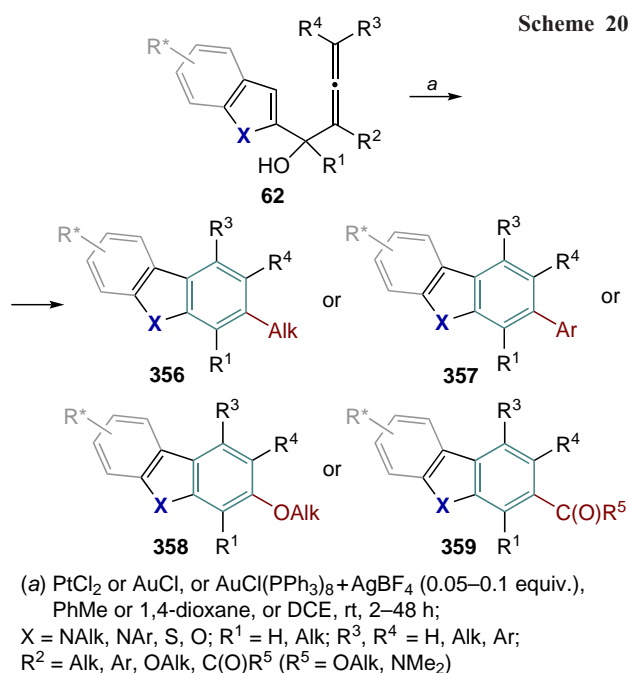
More than 60 representatives of carbazoles, benzothiophenes, benzofurans, and dibenzofurans **356–359**, functionalized with various groups at position C2, were obtained in 31–92% yields by cyclization of 2-substituted hetaryl allenols **62** under the



Scheme 201



Scheme 202



action of platinum(II) chloride,^{290–292} gold(I) chloride²⁹³ or a gold(I) complex in combination with silver(I) tetrafluoroborate^{294,295} (Scheme 202). This method is suitable for the synthesis of 3-substituted substrates and allows variation of alkyl, aryl, alkoxy groups, as well as amide and ester substituents at positions C2 or C3. Product yields using this approach are generally moderate to high.

7. Conclusion

By the end of the first decade of the 21st century, a separate trend had formed in the chemistry of heterocyclic compounds, associated with the development of methods for benzoannulation of heteroaromatic compounds. It is a logical continuation of the classical works on heteroannulation of benzene, carried out in the 19th and 20th centuries, and more recent studies on the development of methods for creating a benzene ring using organometallic reagents (for example, using Fischer carbenes²⁹⁶) and transition metal complexes (for example, by Pd-catalyzed annulation of the triple bond²⁹⁷).

This review is the first attempt to analyze the advances in the field of benzoannulation of aromatic heterocycles over the past twenty-five years. Based on the data obtained, it can be

concluded that the main attention of researchers has been paid to the development of methods for the annulation of indole to give carbazole. This could be explained, on the one hand, by the large number of carbazole-based natural compounds, including a large class of alkaloids, and on the other hand, by the diverse biological activities of such compounds. Methods of benzoannulation of five-membered rings with one heteroatom, *vis.*, pyrrole, thiophene and furan, are much less prevalent. Finally, a limited number of approaches have been described for creating a benzene ring at azoles and pyridine nuclei. In our opinion, this disproportion is primarily due to the researchers' clearly expressed interest in carbazole derivatives and only secondarily to the special reactivity of indole.

Benzoannulation of aromatic heterocycles generally consist of several steps. The key step, cyclization, most often represents an attack of an electrophile on the aromatic system of the heterocycle using Lewis or Brønsted acids or in the presence of transition metal complexes. Much less frequently, cyclizations involving free radicals or electrocyclization reactions (thermal or photochemical) are found in the literature.

Benzoannulation often requires prefunctionalization of the substrates, but a wide range of methods have been developed for the annulation of unsubstituted heterocycles. This strategy has been termed APEX (annulative π -extension).²⁹⁸

Existing benzoannulation procedures provide an access to a variety of substituted benzoheterocycles which can bear substituents such as alkyl and aryl groups, as well as alkene and alkyne moieties. A large number of methods have been developed for the preparation of BAHs with electron-withdrawing substituents, including acyl and carboxyl groups, as well as a nitro group. Some reactions have been described that produce BAHs with electron-donating substituents such as amino, hydroxy, alkoxy groups, as well as halogen atoms.

In general, it can be stated that today it is possible to obtain a wide variety of carbazole and indole derivatives. However, it is undoubtedly necessary to adapt the existing benzoannulation methods for the use of such heterocycles as furan, thiophene, pyridine and azoles (oxazole, imidazole, pyrazole, *etc.*). We hope that this review will stimulate organic synthetic chemists to fill the existing gaps in the development of new synthetic protocols for benzoannulation of the above heterocycles, which will open access to useful polyheterocyclic compounds intended for materials science, engineering and medicine.

Acknowledgement

The project was supported by the Ministry of Science and Higher Education of the Russian Federation within the framework of the Scientific and Educational Center 'Baikal' (Project No. FZZS-2024-0001).

8. List of abbreviations

All — allyl,
 BAH — benzoannulated heterocycle,
 BEA — 2-bromoethylamine hydrobromide,
 Boc — *tert*-butoxycarbonyl,
 BPA — 1,1'-binaphthyl-2,2'-dihydrophosphate,
 bpy — 2,2'-bipyridine,
 BrettPhos — 2-dicyclohexylphosphino-3,6-dimethoxy-2',4',6'-triisopropylbiphenyl,
 Cbz — carboxybenzyl,
 Cp* — pentamethylcyclopentadienyl,
m-CPBA — 3-chloroperoxybenzoic acid,
 CSA — camphorsulfonic acid,

DABCO — 1,4-diazabicyclo[2.2.2]octane,
 DBU — 1,8-diazabicyclo[5.4.0]undec-7-ene,
 DCB — dichlorobenzene,
 DCE — 1,2-dichloroethane,
 DCM — dichloromethane,
 DDQ — 2,3-dichloro-5,6-dicyano-1,4-benzoquinone,
 DG — directing group,
 DIPA — diisopropanolamine,
 DIPEA — *N,N*-diisopropylethylamine,
 DMA — *N,N*-dimethylacetamide,
 DMAD — dimethyl acetylenedicarboxylate,
 DMAP — 4-dimethylaminopyridine,
 DME — dimethoxyethane,
 DMF–DMA — *N,N*-dimethylformamide dimethyl acetal,
 DTBP — di-*tert*-butylperoxide,
 dtbbpy — 4,4'-di-*tert*-butyl-2,2'-bipyridine,
 EDC — 1-ethyl-3-(3-dimethylaminopropyl)carbodiimide,
 EWG — electron-withdrawing group,
 esp — α,α,α' -tetramethylbenzene-1,3-dipropionate,
 hfacac — 1,1,1,5,5,5-hexafluoroacetylacetate,
 HFIP — 1,1,1,3,3,3-hexafluoropropan-2-ol,
 HMIM — 1-hexyl-3-methylimidazolium,
 HMPA — hexamethylphosphoramide,
 IBX — 2-iodoxybenzoic acid,
 Ind — indol-3-yl,
 IPr — 1,3-bis(2,6-diisopropylphenyl)imidazol-2-ylidene,
 JackiePhos — 2-{bis[3,5-bis(trifluoromethyl)phenyl]-phosphino}-3,6-dimethoxy-2',4',6'-triisopropylbiphenyl,
 JohnPhos — 2-(di-*tert*-butylphosphino)biphenyl,
 LDA — lithium diisopropylamide,
 LED — light-emitting diode,
 LHDMS — lithium bis(trimethylsilyl)amide,
 Mes — 2,4,6-trimethylphenyl (mesityl),
 MOM — methoxymethyl,
 Ms — methanesulfonyl (mesyl),
 MS — molecular sieves,
 MW — microwave irradiation,
 NBP — *N*-bromophthalimide,
 NBS — *N*-bromosuccinimide,
 NHC — *N*-heterocyclic carbene,
 NMP — *N*-methyl-2-pyrrolidone,
 NuH — nucleophile,
 PEG — polyethylene glycol,
 Piv — pivaloyl,
 ppy — 2-phenylpyridine,
 Py — pyridyl,
 rt — room temperature,
 SEM — 2-(trimethylsilyl)ethoxymethyl,
 SPhos — 2-dicyclohexylphosphino-2',6'-dimethoxybiphenyl,
 TBAB — tetrabutylammonium bromide,
 TBAF — tetrabutylammonium fluoride,
 TBDMS — *tert*-butyldimethylsilyl,
 TBPMS — *tert*-butyldiphenylsilyl,
 TBHP — *tert*-butylhydroperoxide,
 TEA — triethylamine,
 TEMPO — (2,2,6,6-tetramethylpiperidin-1-yl)oxyl,
 Tf — trifluoromethanesulfonyl (triflyl),
 TFA — trifluoroacetic acid,
 TFAA — trifluoroacetic anhydride,
 TFE — 2,2,2-trifluoroethanol,
 TIPS — triisopropylsilyl,
 TMS — trimethylsilyl,
 Tol — *p*-methylphenyl (tolyl),
 Ts — *p*-toluenesulfonyl (tosyl),

XPhos — 2-dicyclohexylphosphino-2',4',6'-triisopropylbiphenyl.

9. References

- H.-J.Knölker, K.R.Reddy. *Chem. Rev.*, **102** (11), 4303 (2002); <https://doi.org/10.1021/cr020059j>
- C.Sánchez, C.Méndez, J.A.Salas. *Nat. Prod. Rep.*, **23** (6), 1007 (2006); <https://doi.org/10.1039/B601930G>
- A.J.Kochanowska-Karamyan, M.T.Hamann. *Chem. Rev.*, **110** (8), 4489 (2010); <https://doi.org/10.1021/cr900211p>
- L.Le Bozec, C.J.Moody. *Aust. J. Chem.*, **62** (7), 639 (2009); <https://doi.org/10.1071/CH09126>
- M.Millot, A.Dieu, S.Tomasi. *Nat. Prod. Rep.*, **33** (6), 801 (2016); <https://doi.org/10.1039/C5NP00134J>
- G.Diaz, I.L.Miranda, M.A.N.Diaz. In *Phytochemicals – Isolation, Characterisation and Role in Human Health*. (InTech, 2015); <https://doi.org/10.5772/59819>
- B.k.g.Bhatthula, J.r.Kanchani, V.r.Arava, M.C.S.Subha. *Tetrahedron*, **75** (7), 874 (2019); <https://doi.org/10.1016/j.tet.2019.01.003>
- S.Meena, K.S.Rajeev, V.Dwivedi, K.A.Singh, C.S.Chanotiya, M.Q.Akhtar, K.Kumar, K.A.Shasany, D.A.Nagegowda. *Sci. Rep.*, **7** (1), 44126 (2017); <https://doi.org/10.1038/srep44126>
- S.B.Markad, N.P.Argade. *J. Org. Chem.*, **81** (12), 5222 (2016); <https://doi.org/10.1021/acs.joc.6b00702>
- A.W.Schmidt, K.R.Reddy, H.-J.Knölker. *Chem. Rev.*, **112** (6), 3193 (2012); <https://doi.org/10.1021/cr200447s>
- M.Tang, H.Lu, L.Zu. *Nat. Commun.*, **15** (1), 941 (2024); <https://doi.org/10.1038/s41467-024-45140-2>
- B.W.Alexander, N.M.Bartfield, V.Gupta, B.Q.Mercado, M.Del Campo, S.B.Herzon. *Science*, **383**, 849 (2024); <https://doi.org/10.1126/science.adl6163>
- H.Ohno, S.Inuki. *Org. Biomol. Chem.*, **19** (16), 3551 (2021); <https://doi.org/10.1039/D0OB02577A>
- V.N.Charushin, E.V.Verbitskiy, O.N.Chupakhin, D.V.Vorobyeva, P.S.Gribanov, S.N.Osipov, A.V.Ivanov, S.V.Martynovskaya, E.F.Sagitova, V.D.Dyachenko, I.V.Dyachenko, S.G.Krivokolysko, V.V.Dotsenko, A.V.Aksenov, D.A.Aksenov, N.A.Aksenov, A.A.Larin, L.L.Fershtat, V.M.Muzalevskiy, V.G.Nenajdenko, A.V.Gulevskaya, A.F.Pozharskii, E.A.Filatova, K.V.Belyaeva, B.A.Trofimov, I.A.Balova, N.A.Danilina, A.I.Govdi, A.S.Tikhomirov, A.E.Shchekotikhin, M.S.Novikov, N.V.Rostovskii, A.F.Khlebnikov, Yu.N.Klimochkin, M.V.Leonova, I.M.Tkachenko, V.A.Mamedov, V.L.Mamedova, N.A.Zhukova, V.E.Semenov, O.G.Sinyashin, O.V.Borshchev, Yu.N.Luponosov, S.A.Ponomarenko, A.S.Fisyuk, A.S.Kostyuchenko, V.G.Ilkin, T.V.Beryozkina, V.A.Bakulev, A.S.Gazizov, A.A.Zagidullin, A.A.Karasik, M.E.Kukushkin, E.K.Beloglazkina, N.E.Golantsov, A.A.Festa, L.G.Voskresensky, V.S.Moshkin, E.M.Buev, V.Ya.Sosnovskikh, I.A.Mironova, P.S.Postnikov, V.V.Zhdankin, M.S.Yusubov, I.A.Yaremenko, V.A.Vil', I.B.Krylov, A.O.Terent'ev, Yu.G.Gorbunova, A.G.Martynov, A.Yu.Tsivadze, P.A.Stuzhin, S.S.Ivanova, O.I.Koifman, O.N.Burov, M.E.Kletschii, S.V.Kurbatov, O.I.Yarovaya, K.P.Volcho, N.F.Salakhutdinov, M.A.Panova, Ya.V.Burgart, V.I.Saloutin, A.R.Sitdikova, E.S.Shchegravina, A.Yu.Fedorov. *Russ. Chem. Rev.*, **93** (7), RCR5125 (2024); <https://doi.org/10.59761/RCR5125>
- D.S.Ryabukhin, A.V.Vasilyev. *Russ. Chem. Rev.*, **85** (6), 637 (2016); <https://doi.org/10.1070/RCR4550>
- O.A.Omelchuk, A.S.Tikhomirov, A.E.Shchekotikhin. *Russ. Chem. Rev.*, **85** (8), 817 (2016); <https://doi.org/10.1070/RCR4613>
- N.A.Zhukova, V.A.Mamedov. *Russ. Chem. Rev.*, **86** (10), 968 (2017); <https://doi.org/10.1070/RCR4749>

18. N.N.Makhova, L.I.Belen'kii, G.A.Gazieva, I.L.Dalinger, L.S.Konstantinova, V.V.Kuznetsov, A.N.Kravchenko, M.M.Krayushkin, O.A.Rakitin, A.M.Starosotnikov, L.L.Fershtat, S.A.Shevelev, V.Z.Shirinian, V.N.Yarovenko. *Russ. Chem. Rev.*, **89** (1), 55 (2020); <https://doi.org/10.1070/RCR4914>
19. V.A.Bakulev, Y.M.Shafran, N.A.Beliaev, T.V.Beryozkina, N.N.Volkova, M.N.Joy, Z.Fan. *Russ. Chem. Rev.*, **91** (7), RCR5042 (2022); <https://doi.org/10.1070/RCR5042>
20. M.S.Klenov, A.A.Voronin, A.M.Churakov, V.A.Tartakovsky. *Russ. Chem. Rev.*, **92** (8), RCR5089 (2023); <https://doi.org/10.59761/RCR5089>
21. P.G.Sergeev, R.A.Novikov, Y.V.Tomilov. *Russ. Chem. Rev.*, **93** (3), RCR5111 (2024); <https://doi.org/10.59761/RCR5111>
22. P.A.Koutentis, A.Y.Makarov, O.A.Rakitin, A.V.Zibarev. *Russ. Chem. Rev.*, **94** (1), RCR5146 (2025); <https://doi.org/10.59761/RCR5146>
23. E.V.Nosova, S.Achelle, G.N.Lipunova, V.N.Charushin, O.N.Chupakhin. *Russ. Chem. Rev.*, **88** (11), 1128 (2019); <https://doi.org/10.1070/rcr4887>
24. P.P.Zak, V.A.Lapina, T.A.Pavich, A.V.Trofimov, N.N.Trofimova, Y.B.Tsapelev. *Russ. Chem. Rev.*, **86** (9), 831 (2017); <https://doi.org/10.1070/RCR4735>
25. I.S.Antipin, M.V.Alfimov, V.V.Arslanov, V.A.Burilov, S.Z.Vatsadze, Ya.Z.Voloshin, K.P.Volcho, V.V.Gorbachuk, Yu.G.Gorbunova, S.P.Gromov, S.V.Dudkin, S.Yu.Zaitsev, L.Ya.Zakharova, M.A.Ziganshin, A.V.Zolotukhina, M.A.Kalinina, E.A.Karakhanov, R.R.Kashapov, O.I.Koifman, A.I.Kononov, V.S.Korenev, A.L.Maksimov, N.Zh.Mamardashvili, G.M.Mamardashvili, A.G.Martynov, A.R.Mustafina, R.I.Nugmanov, A.S.Ovsyannikov, P.L.Padnya, A.S.Potapov, S.L.Selektor, M.N.Sokolov, S.E.Solovieva, I.I.Stoikov, P.A.Stuzhin, E.V.Suslov, E.N.Ushakov, V.P.Fedin, S.V.Fedorenko, O.A.Fedorova, Yu.V.Fedorov, S.N.Chvalun, A.Yu.Tsivadze, S.N.Shtykov, D.N.Shurpik, M.A.Shcherbina, L.S.Yakimova. *Russ. Chem. Rev.*, **90** (8), 895 (2021); <https://doi.org/10.1070/RCR5011>
26. I.A.Lavrinchenko, T.D.Moseev, M.V.Varaksin, V.N.Charushin, O.N.Chupakhin. *Russ. Chem. Rev.*, **93** (10), RCR5130 (2024); <https://doi.org/10.59761/RCR5130>
27. E.A.Knyazeva, O.A.Rakitin. *Russ. Chem. Rev.*, **85** (10), 1146 (2016); <https://doi.org/10.1070/rcr4649>
28. H.K.Mehta, S.K.Pathan, S.M.Trivedi. *Russ. J. Org. Chem.*, **59** (S1), S123 (2023); <https://doi.org/10.1134/S10704280231301434>
29. L.N.Rogoza, N.F.Salakhutdinov. *Russ. Chem. Rev.*, **84** (1), 98 (2015); <https://doi.org/10.1070/RCR4430>
30. K.V.Balakin, R.Filosa, S.N.Lavrenov, A.S.Mkrtchyan, M.B.Nawrozki, I.A.Novakov. *Russ. Chem. Rev.*, **87** (6), 509 (2018); <https://doi.org/10.1070/rcr4791>
31. D.A.Guk, O.O.Krasnovskaya, E.K.Beloglazkina. *Russ. Chem. Rev.*, **90** (12), 1566 (2021); <https://doi.org/10.1070/RCR5016>
32. M.I.Kharitonova, I.D.Konstantinova, A.I.Miroshnikov. *Russ. Chem. Rev.*, **87** (11), 1111 (2018); <https://doi.org/10.1070/rcr4832>
33. A.D.Rogachev, S.O.Kuranov, N.F.Salakhutdinov. *Russ. Chem. Rev.*, **85** (12), 1313 (2016); <https://doi.org/10.1070/rcr4624>
34. S.S.Laev, O.I.Yarovaya, N.F.Salakhutdinov. *Russ. Chem. Rev.*, **93** (12), RCR5142 (2024); <https://doi.org/10.59761/RCR5142>
35. A.Mullard. *Nat. Rev. Drug Discov.*, **13** (2), 85 (2014); <https://doi.org/10.1038/nrd4239>
36. A.Mullard. *Nat. Rev. Drug Discov.*, **19** (2), 79 (2020); <https://doi.org/10.1038/d41573-020-00001-7>
37. A.Mullard. *Nat. Rev. Drug Discov.*, **20** (2), 85 (2021); <https://doi.org/10.1038/d41573-021-00002-0>
38. A.Mullard. *Nat. Rev. Drug Discov.*, **21** (2), 83 (2022); <https://doi.org/10.1038/d41573-022-00001-9>
39. A.Mullard. *Nat. Rev. Drug Discov.*, **15** (2), 73 (2016); <https://doi.org/10.1038/nrd.2016.15>
40. A.K.Clarke, J.M.Lynam, R.J.K.Taylor, W.P.Unsworth. *ACS Catal.*, **8** (8), 6844 (2018); <https://doi.org/10.1021/acscatal.8b00745>
41. M.Beesu, G.Mehta. *Adv. Synth. Catal.*, **363** (11), 2877 (2021); <https://doi.org/10.1002/adsc.202100170>
42. M.Inman, C.J.Moody. *Chem. Sci.*, **4** (1), 29 (2013); <https://doi.org/10.1039/C2SC21185H>
43. J.S.S.Neto, G.Zeni. *Org. Chem. Front.*, **7** (1), 155 (2020); <https://doi.org/10.1039/C9QO01315F>
44. G.R.Humphrey, J.T.Kueth. *Chem. Rev.*, **106** (7), 2875 (2006); <https://doi.org/10.1021/cr0505270>
45. D.F.Taber, P.K.Tirunahari. *Tetrahedron*, **67** (38), 7195 (2011); <https://doi.org/10.1016/j.tet.2011.06.040>
46. K.Speck, T.Magauer. *Beilstein J. Org. Chem.*, **9**, 2048 (2013); <https://doi.org/10.3762/bjoc.9.243>
47. Y.Fu, P.Sun, G.Li, R.He, L.Shi, N.Xing. *J. Heterocycl. Chem.*, **59** (7), 1135 (2022); <https://doi.org/10.1002/jhet.4451>
48. S.N.Georgiades, P.G.Nicolaou. In *Advances in Heterocyclic Chemistry*. Vol. 129. (Elsevier, 2019). P. 1; <https://doi.org/10.1016/bs.aihch.2018.10.001>
49. Z.Xu, D.Wu, C.Fang, Y.Li. *Des. Monomers Polym.*, **26** (1), 90 (2023); <https://doi.org/10.1080/15685551.2023.2194174>
50. A.Mushtaq, A.F.Zahoor, S.Ahmad, M.J.Saif, A.ul Haq, S.G.Khan, A.A.Al-Mutairi, A.Irfan, S.A.Al-Hussain, M.E.A.Zaki. *ACS Omega*, **9** (19), 20728 (2024); <https://doi.org/10.1021/acsomega.4c02677>
51. M.A.Metwally, M.E.Khalifa, G.A.El-Hiti. *J. Sulfur Chem.*, **31** (3), 205 (2010); <https://doi.org/10.1080/17415991003721266>
52. B.Wu, N.Yoshikai. *Org. Biomol. Chem.*, **14** (24), 5402 (2016); <https://doi.org/10.1039/C6OB00219F>
53. S.Sahu, A.Banerjee, S.Kundu, A.Bhattacharyya, M.S.Maji. *Org. Biomol. Chem.*, **20** (15), 3029 (2022); <https://doi.org/10.1039/D2OB00187J>
54. X.Zhang, S.Ma. *Isr. J. Chem.*, **58** (5), 608 (2018); <https://doi.org/https://doi.org/10.1002/ijch.201700034>
55. T.Aggarwal, S.Sushmita, A.K.Verma. *Org. Biomol. Chem.*, **17** (36), 8330 (2019); <https://doi.org/10.1039/C9OB01381D>
56. T.Mandal, J.Dash. *Org. Biomol. Chem.*, **19** (45), 9797 (2021); <https://doi.org/10.1039/D1OB01471D>
57. X.Y.Liu, Y.L.Liu, L.Chen. *Adv. Synth. Catal.*, **362** (23), 5170 (2020); <https://doi.org/10.1002/adsc.202000930>
58. M.S.Christodoulou, E.M.Beccalli, F.Foschi, S.Giofrè. *Synthesis*, **52** (19), 2731 (2020); <https://doi.org/10.1055/s-0040-1707123>
59. A.Banerjee, S.Kundu, A.Bhattacharyya, S.Sahu, M.S.Maji. *Org. Chem. Front.*, **8** (11), 2710 (2021); <https://doi.org/10.1039/D1QO00092F>
60. L.A.T.Allen, P.Natho. *Org. Biomol. Chem.*, **21** (45), 8956 (2023); <https://doi.org/10.1039/D3OB01605F>
61. S.Munawar, A.F.Zahoor, A.Mansha, T.H.Bokhari, A.Irfan. *RSC Adv.*, **14** (5), 2929 (2024); <https://doi.org/10.1039/D3RA07270C>
62. J.Roy, A.K.Jana, D.Mal. *Tetrahedron*, **68** (31), 6099 (2012); <https://doi.org/10.1016/j.tet.2012.05.007>
63. X.D.Duc. *Curr. Org. Chem.*, **24** (19), 2256 (2020); <http://dx.doi.org/10.2174/1385272824999200820151545>
64. N.A.Harry, S.M.Ujwaldev, G.Anilkumar. *Org. Biomol. Chem.*, **18** (48), 9775 (2020); <https://doi.org/10.1039/D0OB02000A>
65. R.Gujjarappa, N.Vodnala, C.C.Malakar. *Adv. Synth. Catal.*, **362** (22), 4896 (2020); <https://doi.org/10.1002/adsc.202000658>
66. D.Bag, S.D.Sawant. *Chem. – Eur. J.*, **27** (4), 1165 (2021); <https://doi.org/10.1002/chem.202002154>
67. S.Sengupta, S.Pabbaraja, G.Mehta. *Eur. J. Org. Chem.*, (42), e202200945 (2022); <https://doi.org/10.1002/ejoc.202200945>
68. M.Abid, A.Spaeth, B.Török. *Adv. Synth. Catal.*, **348** (15), 2191 (2006); <https://doi.org/10.1002/adsc.200606200>
69. A.Kulkarni, P.Quang, B.Török. *Synthesis*, (23), 4010 (2009); <https://doi.org/10.1055/s-0029-1217052>
70. H.Santos, L.A.Zeoly, R.Rebecchi, J.Arantes, F.Coelho, M.T.Rodrigues. *Adv. Synth. Catal.*, **366** (4), 884 (2024); <https://doi.org/10.1002/adsc.202301328>
71. X.Yi, K.Chen, W.Chen. *Adv. Synth. Catal.*, **360** (23), 4497 (2018); <https://doi.org/10.1002/adsc.201801043>
72. P.Tharra, B.Baire. *Org. Lett.*, **20** (4), 1118 (2018); <https://doi.org/10.1021/acs.orglett.8b00042>

73. D.Roy, P.Tharra, B.Baire. *Chem. Commun.*, **58** (73), 10210 (2022); <https://doi.org/10.1039/D2CC03526J>
74. K.Hayashi, K.Yoshida, A.Yanagisawa. *J. Org. Chem.*, **78** (7), 3464 (2013); <https://doi.org/10.1021/jo400099x>
75. C.R.Reddy, E.Srinivasu, P.Sathish, M.Subbarao, R.R.Donthiri. *J. Org. Chem.*, **86** (1), 1118 (2021); <https://doi.org/10.1021/acs.joc.0c02601>
76. V.Kanchupalli, D.Joseph, S.Katukojvala. *Org. Lett.*, **17** (23), 5878 (2015); <https://doi.org/10.1021/acs.orglett.5b03064>
77. A.Dagar, S.Biswas, S.M.Mobin, S.Samanta. *ChemistrySelect*, **1** (20), 6362 (2016); <https://doi.org/10.1002/slct.201601611>
78. A.Banerjee, S.Sahu, M.S.Maji. *Adv. Synth. Catal.*, **359** (11), 1860 (2017); <https://doi.org/10.1002/adsc.201700092>
79. M.A.Muñoz-Torres, S.Suárez-Pantiga, R.Sanz. *Org. Lett.*, **26** (23), 4969 (2024); <https://doi.org/10.1021/acs.orglett.4c01581>
80. E.Duval, G.D.Cuny. *Tetrahedron Lett.*, **45** (28), 5411 (2004); <https://doi.org/10.1016/j.tetlet.2004.05.053>
81. Y.Li, J.Waser. *Angew. Chem., Int. Ed.*, **54** (18), 5438 (2015); <https://doi.org/10.1002/anie.201412321>
82. T.I.Ho, J.Y.Wu, S.L.Wang. *Angew. Chem., Int. Ed.*, **38** (17), 2558 (1999); [https://doi.org/10.1002/\(SICI\)1521-3773\(19990903\)38:17<2558::AID-ANIE2558>3.0.CO;2-E](https://doi.org/10.1002/(SICI)1521-3773(19990903)38:17<2558::AID-ANIE2558>3.0.CO;2-E)
83. J.-Y.Wu, J.-H.Ho, S.-M.Shih, T.-L.Hsieh, T.-I.Ho. *Org. Lett.*, **1** (7), 1039 (1999); <https://doi.org/10.1021/ol990828u>
84. T.-I.Ho, J.-H.Ho, J.-Y.Wu. *J. Am. Chem. Soc.*, **122** (35), 8575 (2000); <https://doi.org/10.1021/ja0011562>
85. Y.-Z.Chen, C.-W.Ni, F.-L.Teng, Y.-S.Ding, T.-H.Lee, J.-H.Ho. *Tetrahedron*, **70** (9), 1748 (2014); <https://doi.org/10.1016/j.tet.2014.01.035>
86. A.S.K.Hashmi, W.Yang, F.Rominger. *Angew. Chem., Int. Ed.*, **50** (25), 5762 (2011); <https://doi.org/10.1002/anie.201100989>
87. A.S.K.Hashmi, W.Yang, F.Rominger. *Chem. – Eur. J.*, **18** (21), 6576 (2012); <https://doi.org/10.1002/chem.201200314>
88. X.Liu, D.Du, S.Li, X.Wang, C.Xu, M.Wang. *Adv. Synth. Catal.*, **362** (22), 5135 (2020); <https://doi.org/10.1002/adsc.202000865>
89. S.Suresh, H.S.Chien, C.H.Chen, H.Y.Tsai, D.R.Chung, V.Kavala, C.F.Yao. *J. Org. Chem.*, **88** (24), 17505 (2023); <https://doi.org/10.1021/acs.joc.3c01857>
90. S.-M.Yang, J.-J.Shie, J.-M.Fang, S.K.Nandy, H.-Y.Chang, S.-H.Lu, G.Wang. *J. Org. Chem.*, **67** (15), 5208 (2002); <https://doi.org/10.1021/jo0257849>
91. E.Nojman, L.Latos-Grażyński, L.Szterenber. *Eur. J. Org. Chem.*, (22), 4115 (2012); <https://doi.org/10.1002/ejoc.201200644>
92. S.Debnath, T.Das, T.K.Pati, S.Majumdar, D.K.Maiti. *J. Org. Chem.*, **85** (20), 13272 (2020); <https://doi.org/10.1021/acs.joc.0c01670>
93. J.Jia, J.Shi, J.Zhou, X.Liu, Y.Song, H.E.Xu, W.Yi. *Chem. Commun.*, **51** (14), 2925 (2015); <https://doi.org/10.1039/C4CC09823D>
94. L.Shi, X.Zhong, H.She, Z.Lei, F.Li. *Chem. Commun.*, **51** (33), 7136 (2015); <https://doi.org/10.1039/C5CC00249D>
95. Y.-W.Huang, X.-Y.Li, L.-N.Fu, Q.-X.Guo. *Org. Lett.*, **18** (23), 6200 (2016); <https://doi.org/10.1021/acs.orglett.6b03257>
96. S.Wang, Z.Chai, Y.Wei, X.Zhu, S.Zhou, S.Wang. *Org. Lett.*, **16** (13), 3592 (2014); <https://doi.org/10.1021/ol501605h>
97. J.Seo, C.-W.Chen, W.Lee, J.E.Jeon, P.-L.Chen, S.-C.Chuang, J.M.Joo. *Synthesis*, **53** (17), 3001 (2021); <https://doi.org/10.1055/a-1502-3641>
98. H.Kwak, E.Kang, J.Y.Song, G.Kang, J.M.Joo. *Adv. Synth. Catal.*, **363** (21), 4883 (2021); <https://doi.org/10.1002/adsc.202100763>
99. T.Mandal, G.Chakraborti, S.Karmakar, J.Dash. *Org. Lett.*, **20** (16), 4759 (2018); <https://doi.org/10.1021/acs.orglett.8b01827>
100. C.Praveen, P.Perumal. *Synlett*, (04), 521 (2011); <https://doi.org/10.1055/s-0030-1259537>
101. K.Huang, G.Sheng, P.Lu, Y.Wang. *Org. Lett.*, **19** (15), 4114 (2017); <https://doi.org/10.1021/acs.orglett.7b01769>
102. Y.Gu, W.Huang, S.Chen, X.Wang. *Org. Lett.*, **20** (14), 4285 (2018); <https://doi.org/10.1021/acs.orglett.8b01707>
103. M.J.James, R.E.Clubley, K.Y.Palate, T.J.Procter, A.C.Wyton, P.O'Brien, R.J.K.Taylor, W.P.Unsworth. *Org. Lett.*, **17** (17), 4372 (2015); <https://doi.org/10.1021/acs.orglett.5b02216>
104. N.Sun, X.Xie, Y.Liu. *Chem. – Eur. J.*, **20** (24), 7514 (2014); <https://doi.org/10.1002/chem.201400112>
105. Y.Matsuda, S.Naoe, S.Oishi, N.Fujii, H.Ohno. *Chem. – Eur. J.*, **21** (4), 1463 (2015); <https://doi.org/10.1002/chem.201405903>
106. G.Lupidi, B.Bassetti, R.Ballini, M.Petrini, A.Palmieri. *Asian J. Org. Chem.*, **10** (9), 2334 (2021); <https://doi.org/10.1002/ajoc.202100342>
107. R.Ballini, S.Gabrielli, A.Palmieri, M.Petrini. *Adv. Synth. Catal.*, **352** (14–15), 2459 (2010); <https://doi.org/10.1002/adsc.201000394>
108. M.L.Yao, X.Y.Wang, G.C.Feng, J.K.Liu, B.Wu, J.M.Yang. *Org. Lett.*, **25** (25), 4615 (2023); <https://doi.org/10.1021/acs.orglett.3c01186>
109. S.Chen, P.Jiang, P.Wang, Y.Pei, H.Huang, F.Xiao, G.-J.Deng. *J. Org. Chem.*, **84** (6), 3121 (2019); <https://doi.org/10.1021/acs.joc.8b02994>
110. G.Karan, S.Sahu, M.S.Maji. *Chem. Commun.*, **57** (43), 5274 (2021); <https://doi.org/10.1039/D1CC01512E>
111. B.Alcaide, P.Almendros, J.M.Alonso, M.T.Quirós, P.Gadziński. *Adv. Synth. Catal.*, **353** (11–12), 1871 (2011); <https://doi.org/10.1002/adsc.201100209>
112. S.Yadav, S.S.V.Ramasastri. *Chem. Commun.*, **57** (1), 77 (2021); <https://doi.org/10.1039/D0CC06695H>
113. S.Samala, A.K.Mandadapu, M.Saifuddin, B.Kundu. *J. Org. Chem.*, **78** (13), 6769 (2013); <https://doi.org/10.1021/jo400799b>
114. Y.Qiu, W.Kong, C.Fu, S.Ma. *Org. Lett.*, **14** (24), 6198 (2012); <https://doi.org/10.1021/ol3029498>
115. Z.Zhang, X.Tang, Q.Xu, M.Shi. *Chem. – Eur. J.*, **19** (32), 10625 (2013); <https://doi.org/10.1002/chem.201301203>
116. I.Martín-Mejías, C.Aragoncillo, H.Yanai, S.Hoshikawa, Y.Fujimoto, T.Matsumoto, P.Almendros. *Chem. Commun.*, **56** (12), 1795 (2020); <https://doi.org/10.1039/C9CC08930F>
117. M.Solas, M.A.Muñoz-Torres, F.Martínez-Lara, L.Renedo, S.Suárez-Pantiga, R.Sanz. *ChemPlusChem.*, **88** (11), e202300382 (2023); <https://doi.org/10.1002/cplu.202300382>
118. A.Suárez, S.Suárez-Pantiga, O.Nieto-Faza, R.Sanz. *Org. Lett.*, **19** (19), 5074 (2017); <https://doi.org/10.1021/acs.orglett.7b02303>
119. M.A.Muñoz-Torres, F.Martínez-Lara, M.Solas, S.Suárez-Pantiga, R.Sanz. *Adv. Synth. Catal.*, **364** (21), 3716 (2022); <https://doi.org/10.1002/adsc.202200824>
120. S.R.Kandukuri, L.Jiao, A.B.Machotta, M.Oestreich. *Adv. Synth. Catal.*, **356** (7), 1597 (2014); <https://doi.org/10.1002/adsc.201301108>
121. C.-J.Zhou, H.Gao, S.-L.Huang, S.-S.Zhang, J.-Q.Wu, B.Li, X.Jiang, H.Wang. *ACS Catal.*, **9** (1), 556 (2019); <https://doi.org/10.1021/acscatal.8b03595>
122. Q.Huang, R.C.Larock. *J. Org. Chem.*, **68** (19), 7342 (2003); <https://doi.org/10.1021/jo034449x>
123. S.Chen, Y.Li, P.Ni, H.Huang, G.-J.Deng. *Org. Lett.*, **18** (20), 5384 (2016); <https://doi.org/10.1021/acs.orglett.6b02762>
124. J.R.Stepherson, C.E.Ayala, T.H.Tugwell, J.L.Henry, F.R.Fronczek, R.Kartika. *Org. Lett.*, **18** (12), 3002 (2016); <https://doi.org/10.1021/acs.orglett.6b01376>
125. N.Thies, C.G.Hrib, E.Haak. *Chem. – Eur. J.*, **18** (20), 6302 (2012); <https://doi.org/10.1002/chem.201200188>
126. J.Kaufmann, E.Jäckel, E.Haak. *Angew. Chem., Int. Ed.*, **57** (20), 5908 (2018); <https://doi.org/10.1002/anie.201801846>
127. K.Yoshida, K.Hayashi, A.Yanagisawa. *Org. Lett.*, **13** (18), 4762 (2011); <https://doi.org/10.1021/ol201510u>
128. M.Yamashita, K.Hirano, T.Satoh, M.Miura. *Org. Lett.*, **11** (11), 2337 (2009); <https://doi.org/10.1021/ol900736s>
129. M.Yamashita, H.Horiguchi, K.Hirano, T.Satoh, M.Miura. *J. Org. Chem.*, **74** (19), 7481 (2009); <https://doi.org/10.1021/jo9016698>
130. K.Fukuzumi, Y.Unoh, Y.Nishii, T.Satoh, K.Hirano, M.Miura. *J. Org. Chem.*, **81** (6), 2474 (2016); <https://doi.org/10.1021/acs.joc.6b00030>

131. G.Song, X.Gong, X.Li. *J. Org. Chem.*, **76** (18), 7583 (2011); <https://doi.org/10.1021/jo201266u>
132. Q.Li, Y.Wang, B.Li, B.Wang. *Org. Lett.*, **20** (24), 7884 (2018); <https://doi.org/10.1021/acs.orglett.8b03438>
133. O.S.Kim, J.H.Jang, H.T.Kim, S.J.Han, G.C.Tsui, J.M.Joo. *Org. Lett.*, **19** (6), 1450 (2017); <https://doi.org/10.1021/acs.orglett.7b00410>
134. A.Banerjee, M.S.Maji. *Chem. – Eur. J.*, **25** (49), 11521 (2019); <https://doi.org/10.1002/chem.201902268>
135. A.Banerjee, A.Guin, S.Saha, A.Mondal, M.S.Maji. *Org. Biomol. Chem.*, **17** (7), 1822 (2019); <https://doi.org/10.1039/c8ob02875c>
136. S.Saha, M.S.Maji. *Org. Biomol. Chem.*, **18** (9), 1765 (2020); <https://doi.org/10.1039/c9ob02751c>
137. S.Kundu, A.Banerjee, M.S.Maji. *J. Org. Chem.*, **84** (24), 16003 (2019); <https://doi.org/10.1021/acs.joc.9b02474>
138. S.Min, T.Kim, T.Jeong, J.Yang, Y.Oh, K.Moon, A.Rakshit, I.S.Kim. *Org. Lett.*, **25** (23), 4298 (2023); <https://doi.org/10.1021/acs.orglett.3c01336>
139. D.Liu, J.Huang, Z.Fu, W.Huang. *Green Chem.*, **21** (5), 968 (2019); <https://doi.org/10.1039/C9GC00064J>
140. F.Yu, D.Li, Y.Wei, R.M.Kang, Q.X.Guo. *Tetrahedron*, **74** (15), 1965 (2018); <https://doi.org/10.1016/j.tet.2018.02.066>
141. S.Pramanik, S.Chatterjee, R.Banerjee, C.Chowdhury. *Org. Lett.*, **24** (10), 1895 (2022); <https://doi.org/10.1021/acs.orglett.2c00182>
142. J.Baruenga, H.Vázquez-Villa, A.Ballesteros, J.M.González. *Adv. Synth. Catal.*, **347** (4), 526 (2005); <https://doi.org/10.1002/adsc.200404293>
143. J.Baruenga, H.Vázquez-Villa, I.Merino, A.Ballesteros, J.M.González. *Chem. – Eur. J.*, **12** (22), 5790 (2006); <https://doi.org/10.1002/chem.200501505>
144. N.Asao, H.Aikawa. *J. Org. Chem.*, **71** (14), 5249 (2006); <https://doi.org/10.1021/jo060597m>
145. A.A.Merkushev, A.S.Makarov, P.M.Shpuntov, V.T.Abaev, I.V.Trushkov, M.G.Uchuskin. *Eur. J. Org. Chem.*, (8), 1274 (2021); <https://doi.org/10.1002/ejoc.202001608>
146. T.Guo, Q.Jiang, Z.Yu. *Org. Chem. Front.*, **2** (10), 1361 (2015); <https://doi.org/10.1039/C5QO00203F>
147. T.Guo, Q.Jiang, F.Huang, J.Chen, Z.Yu. *Org. Chem. Front.*, **1** (6), 707 (2014); <https://doi.org/10.1039/C4QO00122B>
148. T.Guo, L.Han, T.Wang, L.Lei, J.Zhang, D.Xu. *J. Org. Chem.*, **85** (14), 9117 (2020); <https://doi.org/10.1021/acs.joc.0c01056>
149. T.Guo, Y.Lin, D.Pan, X.Zhang, W.Zhu, X.-M.Cai, G.Huang, H.Wang, D.Xu, F.E.Kühn, B.Zhang, T.Zhang. *Nat. Commun.*, **14** (1), 6076 (2023); <https://doi.org/10.1038/s41467-023-41681-0>
150. J.Ji, C.Ding, S.Li, T.Guo, J.S.Reinhold, S.Meng, W.Zhu, X.Ji, X.-M.Cai, B.Zhang. *Green Chem.*, **26** (6), 3479 (2024); <https://doi.org/10.1039/d3gc004384c>
151. R.-J.Fang, C.Yan, J.Sun, Y.Han, C.-G.Yan. *Beilstein J. Org. Chem.*, **17**, 2425 (2021); <https://doi.org/10.3762/bjoc.17.159>
152. A.Palmieri, S.Gabrielli, D.Lanari, L.Vaccaro, R.Ballini. *Adv. Synth. Catal.*, **353** (9), 1425 (2011); <https://doi.org/10.1002/adsc.201000965>
153. X.-R.Wu, H.-L.Peng, L.-Q.Wei, L.-P.Li, S.-Y.Yao, B.-H.Ye. *Catal. Commun.*, **124**, 12 (2019); <https://doi.org/10.1016/j.catcom.2019.02.023>
154. A.K.Pandey, D.Kang, S.H.Han, H.Lee, N.K.Mishra, H.S.Kim, Y.H.Jung, S.Hong, I.S.Kim. *Org. Lett.*, **20** (15), 4632 (2018); <https://doi.org/10.1021/acs.orglett.8b01910>
155. M.Faltracco, M.Damian, E.Ruijter. *Org. Lett.*, **23** (19), 7592 (2021); <https://doi.org/10.1021/acs.orglett.1c02795>
156. T.Guney, J.J.Lee, G.A.Kraus. *Org. Lett.*, **16** (4), 1124 (2014); <https://doi.org/10.1021/ol403733n>
157. C.-J.Wu, W.-X.Cao, B.Chen, C.-H.Tung, L.-Z.Wu. *Org. Lett.*, **23** (6), 2135 (2021); <https://doi.org/10.1021/acs.orglett.1c00290>
158. R.K.Varshnaya, P.Banerjee. *J. Org. Chem.*, **84** (3), 1614 (2019); <https://doi.org/10.1021/acs.joc.8b02733>
159. I.Cikotiene, R.Buksnaitiene, R.Sazinas. *Tetrahedron*, **67** (4), 706 (2011); <https://doi.org/10.1016/j.tet.2010.11.073>
160. T.G.Back, A.Pandya, J.E.Wulff. *J. Org. Chem.*, **68** (8), 3299 (2003); <https://doi.org/10.1021/jo0268714>
161. K.S.Rathore, M.Harode, S.Katukojvala. *Org. Biomol. Chem.*, **12** (43), 8641 (2014); <https://doi.org/10.1039/C4OB01693A>
162. P.Dastoorani, M.T.Maghsoodlou, M.A.Khalilzadeh, E.Sarina. *Tetrahedron Lett.*, **57** (3), 314 (2016); <https://doi.org/10.1016/j.tetlet.2015.12.021>
163. D.Liu, Y.Gao, J.Huang, Z.Fu, W.Huang. *J. Org. Chem.*, **83** (22), 14210 (2018); <https://doi.org/10.1021/acs.joc.8b02532>
164. W.Huang, S.Chen, J.Yang, A.El-Harairy, X.Wang, M.Li, Y.Gu. *Adv. Synth. Catal.*, **361** (18), 4369 (2019); <https://doi.org/10.1002/adsc.201900425>
165. C.R.Reddy, U.Dilipkumar, M.D.Reddy. *Org. Lett.*, **16** (14), 3792 (2014); <https://doi.org/10.1021/ol501683v>
166. M.Bao, D.V.N.Carriola, H.Arman, M.P.Doyle. *Org. Lett.*, **26** (22), 4793 (2024); <https://doi.org/10.1021/acs.orglett.4c01628>
167. J.-Q.Wu, Z.Yang, S.-S.Zhang, C.-Y.Jiang, Q.Li, Z.-S.Huang, H.Wang. *ACS Catal.*, **5** (11), 6453 (2015); <https://doi.org/10.1021/acscatal.5b01801>
168. S.G.Dawande, V.Kanchupalli, J.Kalepu, H.Chennamsetti, B.S.Lad, S.Katukojvala. *Angew. Chem., Int. Ed.*, **53** (16), 4076 (2014); <https://doi.org/10.1002/anie.201400161>
169. Y.Du, W.Xue, R.Gao, Y.Gu, L.Han. *Tetrahedron Lett.*, **59** (48), 4221 (2018); <https://doi.org/10.1016/j.tetlet.2018.10.029>
170. J.Zhao, P.Li, C.Xia, F.Li. *Chem. – Eur. J.*, **21** (46), 16383 (2015); <https://doi.org/10.1002/chem.201503260>
171. F.Wu, W.Huang, Yiliqi, J.Yang, Y.Gu. *Adv. Synth. Catal.*, **360** (17), 3318 (2018); <https://doi.org/10.1002/adsc.201800669>
172. X.Zheng, L.Lv, S.Lu, W.Wang, Z.Li. *Org. Lett.*, **16** (19), 5156 (2014); <https://doi.org/10.1021/ol5025053>
173. P.M.Amin, W.Wang, C.Wang, J.Zhou, Y.Wang. *Chem. Commun.*, **60** (40), 5326 (2024); <https://doi.org/10.1039/d4cc00176a>
174. S.Lu, R.Xu, Z.Li. *Asian J. Org. Chem.*, **6** (11), 1604 (2017); <https://doi.org/10.1002/ajoc.201700353>
175. S.Saha, A.Banerjee, M.S.Maji. *Org. Lett.*, **20** (21), 6920 (2018); <https://doi.org/10.1021/acs.orglett.8b03063>
176. A.K.Verma, A.K.Danodia, R.K.Saunthwal, M.Patel, D.Choudhary. *Org. Lett.*, **17** (15), 3658 (2015); <https://doi.org/10.1021/acs.orglett.5b01476>
177. R.K.Saunthwal, M.Patel, S.Kumar, A.K.Danodia, A.K.Verma. *Chem. – Eur. J.*, **21** (51), 18601 (2015); <https://doi.org/10.1002/chem.201503657>
178. K.Ozaki, H.Zhang, H.Ito, A.Lei, K.Itami. *Chem. Sci.*, **4** (9), 3416 (2013); <https://doi.org/10.1039/c3sc51447a>
179. S.Kotha, V.R.Aswar, A.K.Chinnam. *Tetrahedron Lett.*, **58** (46), 4360 (2017); <https://doi.org/10.1016/j.tetlet.2017.10.002>
180. F.Fang, H.Zheng, G.Mao, S.Chen, G.-J.J.Deng. *Org. Chem.*, **87** (9), 6052 (2022); <https://doi.org/10.1021/acs.joc.2c00315>
181. R.Sureshbabu, R.Balamurugan, A.K.Mohanakrishnan. *Tetrahedron*, **65** (18), 3582 (2009); <https://doi.org/10.1016/j.tet.2009.03.010>
182. Y.Guo, Z.Wang, Y.Zhu, Q.Zhang, D.Wei, X.Liu, Z.Fu. *Org. Chem. Front.*, **6** (22), 3741 (2019); <https://doi.org/10.1039/C9QO01093A>
183. C.Liu, W.Huang, M.Wang, B.Pan, Y.Gu. *Adv. Synth. Catal.*, **358** (14), 2260 (2016); <https://doi.org/10.1002/adsc.201600185>
184. C.R.Reddy, R.R.Valletti, U.Dilipkumar. *Chem. – Eur. J.*, **22** (7), 2501 (2016); <https://doi.org/10.1002/chem.201503496>
185. C.R.Reddy, R.R.Valletti, P.Sathish. *J. Org. Chem.*, **82** (5), 2345 (2017); <https://doi.org/10.1021/acs.joc.6b02637>
186. Y.B.C.Kumar, P.Samatha, P.S.Mainkar, R.Adepu. *Org. Biomol. Chem.*, **20** (46), 9117 (2022); <https://doi.org/10.1039/D2OB01662A>
187. K.Lu, T.Luo, Z.Xiang, Z.You, R.Fathi, J.Chen, Z.Yang. *J. Comb. Chem.*, **7** (6), 958 (2005); <https://doi.org/10.1021/cc050099b>
188. D.Giomi, M.Cecchi. *Tetrahedron*, **58** (40), 8067 (2002); [https://doi.org/10.1016/S0040-4020\(02\)00958-4](https://doi.org/10.1016/S0040-4020(02)00958-4)
189. L.Zhou, B.Xu, J.Zhang. *Angew. Chem., Int. Ed.*, **54** (31), 9092 (2015); <https://doi.org/10.1002/anie.201503549>

190. S.M.T.Toguem, P.Langer. *Synlett*, (4), 513 (2011); <https://doi.org/10.1055/s-0030-1259552>
191. M.Hussain, T.T.Dang, P.Langer. *Synlett*, (11), 1822 (2009); <https://doi.org/10.1055/s-0029-1217357>
192. S.M.T.Toguem, M.Hussain, I.Malik, A.Villinger, P.Langer. *Tetrahedron Lett.*, **50** (35), 4962 (2009); <https://doi.org/10.1016/j.tetlet.2009.06.057>
193. M.Hussain, N.T.Hung, P.Langer. *Tetrahedron Lett.*, **50** (27), 3929 (2009); <https://doi.org/10.1016/j.tetlet.2009.04.069>
194. S.M.T.Toguem, I.Malik, M.Hussain, J.Iqbal, A.Villinger, P.Langer. *Tetrahedron*, **69** (1), 160 (2013); <https://doi.org/10.1016/j.tet.2012.10.050>
195. A.E.Bober, J.T.Proto, K.M.Brummond. *Org. Lett.*, **19** (7), 1500 (2017); <https://doi.org/10.1021/acs.orglett.7b00155>
196. S.Chen, L.Wang, J.Zhang, Z.Hao, H.Huang, G.-J.Deng. *J. Org. Chem.*, **82** (20), 11182 (2017); <https://doi.org/10.1021/acs.joc.7b02305>
197. S.Chen, Y.Li, P.Ni, B.Yang, H.Huang, G.-J.Deng. *J. Org. Chem.*, **82** (6), 2935 (2017); <https://doi.org/10.1021/acs.joc.6b02892>
198. J.K.Laha, N.Dayal. *Org. Lett.*, **17** (19), 4742 (2015); <https://doi.org/10.1021/acs.orglett.5b02265>
199. T.Gupta, J.B.Singh, K.Mishra, R.M.Singh. *RSC Adv.*, **7** (86), 54581 (2017); <https://doi.org/10.1039/C7RA09447G>
200. J.K.Annor-Gyamfi, E.Ametsetor, K.Meraz, R.A.Bunce. *Molecules*, **25** (21), 5168 (2020); <https://doi.org/10.3390/molecules25215168>
201. C.-W.Kuo, A.Konala, L.Lin, T.-T.Chiang, C.-Y.Huang, T.-H.Yang, V.Kavala, C.-F.Yao. *Chem. Commun.*, **52** (50), 7870 (2016); <https://doi.org/10.1039/C6CC03124B>
202. H.T.Kim, W.Lee, E.Kim, J.M.Joo. *Chem. – Asian J.*, **13** (17), 2418 (2018); <https://doi.org/10.1002/asia.201800558>
203. H.Kim, Y.J.Hwang, I.Han, J.M.Joo. *Chem. Commun.*, **54** (50), 6879 (2018); <https://doi.org/10.1039/C8CC02405G>
204. M.Beesu, G.Mehta. *J. Org. Chem.*, **84** (13), 8731 (2019); <https://doi.org/10.1021/acs.joc.9b00950>
205. S.Singh, R.Samineni, S.Pabbaraja, G.Mehta. *Org. Lett.*, **21** (9), 3372 (2019); <https://doi.org/10.1021/acs.orglett.9b01111>
206. S.Singh, S.Nerella, P.S.abbaraja, G.Mehta. *J. Org. Chem.*, **86** (17), 12093 (2021); <https://doi.org/10.1021/acs.joc.1c01104>
207. S.Sampaulesi, S.Gabrielli, R.Ballini, A.Palmieri. *Adv. Synth. Catal.*, **359** (19), 3407 (2017); <https://doi.org/10.1002/adsc.201700790>
208. P.K.Jaiswal, S.Biswas, S.Singh, S.Samanta. *Org. Biomol. Chem.*, **11** (48), 8410 (2013); <https://doi.org/10.1039/c3ob42034e>
209. N.Ramesh, G.G.Rajeshwaran, A.K.Mohanakrishnan. *Tetrahedron*, **65** (18), 3592 (2009); <https://doi.org/10.1016/j.tet.2009.03.009>
210. A.Pagano, M.Mancinelli, L.Bianchi, G.Giorgi, M.Maccagno, G.Petrillo, C.Tavani. *Tetrahedron*, **75** (33), 4506 (2019); <https://doi.org/10.1016/j.tet.2019.06.038>
211. L.Bianchi, G.Giorgi, M.Maccagno, G.Petrillo, C.Scapolla, C.Tavani. *Tetrahedron Lett.*, **53** (7), 752 (2012); <https://doi.org/10.1016/j.tetlet.2011.11.137>
212. L.Bianchi, M.Maccagno, M.Pani, G.Petrillo, C.Scapolla, C.Tavani. *Tetrahedron*, **71** (39), 7421 (2015); <https://doi.org/10.1016/j.tet.2015.05.046>
213. A.Benzi, L.Bianchi, M.Maccagno, A.Pagano, G.Petrillo, C.Tavani. *Molecules*, **24** (20), 3802 (2019); <https://doi.org/10.3390/molecules24203802>
214. S.Biswas, P.K.Jaiswal, S.Singh, S.M.Mobin, S.Samanta. *Org. Biomol. Chem.*, **11** (41), 7084 (2013); <https://doi.org/10.1039/c3ob41573b>
215. P.Sagar, R.Fröhlich, E.Würthwein. *Angew. Chem., Int. Ed.*, **43** (42), 5694 (2004); <https://doi.org/10.1002/anie.200461031>
216. V.Lyaskovskyy, R.Fröhlich, E.-U.Würthwein. *Synthesis*, (14), 2135 (2007); <https://doi.org/10.1055/s-2007-983752>
217. V.K.Outlaw, C.A.Townsend. *Org. Lett.*, **16** (24), 6334 (2014); <https://doi.org/10.1021/ol503078h>
218. M.Zhang, W.Ruan, H.-J.Zhang, W.Li, T.-B.Wen. *J. Org. Chem.*, **81** (4), 16963 (2016); <https://doi.org/10.1021/acs.joc.5b02340>
219. J.Barluega, A.Gómez, J.Santamaría, M.Tomás. *J. Am. Chem. Soc.*, **131** (41), 14628 (2009); <https://doi.org/10.1021/ja9063712>
220. I.Funes-Ardoiz, J.González, J.Santamaría, D.Sampedro. *J. Org. Chem.*, **81** (4), 1565 (2016); <https://doi.org/10.1021/acs.joc.5b02729>
221. S.Naoe, Y.Suzuki, K.Hirano, Y.Inaba, S.Oishi, N.Fujii, H.Ohno. *J. Org. Chem.*, **77** (11), 4907 (2012); <https://doi.org/10.1021/jo300771f>
222. D.Facoetti, G.Abbiati, E.Rossi. *Eur. J. Org. Chem.*, (17), 2872 (2009); <https://doi.org/10.1002/ejoc.200900039>
223. M.Dell'Acqua, G.Abbiati, A.Arcadi, E.Rossi. *Org. Biomol. Chem.*, **9** (22), 7836 (2011); <https://doi.org/10.1039/c1ob06271a>
224. P.Belmont, T.Belhadj. *Org. Lett.*, **7** (9), 1793 (2005); <https://doi.org/10.1021/ol050380z>
225. M.Tiano, P.Belmont. *J. Org. Chem.*, **73** (11), 4101 (2008); <https://doi.org/10.1021/jo800249f>
226. M.Faltracco, S.Ortega-Rosales, E.Janssen, R.C.Cioc, C.M.L.Vande Velde, E.Ruijter. *Org. Lett.*, **23** (8), 3100 (2021); <https://doi.org/10.1021/acs.orglett.1c00785>
227. T.-T.Wang, L.Zhao, Z.Y.-J.hang, W.-W.Liao. *Org. Lett.*, **18** (19), 5002 (2016); <https://doi.org/10.1021/acs.orglett.6b02460>
228. D.Cao, A.Ying, H.Mo, D.Chen, G.Chen, Z.Wang, J.Yang. *J. Org. Chem.*, **83** (20), 12568 (2018); <https://doi.org/10.1021/acs.joc.8b01876>
229. D.Cao, D.Chen, G.Chen, H.Mo, Z.Xia, K.Li, J.Yang. *Asian J. Org. Chem.*, **9** (10), 1610 (2020); <https://doi.org/10.1002/ajoc.202000303>
230. X.Liu, J.Huang, H.Xu, D.Zhang, Q.Sun, L.He. *Eur. J. Org. Chem.*, (5), 900 (2019); <https://doi.org/10.1002/ejoc.201801297>
231. D.Cao, G.Chen, D.Chen, Z.Xia, Z.Li, Y.Wang, D.Xu, J.Yang. *ACS Omega*, **6** (26), 16969 (2021); <https://doi.org/10.1021/acsomega.1c01992>
232. S.J.Gharpure, S.Somani, D.J.Fartade. *Adv. Synth. Catal.*, **366** (2), 194 (2024); <https://doi.org/10.1002/adsc.202301106>
233. S.Rajasekar, K.Ramachandran, P.Anbarasan. *J. Chem. Sci.*, **134** (4), 1 (2022); <https://doi.org/10.1007/s12039-022-02111-8>
234. A.S.K.Hashmi, S.Pankajakshan, M.Rudolph, E.Enns, T.Bander, F.Rominger, W.Frey. *Adv. Synth. Catal.*, **351** (17), 2855 (2009); <https://doi.org/10.1002/adsc.200900614>
235. D.Mal, B.K.Senapati, P.Pahari. *Tetrahedron*, **63** (18), 3768 (2007); <https://doi.org/10.1016/j.tet.2007.02.060>
236. X.Zhu, R.Guo, X.Zhang, Y.Gao, Q.Jia, Y.Wang. *Adv. Synth. Catal.*, **362** (15), 3190 (2020); <https://doi.org/10.1002/adsc.202000625>
237. B.K.Dinda, S.Basak, D.Mal. *Eur. J. Org. Chem.*, (25), 5521 (2014); <https://doi.org/10.1002/ejoc.201402486>
238. G.H.Zhu, X.Jiang. *Org. Lett.*, **25** (45), 8077 (2023); <https://doi.org/10.1021/acs.orglett.3c03132>
239. S.Yuan, G.Guerra Faura, H.E.Areheart, N.E.Peulen, S.France. *Molecules*, **27** (23), 8344 (2022); <https://doi.org/10.3390/molecules27238344>
240. S.Sakthivel, R.Balamurugan. *J. Org. Chem.*, **83** (19), 12171 (2018); <https://doi.org/10.1021/acs.joc.8b02127>
241. G.Guerra Faura, T.Nguyen, S.France. *J. Org. Chem.*, **86** (15), 10088 (2021); <https://doi.org/10.1021/acs.joc.1c00826>
242. L.Wang, G.Li, Y.Liu. *Org. Lett.*, **13** (15), 3786 (2011); <https://doi.org/10.1021/ol2012154>
243. Y.Qiu, J.Zhou, C.Fu, S.Ma. *Chem. – Eur. J.*, **20** (45), 14589 (2014); <https://doi.org/10.1002/chem.201404769>
244. A.Banerjee, S.Saha, M.S.Maji. *J. Org. Chem.*, **87** (6), 4343 (2022); <https://doi.org/10.1021/acs.joc.2c00042>
245. J.T.R.Liddon, M.J.James, A.K.Clarke, P.O'Brien, R.J.K.Taylor, W.P.Unsworth. *Chem. – Eur. J.*, **22** (26), 8777 (2016); <https://doi.org/10.1002/chem.201601836>

246. J.Liddon, J.Rossi-Ashton, A.Clark, J.Lynam, R.J.K.Taylor, W.Unsworth. *Synthesis*, **50** (24), 4829 (2018); <https://doi.org/10.1055/s-0037-1610181>
247. C.R.Reddy, M.Subbarao, P.Sathish, D.H.Kolgate, R.R.Donthiri. *Org. Lett.*, **22** (2), 689 (2020); <https://doi.org/10.1021/acs.orglett.9b04472>
248. S.Büttner, N.K.Kelzhanova, Z.A.Abilov, A.Villinger, P.Langer. *Tetrahedron*, **68** (18), 3654 (2012); <https://doi.org/10.1016/j.tet.2012.01.101>
249. L.Kong, M.Wang, Y.Wang, B.Song, Y.Yang, Q.Yao, Y.Li. *Chem. Commun.*, **54** (78), 11009 (2018); <https://doi.org/10.1039/C8CC06074F>
250. Y.Yang, J.Huang, H.Tan, L.Kong, M.Wang, Y.Yuan, Y.Li. *Org. Biomol. Chem.*, **17** (4), 958 (2019); <https://doi.org/10.1039/C8OB03031F>
251. Z.-G.Yuan, Q.Wang, A.Zheng, K.Zhang, L.-Q.Lu, Z.Tang, W.-J.Xiao. *Chem. Commun.*, **52** (29), 5128 (2016); <https://doi.org/10.1039/C5CC10542K>
252. Z.Jiang, J.Zhou, H.Zhu, H.Liu, Y.Zhou. *Org. Lett.*, **23** (11), 4406 (2021); <https://doi.org/10.1021/acs.orglett.1c01341>
253. J.Ramaraio, M.Rambabu, S.Suresh. *Org. Lett.*, **26** (9), 1780 (2024); <https://doi.org/10.1021/acs.orglett.3c04249>
254. X.Pan, B.Dong, Y.Wu, B.Gao, C.Song. *J. Org. Chem.*, **89** (12), 8845 (2024); <https://doi.org/10.1021/acs.joc.4c00730>
255. Y.Xiao, H.Xiong, S.Sun, J.Yu, J.Cheng. *Org. Biomol. Chem.*, **16** (45), 8715 (2018); <https://doi.org/10.1039/C8OB02145G>
256. T.Zhou, B.Li, B.Wang. *Chem. Commun.*, **53** (47), 6343 (2017); <https://doi.org/10.1039/C7CC02808C>
257. Q.Li, B.Li, B.Wang. *Chem. Commun.*, **54** (66), 9147 (2018); <https://doi.org/10.1039/C8CC04428G>
258. S.P.Shukla, R.Tiwari, A.K.Verma. *Tetrahedron*, **68** (44), 9035 (2012); <https://doi.org/10.1016/j.tet.2012.08.068>
259. X.Yang, J.Guo, H.Xiao, K.Feng, B.Chen, C.Tung, L.Wu. *Angew. Chem., Int. Ed.*, **59** (13), 5365 (2020); <https://doi.org/10.1002/anie.201916423>
260. B.M.Ramalingam, V.Saravanan, A.K.Mohanakrishnan. *Org. Lett.*, **15** (14), 3726 (2013); <https://doi.org/10.1021/ol4015738>
261. B.M.Ramalingam, A.K.Mohanakrishnan. *Tetrahedron Lett.*, **58** (30), 2919 (2017); <https://doi.org/10.1016/j.tetlet.2017.06.036>
262. M.Kim, E.Vedejs. *J. Org. Chem.*, **69** (20), 6945 (2004); <https://doi.org/10.1021/jo040191e>
263. T.Lu, Y.-T.Jiang, F.-P.Ma, Z.-J.Tang, L.Kuang, Y.-X.Wang, B.Wang. *Org. Lett.*, **19** (23), 6344 (2017); <https://doi.org/10.1021/acs.orglett.7b03186>
264. H.Wang, Z.Wang, Y.-L.Wang, R.-R.Zhou, G.-C.Wu, S.-Y.Yin, X.Yan, B.Wang. *Org. Lett.*, **19** (22), 6140 (2017); <https://doi.org/10.1021/acs.orglett.7b03021>
265. C.Della Rosa, M.Kneeteman, P.Mancini. *Tetrahedron Lett.*, **48** (8), 1435 (2007); <https://doi.org/10.1016/j.tetlet.2006.12.097>
266. T.L.Kishbaugh, G.W.Gribble. *Tetrahedron Lett.*, **42** (29), 4783 (2001); [https://doi.org/10.1016/S0040-4039\(01\)00854-1](https://doi.org/10.1016/S0040-4039(01)00854-1)
267. C.D.Della Rosa, P.M.E.Mancini, M.N.Kneeteman, A.F.Lopez Baena, M.A.Suligoy, L.R.Domingo. *J. Mol. Struct.*, **1079**, 47 (2015); <https://doi.org/10.1016/j.molstruc.2014.09.040>
268. C.D.Della Rosa, J.P.Sanchez, M.N.Kneeteman, P.M.E.Mancini. *Tetrahedron Lett.*, **52** (18), 2316 (2011); <https://doi.org/10.1016/j.tetlet.2011.02.074>
269. M.V.Gómez, A.I.Aranda, A.Moreno, F.P.Cossio, A.de Cózar, Á.Díaz-Ortiz, A.de la Hoz, P.Prieto. *Tetrahedron*, **65** (27), 5328 (2009); <https://doi.org/10.1016/j.tet.2009.04.065>
270. Y.Qiao, X.Wu, Y.Zhao, Y.Sun, B.Li, S.Chen. *Adv. Synth. Catal.*, **360** (11), 2138 (2018); <https://doi.org/10.1002/adsc.201800154>
271. S.Alavi, J.-B.Lin, H.K.Grover. *Org. Lett.*, **23** (14), 5559 (2021); <https://doi.org/10.1021/acs.orglett.1c01965>
272. X.Li, H.Xie, X.Fu, J.Liu, H.Wang, B.Xi, P.Liu, X.Xu, W.Tang. *Chem. – Eur. J.*, **22** (30), 10410 (2016); <https://doi.org/10.1002/chem.201602088>
273. J.Liu, C.J.Simmons, H.Xie, F.Yang, X.Zhao, Y.Tang, W.Tang. *Adv. Synth. Catal.*, **359** (4), 693 (2017); <https://doi.org/10.1002/adsc.201600992>
274. K.Liu, S.Zhang. *ACS Med. Chem. Lett.*, **6** (8), 894 (2015); <https://doi.org/10.1021/acsmedchemlett.5b00158>
275. T.Kono, R.Sasaki, H.Goto, M.Kakuno, Y.Tanabe. *Molecules*, **26** (22) (2021); <https://doi.org/10.3390/molecules26227008>
276. C.-S.Kim, K.C.Russell. *J. Org. Chem.*, **63** (23), 8229 (1998); <https://doi.org/10.1021/jo980879p>
277. J.Wang, H.T.Zhu, Y.F.Qiu, Y.Niu, S.Chen, Y.X.Li, X.Y.Liu, Y.M.Liang. *Org. Lett.*, **17** (12), 3186 (2015); <https://doi.org/10.1021/acs.orglett.5b01590>
278. B.Alcaide, P.Almendros, J.M.Alonso, S.Cembellin, I.Fernández, T.M.Del Campo, M.R.Torres. *Chem. Commun.*, **49** (71), 7779 (2013); <https://doi.org/10.1039/c3cc44073g>
279. B.Alcaide, P.Almendros, J.M.Alonso, E.Busto, I.Fernández, M.P.Ruiz, G.Xiaokaiti. *ACS Catal.*, **5** (6), 3417 (2015); <https://doi.org/10.1021/acscatal.5b00471>
280. B.Alcaide, P.Almendros, C.Aragoncillo, E.Busto, C.G.López-Calixto, M.Liras, V.A.de la Peña O'Shea, A.García-Sánchez, H.V.Stone. *Chem. – Eur. J.*, **24** (30), 7620 (2018); <https://doi.org/10.1002/chem.201801179>
281. I.Martín-Mejías, A.S.Petcu, J.M.Alonso, C.Aragoncillo, P.Almendros. *Adv. Synth. Catal.*, **364** (22), 3954 (2022); <https://doi.org/10.1002/adsc.202200564>
282. S.Yaragorla, R.Dada, D.Bag. *Eur. J. Org. Chem.*, (41), 6983 (2019); <https://doi.org/10.1002/ejoc.201901393>
283. S.Yaragorla, D.Bag, R.Dada, K.V.J.Jose. *ACS Omega*, **3** (11), 15024 (2018); <https://doi.org/10.1021/acsomega.8b02393>
284. S.Yaragorla, T.Khan, S.Chakroborty. *Org. Biomol. Chem.*, **22** (18), 3622 (2024); <https://doi.org/10.1039/d4ob00421c>
285. S.M.Rafiq, R.Sivasakthikumaran, A.K.Mohanakrishnan. *Org. Lett.*, **16** (10), 2720 (2014); <https://doi.org/10.1021/ol501006t>
286. B.Padhi, G.Kang, E.Kim, J.Ha, H.T.Kim, J.Lim, J.M.Joo. *ACS Catal.*, **10** (3), 1792 (2020); <https://doi.org/10.1021/acscatal.9b05177>
287. J.R.Suresh, O.Barun, H.Ila, H.Junappa. *Tetrahedron*, **56** (41), 8153 (2000); [https://doi.org/10.1016/S0040-4020\(00\)00722-5](https://doi.org/10.1016/S0040-4020(00)00722-5)
288. R.Liedtke, F.Tenberge, C.G.Daniliuc, G.Kehr, G.Erker. *J. Org. Chem.*, **80** (4), 2240 (2015); <https://doi.org/10.1021/jo502753s>
289. A.G.Lvov, A.M.Kavun, V.V.Kachala, K.A.Lyssenkov, V.Z.Shirinian. *Org. Biomol. Chem.*, **17** (20), 4990 (2019); <https://doi.org/10.1039/C9OB00690G>
290. W.Kong, C.Fu, S.Ma. *Chem. Commun.*, (30), 4572 (2009); <https://doi.org/10.1039/b909649c>
291. W.Kong, C.Fu, S.Ma. *Org. Biomol. Chem.*, **10** (10), 2164 (2012); <https://doi.org/10.1039/c1ob06474f>
292. W.Kong, Y.Qiu, X.Zhang, C.Fu, S.Ma. *Adv. Synth. Catal.*, **354** (11–12), 2339 (2012); <https://doi.org/10.1002/adsc.201100935>
293. Y.Qiu, D.Ma, C.Fu, S.Ma. *Org. Biomol. Chem.*, **11** (10), 1666 (2013); <https://doi.org/10.1039/c2ob27432a>
294. Y.Qiu, D.Ma, C.Fu, S.Ma. *Tetrahedron*, **69** (30), 6305 (2013); <https://doi.org/10.1016/j.tet.2013.04.099>
295. Y.Qiu, C.Fu, X.Zhang, S.Ma. *Chem. – Eur. J.*, **20** (33), 10314 (2014); <https://doi.org/10.1002/chem.201402423>
296. K.H.Dötz, P.Tomuschat. *Chem. Soc. Rev.*, **28** (3), 187 (1999); <https://doi.org/10.1039/A801442F>
297. M.Rubin, A.W.Sromek, V.Gevorgyan. *Synlett*, (15), 2265 (2003); <https://doi.org/10.1055/s-2003-43334>
298. H.Ito, K.Ozaki, K.Itami. *Angew. Chem., Int. Ed.*, **56** (37), 11144 (2017); <https://doi.org/10.1002/anie.201701058>